

Flash spark plasma sintering of HfB₂ ceramics without pre-sintering

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1 Flash Spark Plasma Sintering of HfB₂ Ceramics without Pre-
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29
30 **Abstract**
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32 Strong HfB₂ powder green bodies with compressive strengths exceeding 20
33 MPa have been produced by incorporating 1 wt% of glassy carbon derived
34 from a pyrolyzed phenolic resin binder. Starting from the green body after
35 debinding in argon, non-textured HfB₂ ceramics (diameter ~20 mm and height
36 ~4 mm) with a relative density of 95.1% were achieved using flash spark
37 plasma sintering (FSPS) with a discharge time of 20 s, peak power of 19 kW
38 and applied pressure of ~16 MPa. The lack of a need for a pre-sintering step
39 resulted in energy savings compared to the densification of HfB₂ ceramics by
40 FSPS.
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55 **Keywords**
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59 Borides; Flash Spark Plasma Sintering; Sintering; Microstructure
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1 Sintering is an essential step in the production of almost all ceramics; however,
2 conventionally it is both time and energy intensive. A significant saving in both
3 factors has been recently realised via “flash sintering (FS)” [1]. This process
4 involves application of an electric field across a ceramic green body to
5 promote Joule heating. Under constant voltage mode, a flash event (a sharp
6 increase in the temperature of the sample) can be triggered when the furnace
7 temperature exceeds a threshold value. This allows ceramics to be densified
8 in a matter of seconds [2-3]. A thermal runaway-based model can predict the
9 heat balance; the flash occurs when the Joule heating significantly exceeds
10 the heat dissipated to the nearby environment [2-6].

25 Previous investigations, however, have recognised that an inhomogeneous
26 temperature distribution exists within samples during flash sintering [7]. Such
27 inhomogeneity is reduced when the green body has a dog-bone shape and
28 this special geometry is also chosen to maintain a uniform field across the
29 sample, as well as a volumetric power dissipation [7-8]. To reduce the need
30 for such a specific geometry requirement, with a cross section limited to a few
31 square millimetres and sample weight to typically below 1 g, a modified
32 configuration for FS has been implemented using a spark plasma sintering
33 (SPS) furnace [9-13]. Various ceramics, including SiC [11], ZrB₂ [9], ZrO₂ [12]
34 and B₄C [13], with diameters up to 60 mm have been consolidated by the
35 flash spark plasma sintering (FSPS) in less than a minute.

53 A green body for a typical FSPS process has to be strong enough to withstand
54 the thermal stresses resulting from the rapid heating, whilst the compressive
55 strength also has to exceed the contact force of the SPS apparatus. The latter

1 is required to ensure good electrical contact and so avoid electrical arcing /
2 overheating at the contacts of the punch-die-ram assembly. Pre-sintering of
3 the pellets at a relatively low temperature and pressure has been commonly
4 adopted to obtain a suitably strong body for FSPS; the relative density of the
5 ceramic body after pre-sintering is normally between 45% and 63.6% [9,12].
6 For example, ZrB₂ and yttria-stabilized zirconia green bodies were pre-
7 sintered by SPS at 1600°C / 16 MPa / 20 min [9] and 750°C / 30 MPa / 1 min
8 [12], respectively. This need clearly reduces the potential advantages of FSPS
9 in terms of overall time and energy savings.
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23 In the present work, the green strength of the ceramics was increased by the
24 incorporation of 1 wt% of glassy carbon into the ceramic matrix. HfB₂, a well-
25 known ceramic that is hard-to-densify [14-16], e.g. hot pressing at 2200°C /
26 30 MPa for 60 mins only yielded 85.8% of the theoretical density [14], was
27 used to demonstrate the potential of this method during FSPS.
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36 HfB₂ (D₅₀ = 3.7 μm, C: 0.1 wt%, O: 0.2 wt%, Treibacher, Althofen, Austria) and
37 2 wt% B₄C (D₅₀ = 0.5 μm, H.C. Starck, grade BF-12, Goslar, Germany)
38 powders were ball mixed together in ethanol for 24 h. In some batches, 2.43
39 wt% phenolic resin (PR, Hexion Inc., Rotterdam, The Netherlands) was added
40 into the slurry before mixing [14,15]. The slurry was subsequently dried at
41 40°C for 0.5 h to remove all the solvent; simultaneous magnetic stirring was
42 employed in order to avoid any possible sedimentation during drying. The
43 resulting dried powder cake was crushed into fine powders, which were then
44 die pressed at 200 MPa into pellets with a diameter of 20 mm and height of 5
45 mm using a steel mould preheated to 40°C to soften the phenolic resin. The
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green body was pyrolyzed at 600°C for 2 h in argon using a heating rate of 1°C/min to convert the phenolic resin into carbon. The compressive strength of the green bodies before and after pyrolysis at 600°C (denoted as P-RT and P-600) were measured using 5 mm $\Phi \times 10$ mm height cylindrical samples under a load displacement rate of 0.5 mm/min.

