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DOI:

[10.1088/1748-605X/ab591d](https://doi.org/10.1088/1748-605X/ab591d)

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Document Version

Peer reviewed version

Citation for published version (Harvard):

Norris, E & Poologasundarampillai, G 2019, 'Electrospinning 3D bioactive glasses for wound healing', *Biomedical Materials*. <https://doi.org/10.1088/1748-605X/ab591d>

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Electrospinning 3D bioactive glasses for wound healing

Elizabeth Norris¹, Carolina Ramos-Rivera², Gowsihan Poologasundarampillai³, Joshua P. Clark⁴, Qun Ju⁵, Akiko Obata⁵, John V. Hanna⁴, Toshihiro Kasuga⁵, Christopher A. Mitchell⁶, Gavin Jell² and Julian R. Jones^{1*}

¹Department of Materials, Imperial College London, South Kensington Campus, London, SW7 2AZ, UK;

²Royal Free Medical School, Division of Surgery, University College London, UK;

³School of Dentistry, University of Birmingham, Birmingham, UK

⁴Department of Physics, University of Warwick, Gibbet Hill Rd., Coventry CV4 7AL, UK

⁵Graduate School of Engineering, Nagoya Institute of Technology, Gokiso-cho, Showa-ku, Nagoya 466-8555, Japan

⁶School of Biomedical Sciences, Ulster University, Cromore Road, Coleraine Co., Londonderry BT52 1SA, UK

E-mail: julian.r.jones@imperial.ac.uk

Received xxxxxx

Accepted for publication xxxxxx

Published xxxxxx

Abstract

An electrospinning technique was used to produce three-dimensional (3D) bioactive glass fibrous scaffolds, in the SiO₂-CaO system, for wound healing applications. Previously, it was thought that 3D cotton wool-like structures could only be produced when the sol contained calcium nitrate, implying that the Ca²⁺ and its electronic charge had a significant effect on the structure produced. Here, fibres with a 3D appearance were also electrospun from compositions containing only silica. A polymer binding agent was added to inorganic sol-gel solutions, enabling electrospinning prior to bioactive glass network formation and the polymer was removed by calcination. While the addition of Ca²⁺ contributes to the 3D morphology, here we show that other factors, such as relative humidity, play an important role in producing the 3D cotton-wool-like macrostructure of the fibres. A human dermal fibroblast cell line (CD-18CO) was exposed to dissolution products of the samples. Cell proliferation and metabolic activity tests were carried out and a VEGF ELISA showed a significant increase in VEGF production in cells exposed to the bioactive glass samples compared to control in DMEM. A novel SiO₂-CaO nanofibrous scaffold was created that showed tailorable physical and dissolution properties, the control and composition of these release products are important for directing desirable wound healing interactions.

Keywords: Electrospinning, Sol-gel, Bioactive glasses, 3D cotton-wool-like structure, Wound healing

1. Introduction

The successful treatment of wounds is a growing challenge for healthcare systems worldwide. In the UK alone, the National Health Service (NHS) treated an estimated 2.2 million wounds in 2012/2013 costing a total of £5.3 billion for

the management and associated comorbidities of the wounds [1].

Wounds that fail to heal in a timely manner or progress through the normal stages of healing are known as chronic wounds. Chronic wounds typically occur as complications from underlying health conditions such as foot ulcers in patients with diabetes [2]. Factors such as age, lifestyle and

medication can also increase the likelihood of occurrence [3] [4]. Chronic wounds typically experience prolonged or excessive inflammation and are at risk from persistent infections [5].

Clinical procedures for healing chronic wounds commonly include compression therapies, skin substitutes or dressings [6]. Tissue-engineered products [7], such as allograft-based bilayered cultured skin equivalents exist, e.g. Apligraf® (Organogenesis), which combines a dermal layer of cultured fibroblasts and bovine type I collagen with an epidermal layer of cultured keratinocytes, producing epidermis [8-9] and an environment similar to that of a normal healing wound [6]. While this produces a well-differentiated epidermis and dermis, it misses some of the components found in healthy skin, such as pigments, hair follicles, sweat glands and immune cells [9]. Tissue engineering is expensive and time consuming [10], therefore cost efficient biodegradable devices are needed which can be used with conventional dressings and stimulate healing of chronic wounds.

Bioactive glasses are biodegradable devices that have traditionally been used in bone regeneration [11-12]. Bioactive glasses can also bond to soft tissue [13] and the original bioactive glass composition, 45 wt% SiO₂, 24.5 wt% CaO, 24.5 wt% Na₂O, and 6.0 wt% P₂O₅, known as 45S5 Bioglass® [14], developed by Hench [14-15], has been investigated for wound healing [17] [18]. Day *et al.* [19] demonstrated that 45S5 can stimulate the expression and release of VEGF and β -FGF by CD-18Co human fibroblasts, suggesting that by the release of growth factors, Bioglass could potentially stimulate neovascularisation. Other bioactive glass compositions may also stimulate VEGF, providing an application for bioactive glasses in wound healing.

Mo-Sci Corp. (Rolla, MO) developed borate based melt derived bioactive glass fibres which have a 3D appearance similar to cotton-wool (Mirragen®, ETS Woundcare, Rolla, MO). They used a melt blowing technique [20] [21] to produce borate glass fibres and the device showed successful clinical results in healing diabetic foot ulcers that were not healing under conventional treatment [22]. Importantly, the device was not a dressing: the cotton wool-like glass was packed into the wound bed and a dressing was placed over it and the wound. The device and the dressing were replaced approximately every two days. The aim here was to produce a silicate based bioactive glass scaffold cotton wool-like device with tighter control over the fibre diameter and to carry out preliminary cell studies to investigate its potential for wound healing applications. Combination of the sol-gel process and the electrospinning technique were hypothesised to give tight control over the glass composition and fibre diameter.

Sol-gel derived silica glasses are made through the hydrolysis of alkoxide precursors, to make a sol, and gelation through polycondensation of silicate species [23]. They have

an inherent nanoporosity [21], which increases the surface area of the glass compared with dense melt-derived glasses. This enables control over the dissolution rates [20] of the glasses which is beneficial for wound healing applications, where short recovery times are imperative [4]. There are various methods for processing of sol-gel glasses into 3D macrostructures, such as by foaming [24] or as fibres [25].

Electrospinning can be used to make fibrous structures at a scale that can mimic some of the key structural features of the extracellular matrix (ECM) [26] [27] of the skin [28], [29]. The collagen fibrous structure is formed by multi-fibrils, organised in a three-dimensional fibre network.

Electrospinning is versatile [30] but typically produces a 2D fibremat structure, which could be used in shallow wounds [31]. Poologasundarampillai *et al.* [25] electrospun a sol-gel 70 mol% SiO₂ and 30 mol% CaO (70S30C) bioactive glass composition directly from the sol and produced a 3D structure that had an appearance similar to cotton-wool or cotton candy. Their focus was on bone regeneration and bioactivity with scaffolds of high Ca content for bone bonding and mineralisation. With the ability to be packed into spaces, the morphology may be more suitable for deep wounds in the skin. The production of a 3D structure with new compositions and the potential to mimic the natural extracellular matrix could potentially impact the behaviour and proliferation of cells [32]. This '3D' structure is different from some literature definitions which define '3D' as 2D fibremats which are thicker (~0.5–2 mm) than conventional fibremats (<0.5 mm) [33], [34].

The formation of a 3D structure was attributed to two factors: i) the rheological properties and surface tension of the sol; ii) the distribution of conductive ionic species (Ca²⁺ ions). They proposed that the application of an electric field to the 70S30C sol caused the separation of calcium and silicate ions resulting in a heterogeneous distribution of Ca²⁺ and SiO₄⁴⁻ at the Taylor cone. The position and quantity of separated charges at the cone determines the forces on the fibres as they spin and consequentially affects how the fibres are whipped as they move towards the collector plate. A similar effect was produced by electrospinning certain polymers [35], [36], e.g. polyaniline, when an external electric field concentrates charge carriers at favourable points within fibres through a conductive medium. Spinning a 100S sol (100 mol% SiO₂) composition, not containing Ca ions, produced the traditional, 2D structure.

Our hypothesis here was that more parameters affect the morphology of the scaffolds, such as the solution viscosity [37]; the voltage applied and distance between the needle tip and the collector plate [29], [36]. Environmental factors such as the relative humidity [27–29] and temperature [41] of the surroundings can also affect the morphology of electrospun fibres. Our initial studies on electrospinning sol-gel, found the viscosity to be partly dependent on the degree of crosslinking

(gelation) induced by the hydrolysis and condensation of the silica sol-gel network [37] and partly by the amount of polyvinyl butryal additive used.

Here, the aim was to develop a robust protocol for production of a bioactive glass scaffold with a 3D porous structure intended for use in wound healing and to investigate VEGF production by a human dermal fibroblast cell line (CD-18CO). The role of calcium in the fibres was a key focus so the quantity of calcium was varied compositions of 0 – 40 mol% CaO.

2. Materials and methods

2.1 Sample preparation

All reagents were obtained from Sigma-Aldrich (UK) unless otherwise stated. An electrospinning step was added to the conventional sol-gel process. Bioactive glass sols were produced with nominal compositions based on the 100- $XSiO_2$ - $XCaO$ system (molar ratios, coded 100- $XSXC$), 100S, 60S40C, 70S30C, 80S20C and 90S10C. Sols were produced by mixing tetraethyl orthosilicate (TEOS), ethanol, water with 1 M nitric acid (HNO_3) using a magnetic stirrer. Calcium nitrate tetrahydrate was added after 1 h (Table 1). The sol was stirred at room temperature for 24 h and then left to rest for a following 24 h. Polyvinylbutyral (Butvar-98) was combined in equal parts with the sol, in order to obtain a viscosity within the range for electrospinning long continuous fibres of 0.56–0.95 Pa.s, as identified by Poologasundarampillai *et al.* [25]. An AntonPar Compact Rheometer MCR was used to measure viscosity. 1 mL of solution was placed on the centre of the dish before a flat-headed probe was automatically lowered onto the sample. Twelve readings for viscosity were taken per sample.

The hybrid solution was loaded into a plastic syringe with a gauge 22 stainless steel needle which was connected to a high voltage power supply set at 10 kV. A syringe pump was used at a feed rate of 0.4 mL h^{-1} . The sol was electrospun onto a ground aluminium foil sheet at a distance of 100–120 mm from the end of the needle. The fibres were transferred to a ceramic container for stabilisation, where they were heated to 600 °C at a rate of 1 °C min^{-1} with a dwell at 600 °C for 3 h before cooling naturally.

2.2 Morphology and physical structure

Relative humidity was measured using a humidity monitor, and was taken with each sample at the start and end of every electrospinning session. If electrospinning took place over a number of hours, and the relative humidity had changed, an average of several results was taken. The chemical compositions of the samples were determined through the use of a lithium metaborate flux: 50 mg of the glass fibers were combined with 150 mg of the lithium metaborate in a platinum crucible and heated in to 1050 °C. When cooled the combination of flux and sample was dissolved in 2 M nitric

acid. The dissolved sample was then analysed using inductively coupled plasma-optical emission spectroscopy (ICP-OES) to determine the actual composition of the samples produced.

Scanning electron microscopy (SEM) was used to produce images of the fibres (JEOL-6V and LEO-1525). Samples were sputter coated with chromium to a maximum coating of 20 nm, and were observed under a 5 kV accelerating voltage and at a working distance of 8 mm. Fibre diameters were measured ($n = 10$) from SEM and analysed on ImageJ.

Nitrogen sorption (Autosorb1; Quantachrome, Fleet, UK) was used to investigate the surface area and the mesoporosity of the electrospun bioactive glass samples after degassing for at least 12 h. Attenuated total reflectance-Fourier transform infrared spectroscopy (ATR-FTIR Nicolet iS10) and X-ray diffraction (XRD, Bruker D2 PHASER) were carried out on the bioactive glass. XRD uses a step scanning with Cu radiation, at 30 kV and 10 mA, with 0.035° 2θ step size and a count rate of 12 second per step, from 6° to 70° 2θ . FTIR spectra were collected in absorbance mode using a diamond ATR accessory in the range of 500–4000 cm^{-1} .

Solid-state ^{29}Si magic angle spinning nuclear magnetic resonance (MAS-NMR) was used to measure the atomic and chemical structure of the bioactive glass fibres. The fibres were ground into a powder and samples were recorded on a 7.0 T Varian/Chemagnetics InfinityPlus spectrometer operating at 59.62 MHz. A Bruker 7 mm probe was used with an MAS frequency of 5 kHz. Single pulse acquisition used a minimum of 230 transients (depending on sample amount) and were collected using a $\pi/2(^{29}Si)$ pulse of 4.3 μs with a 240 s recycle delay. In addition, 1H (71 kHz) radiofrequency (RF) decoupling was utilised during the acquisition. Deconvolution using OriginPro (OriginLab Corp.) isotropic chemical shift assignment was obtained from cross polarization measurements where a $\pi/2(^1H)$ of 3.5 μs and a contact time of 4 ms were optimised on the chemical shift reference (kaolinite, -92 ppm) in addition to the aforementioned decoupling.

2.3 Ion concentration

The ion concentration of the samples was measured to ascertain the concentration of ions released into the samples after the samples were placed in media conducted in 5 mM tris(hydroxymethyl)aminomethane, (TRIS) ($(CH_2OH)_3CNH_2$) buffer solution. In the study, 37.5 mg of 70S30C and 80S20C 3D cotton-wool-like fibres were immersed in 25 mL of TRIS buffer solution, and stored in an incubator shaker at 37 °C and 120 rpm according to the recommendations of Maçon *et al.* [42], all samples were produced in triplicate. A quantity of 0.5 mL of TRIS supernatant from each sample was collected at each time point (0.5, 1, 2, 3, 4, 8, 24, 48, 72, 96, and 168 h) for ICP-AES analysis and then 0.5 mL of fresh TRIS was added to keep the volume of the solution 25 mL. Prior to ICP analysis the 0.5 mL of sample removed was filtered with a 0.2

μm surfactant-free cellulose acetate filter (Corning, USA) and diluted with 4.5 mL of 2 M nitric acid. Samples were analysed using ICP-AES analysis at wavelengths of 317.9 nm (Ca) and 251.6 nm (Si). The pH of each sample was measured at each time point by a pH meter with a thermal probe (Oakton pH 11 meter with Oakton general purpose glass pH probe, Ag/AgCl electrode, and Oakton ATC probe) attached to it.

Table 1. Summary of the electrospinning solution properties and conditions used for each sol composition (100-X SiO₂, XCaO). All samples had the same voltage, flow rate, distance between the collector plate and viscosity. These were set up to be 10.0 kV, 0.4 mL/h, 100-120 mm and 0.12-0.20 Pa.s respectively.

Composition	TEOS [mL]	Nitric acid [mL]	Ethanol [mL]	Ca(NO ₃) ₂ •4H ₂ O [g]	Water [mL]
60S40C	6.2	1.5	3.1	4.3	0.0
70S30C	7.2	1.7	3.7	3.2	0.0
80S20C	8.2	1.9	4.2	2.1	0.3
90S10C	9.3	2.2	4.8	1.1	0.7
100S	7.2	1.7	3.7	-	

To understand the concentration of ions that cells will encounter, the ion concentration study was repeated in Dulbecco's Modified Eagle's Medium (DMEM). 37.5 mg of 70S30C and 80S20C 3D cotton-wool-like fibres were immersed in 25 mL of DMEM for 1-168 h.

The immersed samples were kept in an incubator at 36.5 °C to mimic physiological conditions. A quantity of 1 mL of DMEM supernatant was collected from each sample at each time point (4, 24, 72 and 168 h) and filtered for ICP-AES analysis and then 1 mL of fresh DMEM was added to each sample to keep the solution volume at 25 mL. Solid sample was collected from the filters, washed with PBS and rinsed with deionised water, and then dried at 40°C.