FSPS runs were conducted in a commercial SPS furnace (FCT HPD 25; FCT Systeme GmbH, Rauenstein, Germany) under vacuum. The general details are given elsewhere [11]. The as-pyrolyzed, pressed HfB₂ green bodies were wrapped with a layer of 4 mm thick graphite felt to avoid the effect of radiative heat loss as much as possible and to minimize the temperature difference across the sample during heating. Each sample was placed between the two graphite punches and a pre-load of 5 kN was applied in order to achieve good electrical contact. After some initial trials, a maximum power of between 10-20 kW, depending on the discharging time and the relative pre-set power of the SPS, was chosen. Two discharging times, 20 and 35 s, were finally adopted whilst the corresponding relative power was set at 100% and 80%, respectively. During FSPS the temperature was measured using two pyrometers, one was located at the top of the furnace and focused on the graphite punch at a distance of 4 mm from the sample, whilst the second was located on the side of the furnace and focused on the graphite felt covering the sample.

The density of the resultant samples was measured using the Archimedes' method in water and converted to a relative density using a theoretical value calculated using the rule of mixtures for appropriate volume proportions of

1 HfB₂, B₄C and C. Scanning electron microscopy, SEM (JEOL 7000, Tokyo,
2 Japan), was used to observe the fracture surfaces of the green bodies as well
3 as polished surfaces of the sintered bodies. A dual beam FIB / FEG-SEM
4 equipped with an EBSD detector was employed to look for the evidence of
5 crystallographic texture in the flash sintered samples. Orientation maps were
6 obtained from the surfaces perpendicular to the loading direction and the step
7 size for the EBSD scans was 0.2 μm. A misorientation angle larger than 5°
8 was used to determine the mean grain size and its distribution in the sintered
9 ceramics. A 200 kV JEOL TEM (2100F) equipped with electron energy loss
10 spectrometry (EELS, GIF 963, Gatan, USA) was used to detect the features
11 and locations of any as-pyrolyzed carbon in the HfB₂ matrix.
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27 Without phenolic resin additions, HfB₂ – 2 wt% B₄C powders could not be
28 pressed into crack-free pellets. According to the manufacturer, the phenolic
29 resin used in this work had a pyrolytic carbon (PyC) yield of 41 wt% after
30 pyrolysis. Using this, the calculated composition in the green body after
31 pyrolysis was HfB₂ – 2 wt%B₄C – 1 wt%C. Average compressive strengths of
32 9.6±0.3 and 20.2±0.7 MPa were measured for P-RT and P-600, respectively.
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66 The behaviours of P-RT and P-600 (the relative density of both was ~55%)
67 under uniaxial compression are compared in Fig. 1a. P-RT shows classic
68 ductile behaviour, which explains the better shaping behaviour of the powder
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1 mixture when PR was present. In contrast, no apparent plastic deformation
2 prior to failure was observed in P-600 indicating that these samples fractured
3 in a brittle manner with an enhanced compressive strength value.
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7 As may be observed within the backscattered SEM image of P-RT in Fig. 1b,
8 there were a number of grains with slightly darker contrast that were well
9 distributed within the HfB₂ matrix. The average HfB₂ primary particle size from
10 Fig.1b is 1.25 μm, significantly smaller than the D₅₀ value (3.7 μm) provided
11 by the manufacturer, indicating agglomeration existed in the raw powders.
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13 Considering that the mean atomic numbers of both PR and B₄C are much
14 smaller than for HfB₂, these darker grains must be either PR or B₄C. The EDS
15 analysis results shown in Fig. 1d & e revealed that point A is PR whilst point B
16 is B₄C. The oxygen and carbon signals in PR, Fig. 1d, are much stronger than
17 those in B₄C, Fig. 1e. After pyrolysis, flakes with thickness less than 100 nm
18 were observed occasionally on the fracture surface of P-600; highlighted by
19 arrows in Fig. 1c. It seems that these nanostructures were derived from the