2.4 Cell studies

A CD-18CO cell line (ATCC, UK) was expanded in DMEM supplemented with 10% (v/v) foetal bovine serum (FBS), 1% penicillin streptomycin.

For cell culture, electrospun 70S30C and 80S20C cotton-wool-like scaffolds were preconditioned in DMEM for 4 h in an atmosphere of 37°C, 5% CO₂ and 21% O₂. After 4 h, the media was filtered using sterile 0.2 μm filters to remove all fibres from the dissolution product. The dissolution product was later supplemented with 10 % (v/v) FBS and 1% (v/v) penicillin/streptomycin. A Si control was also prepared from sodium metasilicate based upon the Si ppm value at 4 h in DMEM for the 70S30C and 80S20C, an average of these values was used to determine the quantity of sodium

metasilicate used for the Si Control sample. The Si control was conducted to understand if there was a correlation between increase the silica and increasing cell proliferation. The Si was used to compare to see if the bioactive glass (with its Si and Ca components) is beneficial or if it is the silica alone which provides the benefit. A DMEM vehicle control was also produced alongside the Si control. This was comprised of DMEM supplemented with 10 % (v/v) FBS and 1% (v/v) penicillin/streptomycin.

CD-18CO cells were suspended in DMEM at a concentration of 20, 000 cells per mL. 500 μL of cell suspension was added to each well in a 48 well-plate. After 18 h, the cell media was replaced with the supplemented 70S30C

and 80S20C dissolution product and DMEM was used as control. At day 1, 3 and 7, the supernatant was collected for further analysis and metabolic activity was tested using AlamarBlue[®] cell viability (Thermo Fisher Scientific, USA) reagent at a 10% concentration in each conditioned media. Following an incubation period of 4 h, 200 μL of the solution was transferred into a black well plate and fluorescence was measured at an excitation wavelength of 540 nm and an emission wavelength of 620 nm using a fluorescent microplate reader (Fluoroskan Ascent FL, Thermo Labsystems, UK).

Total DNA (tDNA) was quantified using the Hoechst H332598 fluorescent dye according to manufacture instructions (Total DNA, Sigma-Aldrich, UK). After the AlamarBlue[®] cell viability reagent was removed, the cells were washed with PBS and 200 μL of DI H₂O was added to each cell well. The cells were then lysed with 5 freeze-thaw cycles. A standard curve was prepared with calf thymus DNA and producing a DNA standard serial dilution from 0-10 μg DNA/mL. The assay was prepared at a dilution of 1:1000 with 10% assay buffer and water. 100 μL of the lysates and standard dilution was transferred to a black 96 well plate and 100 μL of the Hoechst dye assay was added. Fluorescence was measured using an excitation wavelength of 360 nm and an emission wavelength of 460 nm in a microplate reader (Fluoroskan Ascent FL, Thermo Labsystems, UK).

A Quantikine[®] ELISA kit was used to determine the amount of human vascular endothelial growth factor (VEGF) concentrations found in the cell supernatant. The assay was

performed according to the manufacturer's instructions (R&D Systems, UK) using 200 μL of supernatant collected at each time point. A standard dilution series of VEGF was additionally produced ranging from 0 pg/mL to 1000 pg mL^{-1} . The optical density was measured with a microplate reader (BioTek, Instruments) set to 450 nm and correction wavelength 562 nm.

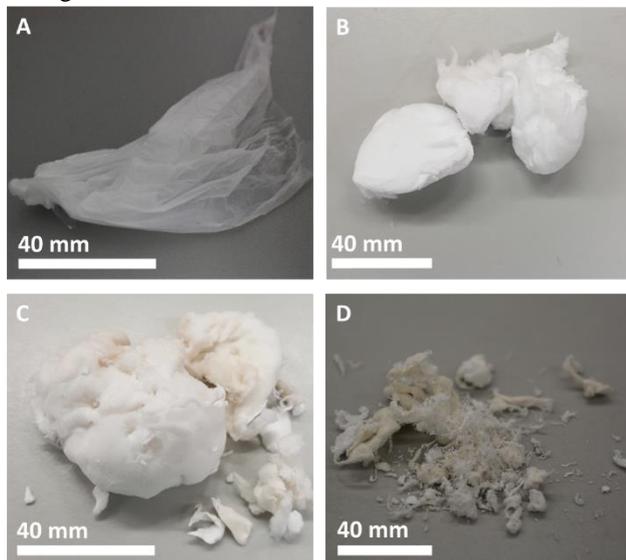


Figure 1. (A-D) Photographs showing the morphology of 70S30C sol electrospun at different relative humidity values (A) 27%, (B) 35%, (C) 55% and (D) 65%.

Metabolic activity was determined using an AlamarBlue assay according to manufacturer instructions (R&D Systems, UK) study, twelve repeats were used for each sample, the tDNA and VEGF ELISA used nine repeats over two experimental repeats.

The statistical significance of the results was compared using a Kruskal-Wallis test followed by Dunn's Multiple Comparison was used to detect significance of each composition with respect to DMEM at each time point (* $P < 0.05$, ** $P < 0.01$, *** $P < 0.001$, **** $P < 0.0001$).

3. Results and discussion

3.1 Processing windows and fibre morphology

To create the 3D cotton-wool-like fibrous structure, sols were prepared according to (Table 1), and equal amounts of binder and sol were mixed to fill a 10 mL syringe for electrospinning. The selected viscosity of the 70S30C sol/binder mixture was 0.069 ± 0.025 Pa.s.

Viscosities below 0.05 Pa.s prevented the formation of fibres, resulting in sol spraying from the end of the needle. Viscosities greater than 0.6 Pa.s failed to electrospin. The inclusion of the binder to obtain the desired viscosity range

made the successful electrospinning of the sol and formation of a fibremat easily repeatable.

The 70S30C/binder sol was electrospun at a range of relative humidity values (Figure 1). At 27% relative humidity, a 2D fibremat structure was produced (Figure 1A). Increasing the relative humidity to 35% (Figure 1B) and 55% relative

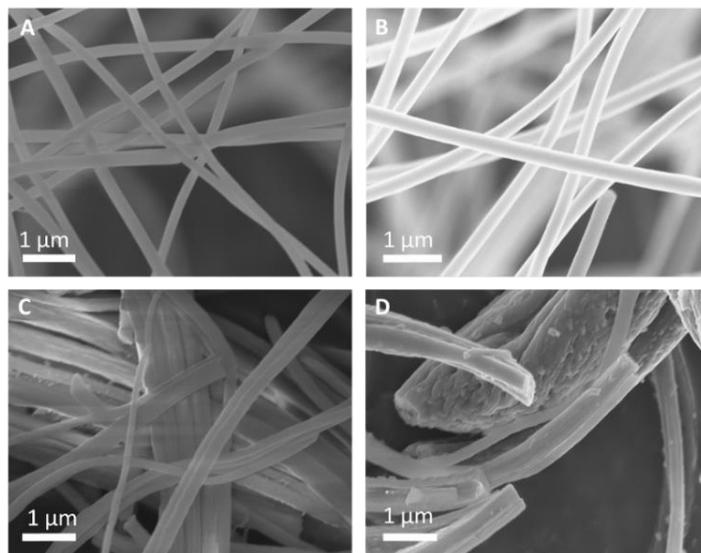


Figure 2. SEM images of the morphology of 70S30C electrospun samples. The relative humidity in each of the samples (A), (B), (C) and (D) was found to be (A) 27%, (B) 35%, (C) 55% and (D) 65%.