20 pyrolysed PR. Previous work has shown that the densification of HfB₂ by hot
21 pressing starts above 1650°C [14], therefore, the strength improvement in P-
22 600 compared to P-RT cannot be ascribed to partial sintering since the heat
23 treatment only occurred at 600°C. As a result, the different behaviour under
24 compression between P-RT and P-600 must be due to the conversion from
25 PR to PyC. To illuminate this further, the bonding characteristics between the
26 PyC and HfB₂ particles were further investigated using TEM to explain the
27 strengthening mechanism found in the P-600 samples.
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1 Fig. 2a clearly shows two HfB₂ particles connected by a nanosheet stack.
2 Although these sheets exhibit a layered structure, no clear lattice fringes could
3 be observed from their basal plane, Fig. 2b. In an enlarged view of the
4 junction between the nanosheets and HfB₂ grains, Fig. 2c, some particles less
5 than 5 nm in size but with good crystallinity were found. Some of the
6 interplanar spacings of these particles are shown in Fig. 2c; some fit with B₄C
7 (4.50, 3.78 and 2.37 Å), whilst the others (3.36 and 3.37 Å) are similar to the
8 (0002) spacing of graphite (3.38 Å). The nanosheets not only bridge the
9 adjacent particles, arrowed in Fig. 2a and 2e, they also coat the surface of
10 some of the HfB₂ grains, as revealed by the HRTEM image in Fig. 2d.
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26 According to the EELS maps of carbon and boron, Fig. 2g & h, a layer only
27 containing carbon, region I in Fig. 2h, was observed to cover the edge of an
28 HfB₂ particle, arrowed in Fig. 2f. Between the carbon layer and the HfB₂
29 particle, region III in Fig. 2h, an ~50 nm transition zone was observed
30 composed of B and C, region II in Fig. 2h. The bonding between the carbon
31 and HfB₂ appears to be quite strong since the carbon coating seems to be
32 chemically anchored.
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45 Ionisation edges corresponding to C-k were also found in the EELS patterns
46 of the nanosheets, which show two peaks centred at about 287 and 298 eV,
47 respectively, Fig. 2i. The first sharp peak belongs to the transitions of the π*
48 molecular orbital due to the existence of sp² bonded carbon, whilst the second
49 more intense peak at 298 eV was induced by the σ* orbitals transitions of
50 carbon [17]. The slight broadening of the latter peak is a characteristic for
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glassy carbon; in amorphous carbon it is normally significantly broader compared to glassy carbon [17]. Glassy carbon is a non-graphitic carbon mainly constituted of sp^2 bonded atoms. It can exist as clusters or ribbons of graphite-like sheets interspersed with regions of amorphous carbon [18]. Both the EELS and HRTEM observations in Fig. 2 suggest that the PyC flakes derived from the PR are likely to be glassy carbon, which has a relatively high modulus of ~30 GPa and flexural strength of ~190 MPa [19]. If correct, both the bridging and coating of glassy carbon nanosheets on the HfB_2 particles will have been responsible for the higher strength in the P-600 compared to the P-RT.

The change in the sintering parameters during FSPS, including electrical resistance, displacement, electric current, SPS heating power and resulting front pyrometer temperature, are shown in Fig. 3. When a sample with an electrical resistance of R is subjected to a voltage of V , the sample heating rate (dT/dt) may be expressed as Eq.1 [5].

$$\frac{dT}{dt} = \frac{1}{C} \left[\frac{V^2}{R(T)} - \varepsilon\sigma(T^4 - T_f^4) \right] \quad (1)$$

Where C and ε represent the sample's heat capacity and emissivity, σ is the Stefan–Boltzmann constant, T and T_f are the temperature for the sample surface and furnace, respectively. It should be noted that Eq.1 is not strictly applicable to current FSPS set-up as it neglects the Joule heating of the carbon felt and its low thermal conductivity. Unlike most oxides [2,6], the electrical resistance of metal diborides such as HfB_2 increases linearly with temperature [20]. Therefore, dT/dt will decrease when a constant voltage is