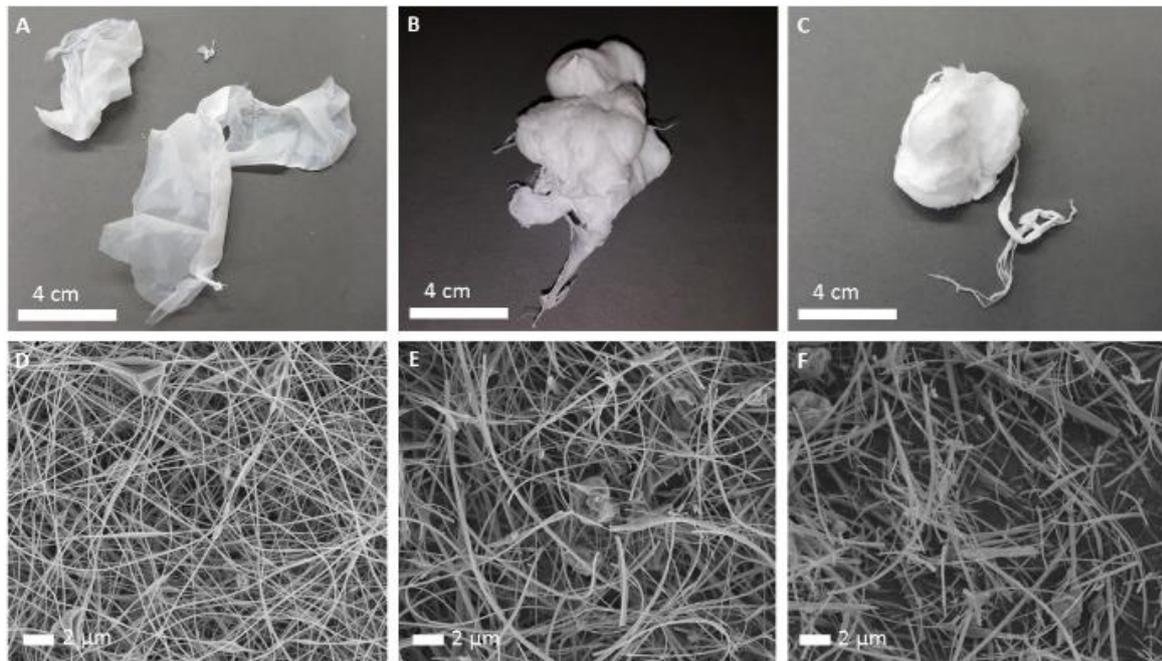
humidity (Figure 1C) produced a structure with a 3D cotton-wool-like appearance, similar to that attained by Poologasundarampillai *et al.* [25]. At a relative humidity of 65% (Figure 1D), the bioactive glass also had an appearance similar to that of a 3D structure. SEM images (Figure 2) of the fibre morphology present the fibres electrospun at the same relative humidity values. Fibres electrospun at 27% relative humidity (Figure 2A) were found to be long, uniform and straight without the presence of beads. Fibres electrospun at 35% and 55% relative humidity (Figure 2B and 2C) were much more undulating, creating the appearance of the 3D structure observed in Figure 1B and 1C. Figure 2D showed fibres which are non-uniform in length and diameter and beading was observed in the fibres.

The fibres electrospun at 55% and at 65% humidity (Figure 2C and 2D) contained fragments of broken fibre attached to the longer, formed fibres. This was much more prevalent in the structure electrospun at 65% humidity. Xie *et al.* [43] suggested that, for collagen fibres, short fibre lengths, such as those in Figure 2C and 2D, among other factors such as high fibre stiffness in collagen gels, prevented the spreading and proliferation of human mesenchymal stem cells (hMSCs). The small fragments of fibres are created by capillary instability, which converts a cylindrical jet into droplets of sol, which solidify as beads when they reach the collector [36].

Reneker *et al.* [36] reported them as a bead-on-a-string structure in polymers [44] due to capillary instability. In this case the beads were spaced at similar, repeating distances and patterns along the length of the fibres.

uniform fibres, similar to 70S30C fibres produced at 27% relative humidity (Figure 3A).

The 100S structures electrospun at 66% and 80% humidity



(Figure 3B, C) had the

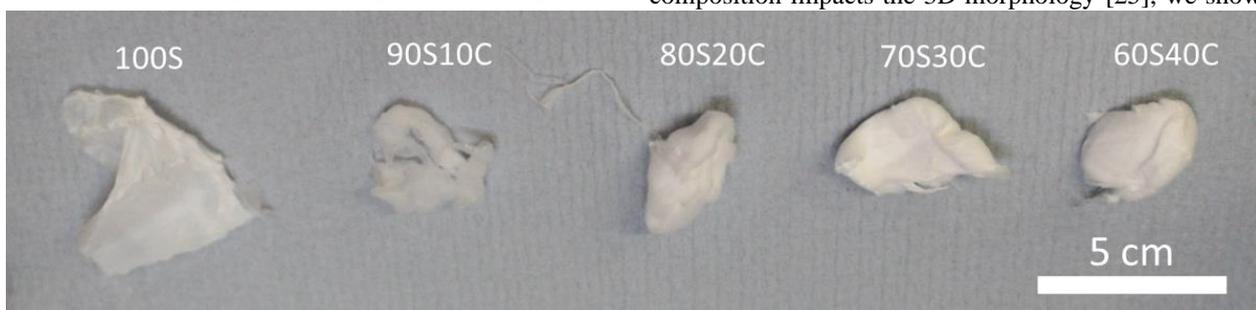
Figure 3. (A-C) Photographs showing the morphology of 100S (100 mol% SiO₂) electrospun cotton wool at relative humidity values of (A) 36%, (B) 66% and (C) 80%. (D-F): SEM images show the fibre morphology at relative humidity values of (D) 33% (E) 66% and (F) 80%.

The fragments formed by electrospinning the 70S30C and 100S sol-gel solutions, at relative humidity values above 55%, did not have this appearance. The beads here resulted from the sputtering of the sol-gel while electrospinning, resulting in the formation of bead-like drops of solid sol. Our working hypothesis is that the cotton wool-like entangled fibres are preferable for wound healing applications.

When the 100S bioactive glass samples were electrospun, changes in the fibre morphology were identified (Figure 3). The 100S sample electrospun at a relative humidity of 36% (Figure 3A) had the appearance of a 2D, fibremat-like structure, and SEM images show (Figure 3D) straight and

appearance of a 3D cotton-wool-like structure despite no presence of Ca²⁺ in the sol. The corresponding SEM images however show differences between the fibres of the three samples: the 100S sample electrospun at 36% relative humidity (Figure 3D) had long, uniform, undulating fibres with the presence of some small beads. The 100S sample electrospun at 66% (Figure 3E) had long, heterogeneous fibres with beads. The fibres electrospun at 80% relative humidity (Figure 3F) were short, with a large quantity of beads and some presence of whiskers.

Sol-gel bioactive glass compositions were electrospun with different quantities of Ca²⁺ added to produce a range of different compositions. Figure 4 shows variations in the morphology of the 60S40C, 70S30C, 80S20C, 90S10C and 100S bioactive glass compositions samples electrospun at similar values of relative humidity (30-40%). While this result confirms that the presence of calcium nitrate in the sol-gel composition impacts the 3D morphology [25], we show that



the relative humidity additionally determines the 2D or 3D nature of fibremat produced, even without calcium (Figure 3).

Figure 4. Photographs showing the morphology of bioactive glass fibres: 100S, 90S10C, 80S20C, 70S30C and 60S40C electrospun at a relative humidity of 30-40%.

3.2 Chemical composition

Sample	Composition		Surface Area [m ² g ⁻¹]	Fibre Diameter
	SiO ₂	CaO		[μm]
70S30C	68.1 ± 0.2	31.9 ± 0.2	43.1	0.34 ± 0.11
80S20C	77.4 ± 0.2	22.6 ± 0.2	40.0	0.21 ± 0.02

Table 2. Electrospun sol-gel bioactive glass composition (from ICP), fibre diameter (from SEM) and specific surface area (determined by N₂ sorption).