1 applied to HfB₂-based samples and flash heating (an exponential increase in
2 heating rate) will not occur without an additional sharp increase in the voltage.
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7 As has been discussed previously [11], at the beginning of FSPS the current
8 preferentially passed through the carbon felt, as shown by the initial
9 temperature increase recorded by the front pyrometer (T_f), Fig. 3a. This
10 effectively caused the samples to be preheated. Subsequently, the current
11 began to flow through the sample, leading to a drop in T_f and a simultaneous
12 decrease in electrical resistivity for the circuit, Fig. 3a. When the discharge
13 began, a simultaneous increase in displacement and the top temperature, T_t,
14 was observed, Fig. 3a. The densification behaviour of HfB₂ ceramics during
15 FSPS was strongly depending on the flash; Fig. 3b shows that a shorter
16 discharge time and higher relative power was helpful in causing a more
17 obvious flash and, consequently, an enhanced shrinkage rate. The porosity of
18 the HfB₂ decreased to <5 vol% when discharge occurred in 20 s, whilst when
19 it took 35 s the porosity did not decrease below 20 vol%, Fig. 3b.
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41 Most of the HfB₂ samples sintered using FSPS were crack-free in this work,
42 Fig. 3e. A mean relative density of 95.1% was obtained for the ceramics
43 densified using a discharge time of 20 s and a relative power of 100%.
44 Residual pores can be clearly seen, they are visible as dark regions with
45 slightly bright edges from charging and are highlighted in some cases by
46 arrows in Fig. 4a. Other areas with a dark contrast indicate B₄C and carbon
47 grains. Both pores and B₄C grains are mainly located at the grain boundaries,
48 as shown in the grain orientation map, Fig. 4b. The latter also confirms that no
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1 texturing was generated during densification. The average HfB₂ grain size in
2 the sintered bodies was ~4 μm, Fig. 4c, slightly larger than that of the reported
3 average particle size of 3.7 μm. This value is much larger than the primary
4 particle size of HfB₂ in the green body determined by SEM, ~1.25 μm,
5 indicating densification and grain growth still compete with each other during
6 FSPS as in conventional sintering. The final microstructure had equiaxed
7 morphology with an average aspect ratio of ~1.2 for the HfB₂ grains, Fig. 4d.
8 The number of neighbouring grains calculated, Fig. 4e, fits a normal
9 distribution and the average value is close to six. Although the HfB₂ ceramics
10 were rapidly densified in only 20 s during FSPS, the resultant microstructures
11 still resembled those sintered more conventionally. Based on the calculations
12 in Table 1, by utilizing a proper design and optimization process for the green
13 forming, the energy cost of FSPS could be further reduced by up to 80%,
14 compared to FSPS using normal pre-heating.

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36 In summary, starting with coarse HfB₂ powder, >95% dense HfB₂ ceramics
37 were achieved by FSPS in a discharge time of just 20 s, using 2 wt% B₄C and
38 1 wt% C as sintering aids. This work also demonstrates that pre-sintering,
39 which is routinely adopted for flash SPS, is not necessary if ceramic green
40 bodies strengthened by the presence of 1 wt% of glassy carbon are used.

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43 Science Research Council entitled 'Materials Systems for Extreme
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Figure and Table Captions

Figure Captions

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Figure 1a. Compressive strain-stress curve for P-RT and P-600. b. & c. fracture surfaces of P-RT and P-600, respectively. d. & e. EDS analysis of point A and B highlighted in Fig. 1b.

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Figure 2a. TEM and b. – d. HRTEM images of P-600. g & h EELS C-k and B-k edge maps of local area shown in e. & f. in P-600. The EELS carbon k-edge fingerprint spectrum for glassy carbon and B₄C are compared in i.

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Figure 3 SPS output curves recorded during FSPS of P-600. a. The set relative power is 80% with a discharging time of 35 s, b. the densification behaviour comparison of two samples with different FSPS parameters, discharging time 35 s, relative power 80% and discharging time 20 s, relative

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power 100%. (c) The digital images of the set-up for FSPS in a conventional SPS furnace, (d) P-600 before and (e) after FSPS.

Figure 4a. The polished surface of a flash-SPSed P-600 sample produced using a discharge time of 20 s and 100% of the relative power. The orientation map from EBSD is shown in b. (Note that a. & b. were collected from slightly different areas of the same sample since the FIB used did not have a BSE detector and the image was tilted to 70°; it is extremely difficult to find the same area for EBSD using a different SEM). c. – e. show the distribution of the HfB₂ initial powder particle size determined by dynamic light scattering (DLS) and SEM, and the subsequent ceramic grain size, aspect ratio and number of neighbouring grains calculated based on the EBSD measurements respectively.

Table Captions

Table 1 Preliminary calculations for the energy consumption used when densifying HfB₂-based ceramics using conventional SPS, FSPS and modified FSPS in this work.

Figure(1)

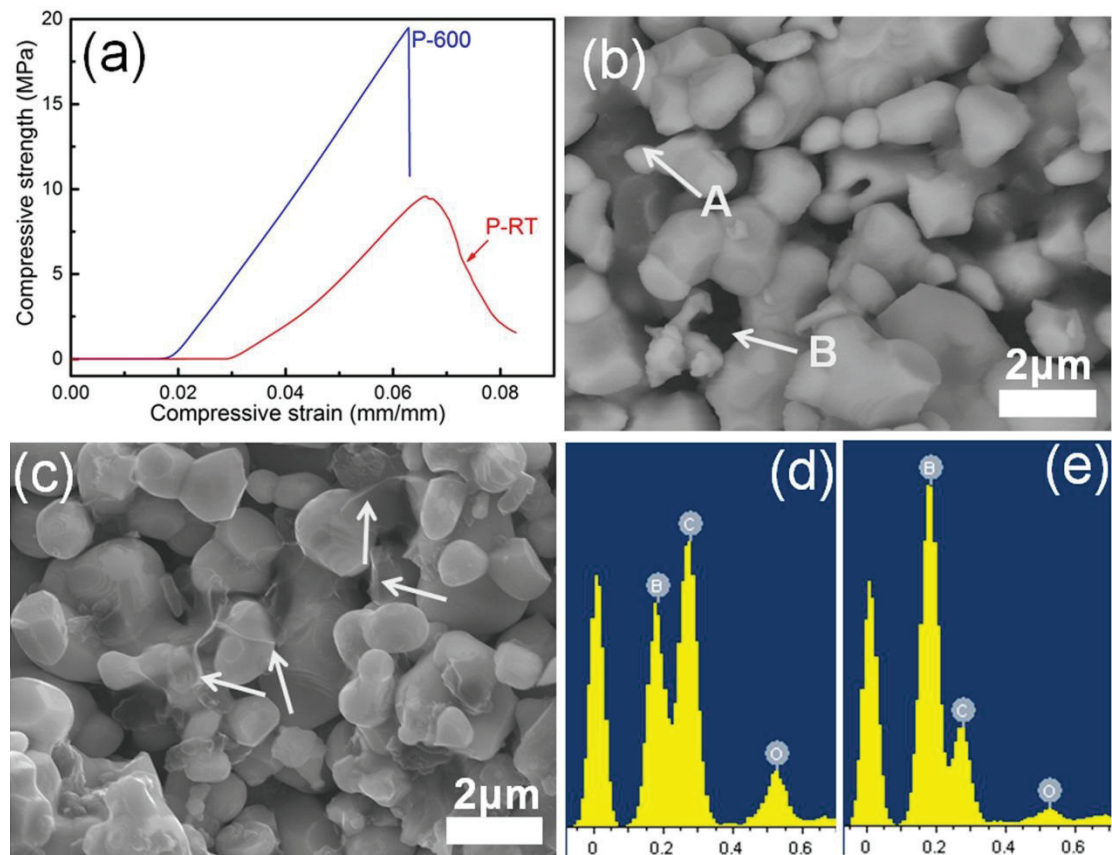


Figure 1a. Compressive strain-stress curve for P-RT and P-600. b. & c. fracture surfaces of P-RT and P-600, respectively. d. & e. EDS analysis of point A and B highlighted in Fig. 1b.