FTIR spectra were collected for each of the compositions. A main Si-O stretch vibration band between 1000 and 1070 cm⁻¹ was present in each of the samples (Figure 5A). There was no evidence of un-hydrolysed TEOS in any of the spectra, which indicates that all the hydrolysis and condensation reactions had taken place and that residues from the precursor reagents and binder were removed through heating. The FTIR spectra for 100S, 90S10C, 80S20C, 70S30C and 60S40C compositions show bands corresponding to Si-O at 1058.73, 1053.91, 1038.48, 1027.39 and 1037.03 cm⁻¹ and for Si-OH at 801.67, 801.28, 796.56, 787.78 and 942.54 cm⁻¹. An increase in the calcium content of bioactive glasses corresponded to a shift in wavenumber of the bands attributed to Si-O stretching [45]. The 60S40C and 70S30C spectra show two bands (1500 and 1410 cm⁻¹) which can be attributed to CO₃²⁻ groups [46]. This band was more intense in samples with higher calcium content, where the increased calcium content was responsible for the presence of carbonate, attributed to a carbonation process of the material due to the atmospheric CO₂ and moisture present [47].

XRD patterns (Figure 5B) of the electrospun sol-gel derived compositions of the 60S40C, 70S30C, 80S20C, 90S10C and 100S compositions all showed amorphous silica halos. Multiple peaks at 29°, 32°, 34°, 37° and 48° were present in the 60S40C composition, which can be attributed to crystalline phases of calcium silicates, such as Wollastonite. PANalytical X'Pert HighScore peak fitting software identified that these peaks could be either anorthic (00-014-0693) or orthorhombic (00-024-0234) calcium silicates. The 60S40C fibres therefore had inhomogeneous structures but these peaks were not found for samples with lower calcium content. Martinez *et al.* [46] showed sol-gel derived 60S40C discs gave

XRD diffraction patterns with amorphous halos without peaks. The peaks formed herein could be attributed to the higher surface area due to the fine fibres or the electrospinning process itself, with the electric field provoking a heterogeneous distribution of Ca²⁺ within the silicate.

60S40C, 70S30C, 80S20C and 90S10C powders, made without electrospinning, are all previously reported to form carbonate apatite layer formation following immersion in SBF [46]. For practical experimental reasons, certain compositions

were disregarded at this point. The relative humidity range that needed for electrospinning 90S10C sol to produce a 3D cotton-wool-like structure was difficult to achieve. The 60S40C sample was also dismissed due to the presence of peaks in the XRD indicating calcium silicates (Figure 5B). Therefore further studies were carried out on the 70S30C and 80S20C compositions only.

The measured compositions of the 80S20C and 70S30C bioactive glass samples were very similar to the nominal compositions (Table 2). The mean fibre diameter of the 80S20C and 70S30C were 0.21 ± 0.02 μm and 0.34 ± 0.11 μm respectively. Nitrogen sorption was carried out to test the specific surface areas (BET method) of the 70S30C and 80S20C fibres (Table 2), which were 43.1 m² g⁻¹ and 40.0 m² g⁻¹ respectively.

Data obtained by ²⁹Si MAS NMR enabled the identification of Qⁿ species defining the glass network, where Qⁿ corresponds to the structural species Si(OSi)_nO(R)_{4-n} (with R representing either Ca or H in the case of these glass formulations). The 70S30C and 80S20C glass compositions prepared at relative humidity values between 30-45% electrospinning conditions were compared, with both compositions exhibiting typical ²⁹Si chemical shifts averaging at -83.3, -92.3, -100.3 and -109.9 ppm which were assigned to Q¹, Q², Q³ and Q⁴ units, respectively (Table 3).

The broad ²⁹Si resonances characterising these Qⁿ species in Supplementary Figure S1 (SF1) are indicative of large structural distributions and disorder influencing the Si positions within the network. This observation is consistent with previous work by Lin *et al.* [48], who demonstrated that the ²⁹Si resonances corresponding to Qⁿ species in 70S30C glasses

Sample	Q ₁		Q ₂		Q ₃		Q ₄		D _c [%]	Connectivity
	δ [ppm]	I [%]								
70S30C	-84.42	10.00	-93.10	12.10	-100.56	17.90	-109.66	60.00	81.98	3.28
80S20C	-82.18	4.60	-91.44	16.70	-100.02	23.60	-110.15	55.00	82.20	3.29

heated to 700 °C were also influenced by large structural distributions and disorder in the SiO₂-CaO glass network [48].

Table 3. Relative Qⁿ unit distribution in the glass structure (from solid-state ²⁹Si MAS-NMR) and degree of condensation of the 70S30C and 80S20C bioactive glass samples.

The electrospun 70S30C and 80S20C samples were found to have a larger proportion Q⁴ (both ~82%) and a much smaller percentages of Q¹ and Q² species in comparison to conventionally prepared sol-gel 70S30C glass [48]. The degree of condensation, D_c, was calculated by equation [49]:

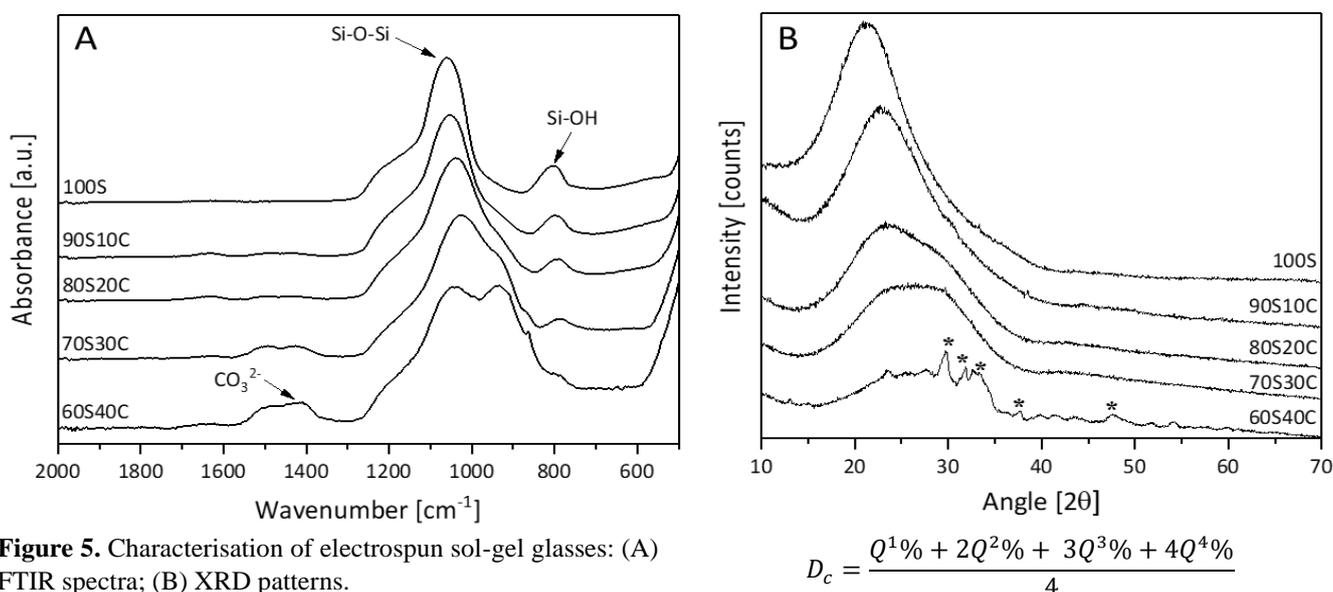


Figure 5. Characterisation of electrospun sol-gel glasses: (A) FTIR spectra; (B) XRD patterns.

$$D_c = \frac{Q^1\% + 2Q^2\% + 3Q^3\% + 4Q^4\%}{4}$$

D_c of the 70S30C and 80S20C glass compositions were similar (Table 3), and much higher than that found in the conventional Ca-containing 70S30C glasses [48], which exhibited a network connectivity of 2.62 and a D_c of 65%. The pre-stabilisation network connectivity of their glass composition (only dried at 130°C) was calculated to be 3.65 with a D_c of 91.3%. Ca was not incorporated into the network as a network modifier until a stabilisation temperature of ~450 °C was reached [48]. Poologasundarampillai *et al.* calculated the degree of condensation in their unstabilised (only dried) 70S30C fibres to be higher ~90%, suggesting that the calcium was not incorporated in the silica network [28].

While 70S30C and 80S20C fibres prepared herein had a reduced network connectivity and D_c compared with the 70S30C fibres of Poologasundarampillai *et al.* [25], they were not as low as those exhibited in monolithic sol-gel derived 70S30C composition.