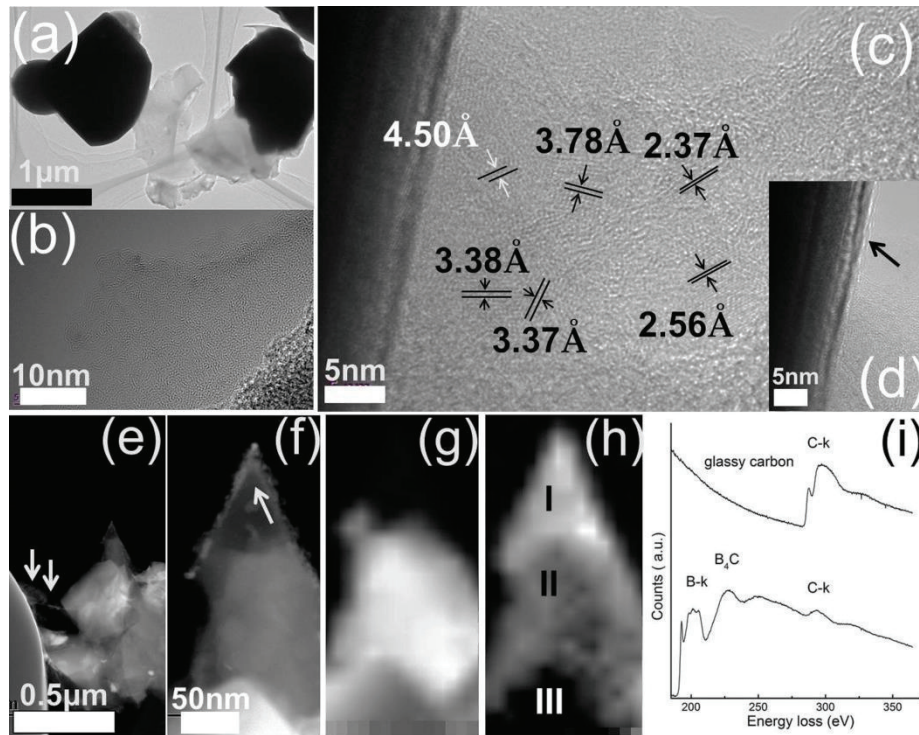


Figure 2a. TEM and b. – d. HRTEM images of P-600. EELS C-k and B-k edge maps of local area shown in e. & f. in P-600 are shown in g. & h. The EELS carbon k-edge fingerprint spectrum for glassy carbon and B₄C are compared in i.

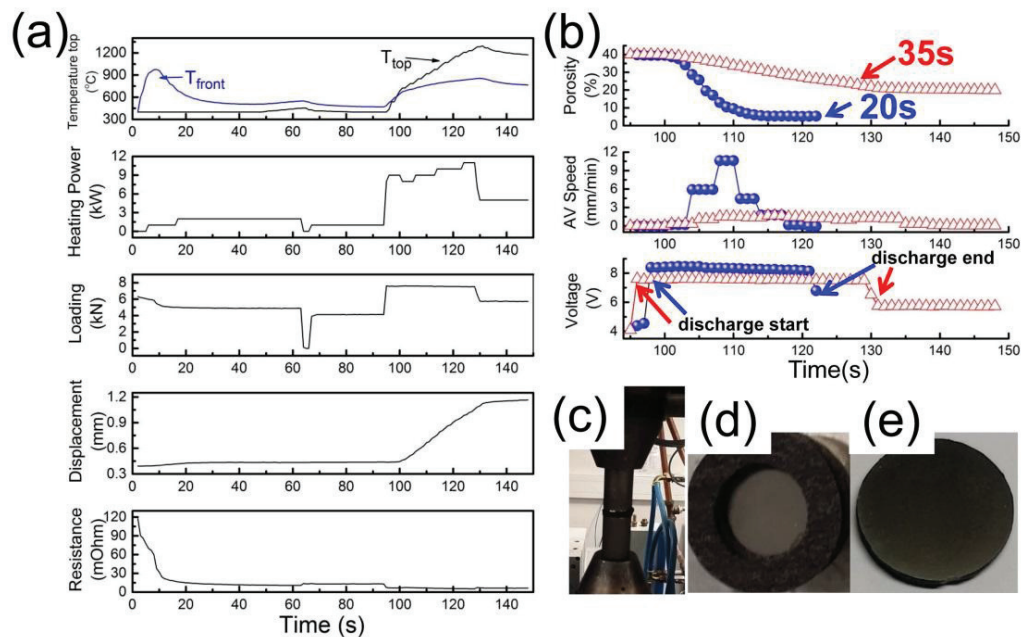


Figure 3 SPS output curves recorded during FSPS of P-600. a. The set relative power is 80% with a discharging time of 35 s, b. the densification behaviour comparison of two samples with different FSPS parameters, discharging time 35 s, relative power 80% and discharging time 20 s, relative power 100%. The digital images of the set-up for FSPS in a conventional SPS furnace, P-600 before and after FSPS are shown in c. – e., respectively.