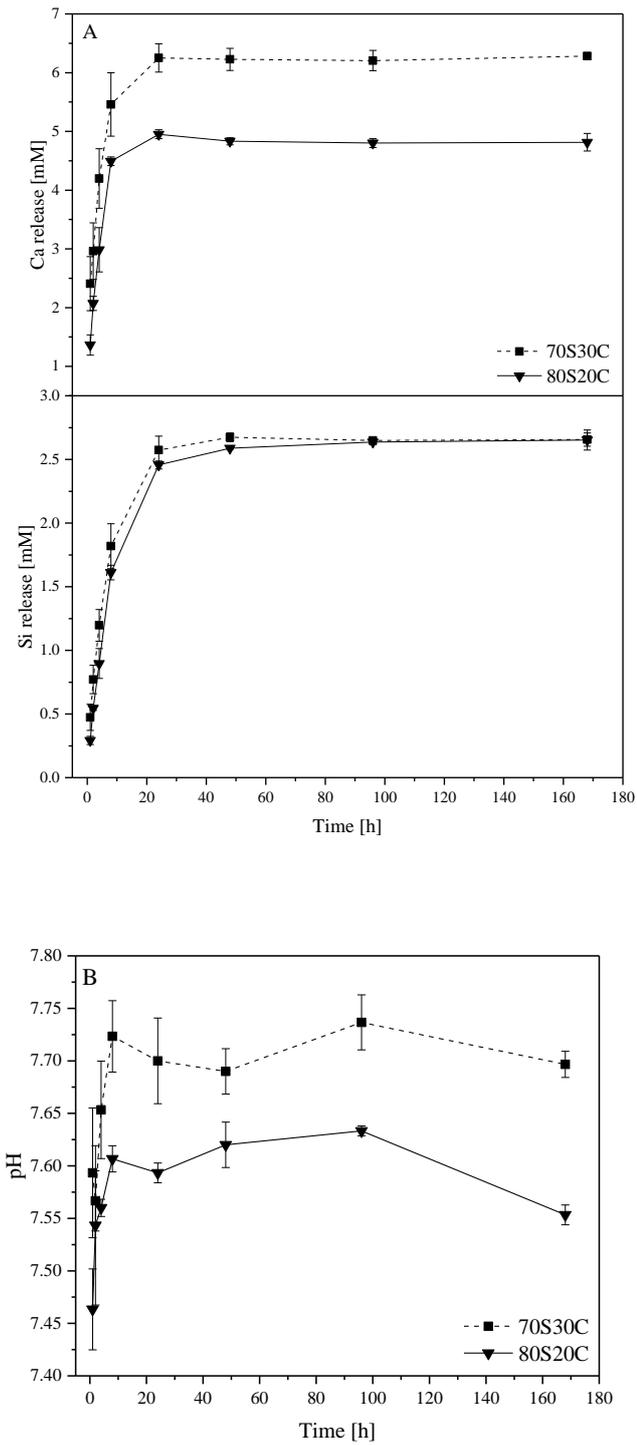


Figure 6. (A) Dissolution profiles (ICP-OES) of TRIS containing electrospun sol-gel glass samples as a function of time, (B) pH profile of the TRIS during glass dissolution.

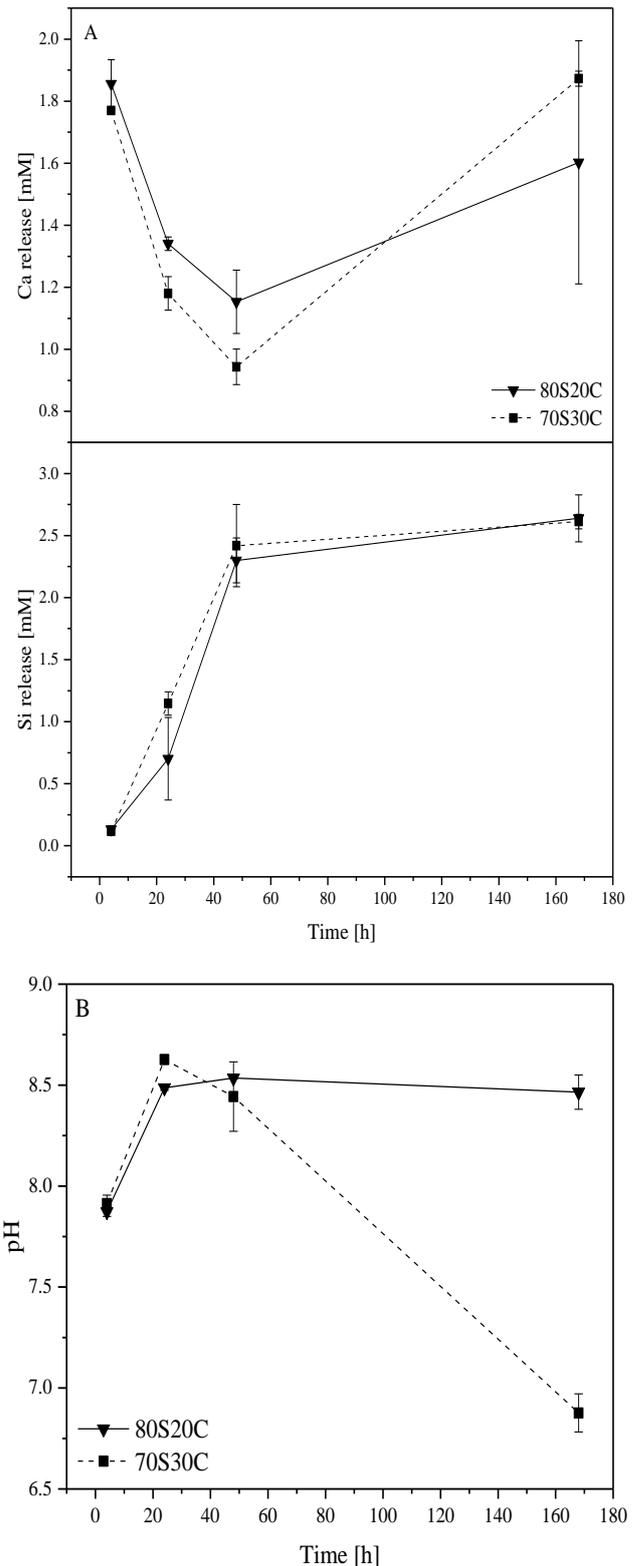


Figure 7. (A) Dissolution profiles (ICP-OES) of electrospun sol-gel glasses in DMEM at time points of 4 h, 1 day, 3 day and 7 day, (B) pH profile of DMEM during glass dissolution.

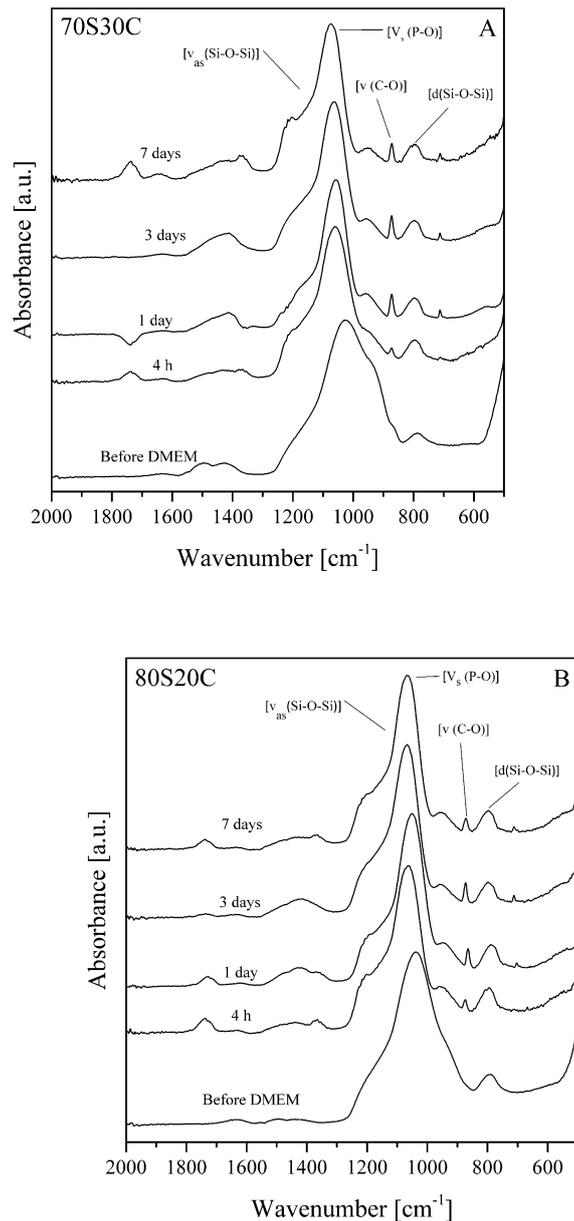


Figure 8. FTIR spectra of 70S30C (A) and 80S20C (B) electrospun sol-gel glasses before and after being soaked in DMEM for 7 days.