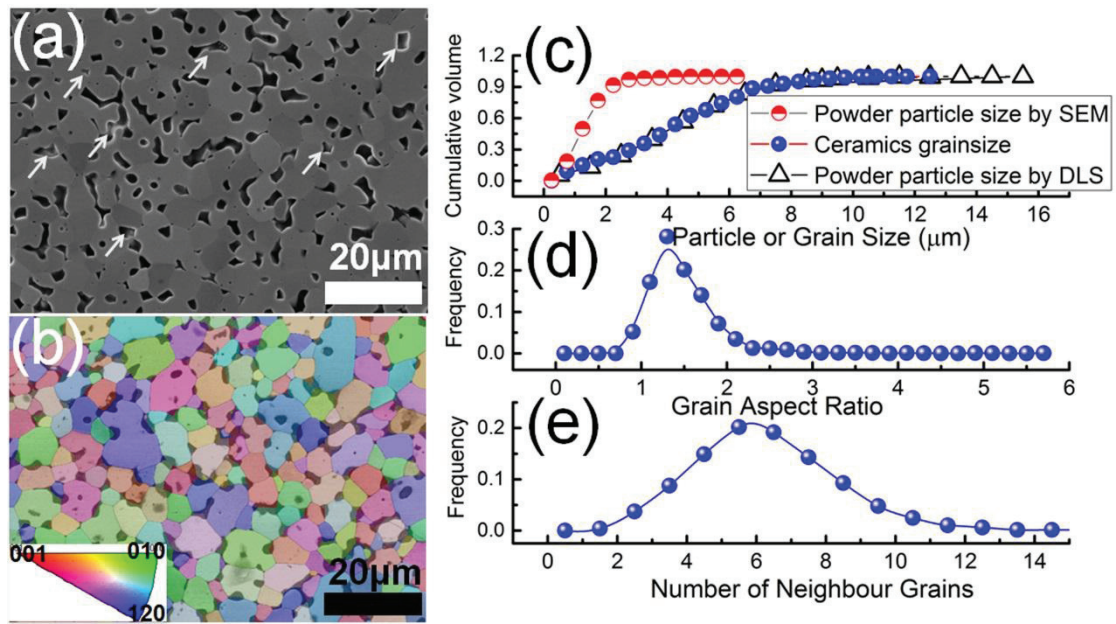


Figure 4a. The polished surface of a flash-SPSed P-600 sample produced using a discharge time of 20 s and 100% of the relative power. The orientation map from EBSD is shown in b. (Note that a. & b. were collected from slightly different areas of the same sample since the FIB used did not have a BSE detector and the image was tilted to 70°; it is extremely difficult to find the same area for EBSD using a different SEM). c. – e. show the distribution of the HfB₂ initial powder particle size determined by dynamic light scattering (DLS) and SEM, and the subsequent ceramic grain size, aspect ratio and number of neighbouring grains calculated based on the EBSD measurements respectively.

Table 1 Preliminary calculations for the energy consumption used when densifying HfB₂-based ceramics using conventional SPS, FSPS and modified FSPS in this work.

	Conventional SPS	Flash SPS (FSPS)	Modified flash SPS in this work
Sintering conditions	Heating from RT to 2100°C over 20 mins followed by a hold at 2100°C for 30 mins	Holding at 1600°C for 20 mins by SPS plus FSPS at 20 kW for 20 s	Holding at 600°C for 2 h plus FSPS at 20 kW for 20 s
Approx. energy consumption per 100 pellets with the same size as used in this work / kWh	$7 \times 100 = 700$	$(2 + 0.2) \times 100 = 220$	$(2 \times 10/100 + 0.2) \times 100 = 40$