3.3 Dissolution studies

A high proportion of the calcium ions and soluble silica were released to the surrounding solution within 50 h of soaking in TRIS (Figure 6A). The concentration of soluble silica was similar between the two compositions but the 70S30C glass produced a greater concentration of calcium ions compared to the 80S30C composition. This was expected as the 70S30C contained more calcium, and both of the glasses induced a pH change in TRIS (Figure 6B). After 168 h, TRIS containing 70S30C had a pH of ~ 7.70 and that containing 80S20C was ~ 7.55 . The pH increased [48],[49] during

bioactive glass degradation as result of the exchange between modifier cations, such as Ca^{2+} , from the glass and protons from the dissolution media [50]. During dissolution studies, changes in the pH therefore provide indication of the amount of ion exchange occurring [51].

The dissolution study in DMEM (Figure 7A) shows similar concentration profiles for the two compositions for silicon and for calcium. Si levels were similar to what they were for the TRIS immersion, however the Ca decreased for the first three days, before increasing again by day seven. Ca concentration at day 7 for the 80S20C sample was ~ 1.60 mM and Kulesz-Martin *et al.* [52] showed that dermal fibroblasts grew best at concentrations of 1.4 mM Ca^{2+} . The dissolution in DMEM (Figure 7) shows that the 70S30C and 80S20C compositions were nearest this value at the 4 h time point and therefore dissolution product at this time point was determined to be the most appropriate for use in the cellular cell studies.

The decrease in concentration of calcium ions in SBF is characteristic of the species being involved in both the dissolution and precipitation (such as hydroxycarbonate apatite formation) [53]. Figure 7B shows a higher pH value for both in DMEM, at all time points, compared with that shown in TRIS (Figure 5B).

FTIR spectra of the samples after seven days immersion (Figure 8) identified an additional band at ~ 872 cm^{-1} , which was attributed to the C-O stretching peak of the CO_2^{3-} group typically associated with the formation of hydroxycarbonate apatite on glasses [54]. Other bands associated with C-O at ~ 1400 cm^{-1} were however not identified [55].

XRD patterns (Figure 9) for the 70S30C fibres contained peaks after 4 h in DMEM. While some of these peaks could be attributed to hydroxyapatite, which displays peaks at $\sim 28^\circ$, $\sim 32^\circ$, $\sim 39^\circ$, $\sim 43^\circ$ and $\sim 49^\circ$, hydroxyapatite also has additional peaks which were not found in the XRD pattern presented in Figure 9. Crucially, the primary peak associated with hydroxyapatite ($2\theta = \sim 32^\circ$) [56] was not present in the patterns produced from this study. Peak fitting software (PANalytical X'Pert HighScore Plus) and comparisons in literature showed that the pattern of peaks fitted most loosely with Calcite [56].

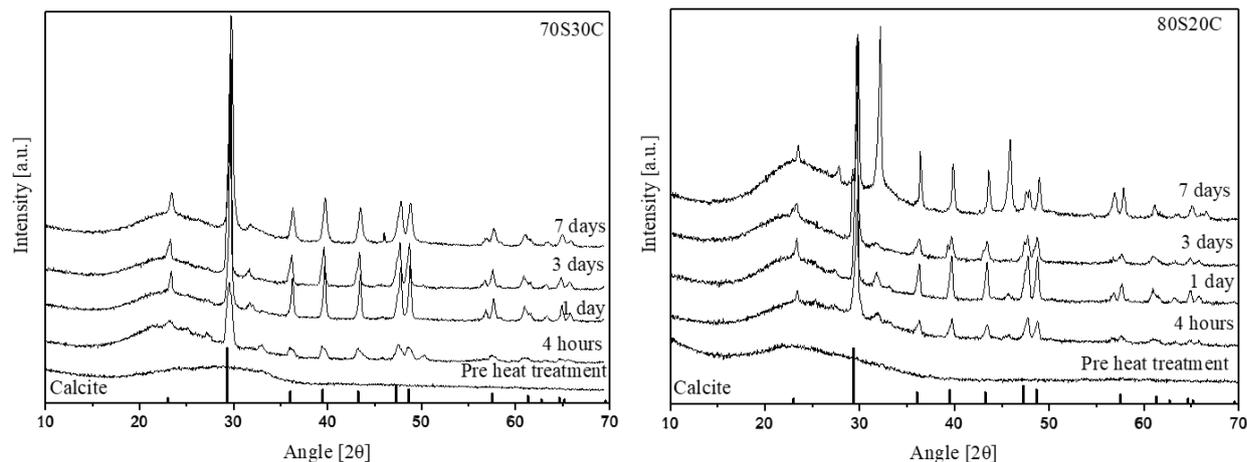


Figure 9. XRD patterns of 70S30C and 80S20C electrospun sol-gel glasses before and after being immersed in DMEM for 7 days. 70S30C and 80S20C patterns contain peaks attributed to calcite (ICDD reference 00-005-0586). The 80S20C pattern additionally shows the presence of halite (NaCl, ICDD reference of 00-001-0994).

These peaks were identified at 4 h, 1 day, 3 days and 7 days immersion of 70S30C fibres in DMEM, indicating the formation of calcite happened very quickly after immersion. The XRD patterns for 80S20C shows a similar array of peaks at each of the 4 h, 1 day 3 day and 7 day time points, indicating that the formation of calcite was similar. The day 7 pattern of the 80S20C sample additionally showed peaks at 33°, 46° and 56° which were attributed to halite (cubic sodium chloride).

3.4 Cell studies

Angiogenesis, which is critical for tissue regeneration, involves the sprouting of endothelial cells from pre-existing vessels, a process which is regulated by the local tissue balance of inhibitory molecules (such as thrombospondin), including growth factors, most notably VEGF.

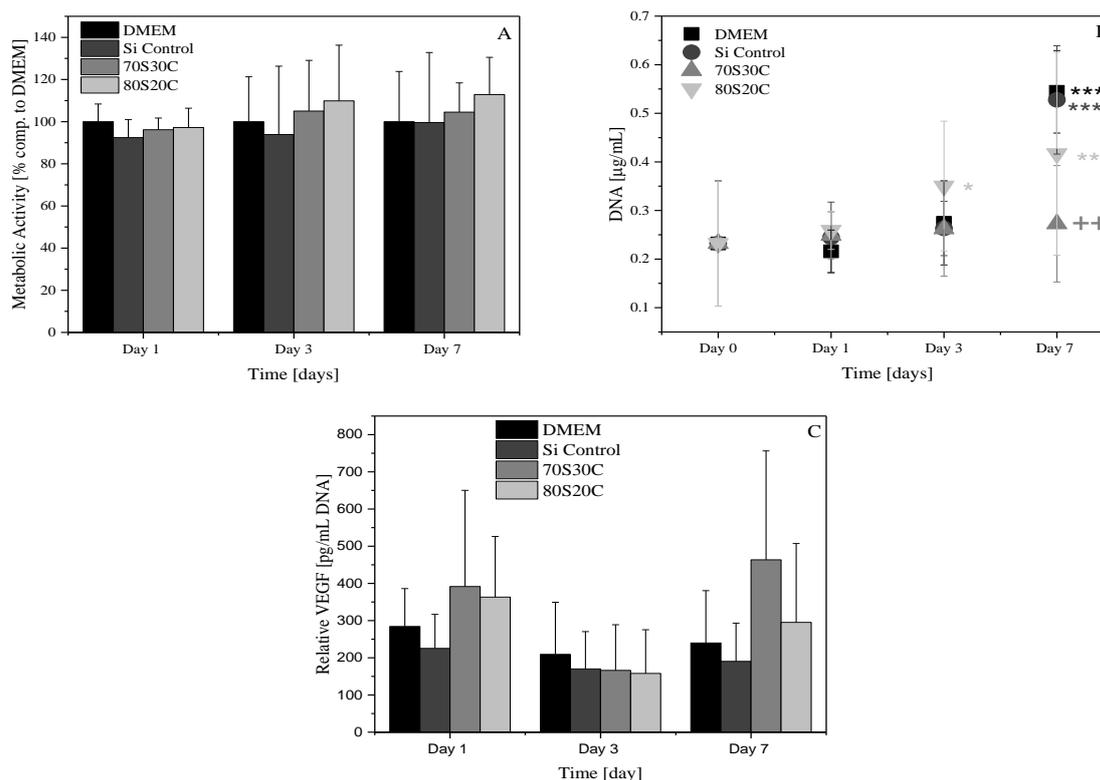
A large number of *in vivo* studies have demonstrated that VEGF plays a central role in angiogenesis, although studies on the *in vivo* induction of angiogenesis by bioactive glasses are rare, such as Midha *et al.* [57]. *In vitro* studies carried out by Day *et al.* showed that dissolution products from bioactive glasses stimulate fibroblasts to secrete VEGF [19].

In addition to release of ions, glass particle size in silicate bioactive glasses is also as an important determinant of cellular growth factor production *in vitro*, as smaller particles (1-2 μm) are more effective than bigger particles (2-3.15 μm) for inducing VEGF secretion by human CD-18CO fibroblasts [58]. An aim herein was to determine whether the increased

surface area of the sub-micrometre scale electrospun fibres induced greater growth factor expression.

The AlamarBlue® assay (Figure 10A) revealed no significant differences in metabolic activity between Si or vehicle only DMEM control cultures and those exposed to 70S30C and 80S20C bioactive glass dissolution products at days 1, 3 or 7.

The tDNA values from control samples in cultures from CD-18CO fibroblast cells were consistent with linear cell growth on days 1 - 3 and exponential growth by day 7. Cultures exposed to either 70S30C or 80S20C dissolution products were not significantly different from those of vehicle-only (or Si control) cultures at day 1 (Figure 10B) when compared with the pre-exposed tDNA at day 0. At day 3, tDNA values were significantly higher in 80S20C samples than for vehicle only or Si controls (**p* < 0.05). By day 7, the tDNA content from both vehicle-control and Si-control cultures was significantly higher than that of cultures exposed



to either 80S20C (* $p < 0.05$) or 70S30C dissolution products, consistent with an inhibitory effect on cell growth.

Statistical analysis was also carried out to compare each of the conditions against the DMEM control vehicle condition at each time point. Day 1 and day 7 showed no significant difference in any of the conditions compared with the DMEM control vehicle. At day 7, the 70S30C condition had a significantly lower tDNA compared to the DMEM control sample. This reduction in significance indicates that the 70S30C composition inhibited the tDNA of the cell cultures by day 7 of exposure.

In order to compare absolute VEGF levels across cultures exposed to different bioactive glass compositions, we normalized VEGF values in each culture to tDNA content for that sample (Fig. 10C). There were no significant differences in the amount of VEGF/pg DNA between any of the sample conditions (NS; $p < 0.05$) or between similar samples across time points (NS; $p > 0.05$). This suggests that the presence of the 3D cotton-wool-like bioactive glass compositions did not inhibit the production of VEGF. Previous studies show that the presence of certain ions can increase VEGF production in a variety of cell types [59]. Therefore, the compositional flexibility offered by the sol-gel derived, bioactive glass 3D structure could provide use as a vehicle to deliver additional therapeutic ions into the wound bed in a structure which mimics that of the extracellular matrix. These additional ions may be needed to provoke a true angiogenic effect.

Figure 10. The 70S30C and 80S20C bioactive glass fiber compositions did not affect metabolic activity on CD-18CO cultures compared to controls over a seven day study. (A) Metabolic activity [% comp. DMEM Control] of HDF cultures exposed to DMEM (DMEM, FBS and P/S) containing the dissolution product of bioactive glasses 70S30C and 80S20C, the Si and DMEM control samples after 1 day, 3 day and 7 day. A Kruskal-Wallis test followed by Dunn's Multiple Comparison were used to detect statistical significance of the samples. Metabolic activity of samples were normalized to respective DMEM controls at each time point ($n=12$, NS; * $P < 0.05$); (B) Proliferation [$\mu\text{g/mL}$] of HDF cultures exposed to DMEM containing the dissolution product of bioactive glass glasses 70S30C and 80S20C, the Si control samples and DMEM after 1 day, 3 day and 7 day. A Kruskal-Wallis test followed by Dunn's Multiple Comparison was used to detect statistical significance of each composition with respect to the pre-exposed tDNA at day 0 ($n=9$, * $P < 0.05$, ** $P < 0.01$, *** $P < 0.001$, **** $P < 0.0001$). A second Kruskal-Wallis test followed by Dunn's Multiple Comparison was used to detect statistical significance of each composition with respect to the result of the DMEM at each time point. ($n=9$, + $P < 0.05$, ++ $P < 0.01$, +++ $P < 0.001$, ++++ $P < 0.0001$); (C) Normalised VEGF production [pg/ μL] of HDF cultures exposed to DMEM containing the dissolution product of bioactive glass glasses 70S30C and 80S20C, the Si control samples and DMEM after 1 day, 3 day and 7 day normalized

to each sample's tDNA [$\mu\text{g/mL}$] value. A Kruskal-Wallis test followed by Dunn's Multiple Comparison was used to detect statistical significance of each composition with respect to DMEM at each time point.

3. Conclusions

An inorganic sol-gel derived material was electrospun from its sol, producing 3D cotton-wool-like structures with long and homogenous fibre diameters for use as a wound healing scaffold. The 3D cotton-wool-like morphology was formed not solely due to the calcium content of the sol, as previously thought, but was also dependent on the relative humidity in which it is electrospun. This meant that calcium-free silica sols could also be spun into 3D architectures. A humidity range for spinning each of the five-sol compositions into 3D architectures was identified. The 90S10C and 60S40C compositions were discarded due to laboratory humidity constraints and the presence of peaks in XRD scans. Two compositions (70S30C and 80S20C) were studied further to determine their suitability for use in wound healing applications. From ^{29}Si MAS NMR data, electrospun 70S30C and 80S20C glass fibres had increased network connectivity in comparison conventional sol-gel derived 70S30C glasses. Studies in TRIS buffer and DMEM both lead to the rapid release of silicon and calcium. Cell studies using a CD-18CO cell line were carried out on the dissolution product of 80S20C and 70S30C samples and were both found not to inhibit the expression of VEGF from CD-18CO cells over a period of seven days.

4. Acknowledgements

The authors would like to thank EPSRC for the funding through the CDT in Advanced Characterisation of Materials (CDT-ACM) (EP/L015277/1). JVH acknowledges support for the solid state NMR instrumentation at Warwick used in this research which was funded by EPSRC and the University of Warwick, with additional partial funding being provided through the Birmingham Science City AM1 and AM2 projects which were supported by Advantage West Midlands (AWM) and the European Regional Development Fund (ERDF). Raw data is available on request from rdm-enquiries@imperial.ac.uk.

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6. Supplementary Figures

Figure S1. Solid-state ^{29}Si MAS-NMR data of electrospun sol-gel glass 70S30C and 80S20C samples.

