

# Development of Fluorinated Analogues of Perhexiline with Improved Pharmacokinetic Properties and Retained Efficacy

Madhani, Melanie

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# Development of Fluorinated Analogues of Perhexiline with Improved Pharmacokinetic Properties and Retained Efficacy

Chih-Chung Tseng,<sup>a</sup> Hannah Noordali,<sup>b</sup> Monica Sani,<sup>c,d</sup> Melanie Madhani,<sup>b</sup> Denis M. Grant,<sup>e</sup> Michael P. Frenneaux,<sup>a,f</sup> Matteo Zanda,<sup>a,c,\*</sup> Iain R. Greig<sup>a,\*</sup>

<sup>a</sup> University of Aberdeen, Kosterlitz Centre for Therapeutics, Foresterhill, Aberdeen, AB25 2ZD (UK)

<sup>b</sup> University of Birmingham, Institute of Cardiovascular Sciences, Edgbaston, Birmingham, B15 2TT (UK)

<sup>c</sup> C.N.R. – I.C.R.M., via Mancinelli 7, 20131 Milan (Italy)

<sup>d</sup> KemoTech s.r.l., Parco Scientifico della Sardegna, Edificio 3, Loc. Piscinamanna, 09010 Pula, CA, Italy

<sup>e</sup> University of Toronto, Department of Pharmacology & Toxicology, Toronto M5S 1A8 (Canada)

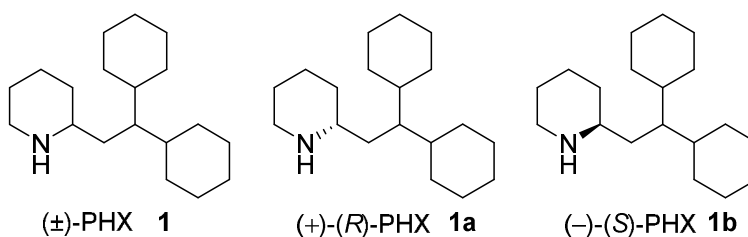
<sup>f</sup> University of East Anglia, Norwich Medical School, Norwich NR4 7UQ

## ABSTRACT

We designed and synthesized perhexiline analogues that have the same therapeutic profile as the parent cardiovascular drug, but lacking its metabolic liability associated with CYP2D6 metabolism. Cycloalkyl perhexiline analogues **6a-j** were found to be unsuitable for further development as they retained a pharmacokinetic profile very similar to that shown by the parent compound. Multi-step synthesis of perhexiline analogues incorporating fluorine atoms onto the cyclohexyl ring(s) provided a range of different fluoro-perhexiline analogues. Of these, analogues **50** (4,4-*gem*-difluoro) and **62** (4,4,4',4'-tetrafluoro) were highly stable and showed greatly-reduced susceptibility to CYP2D6-mediated metabolism. In vitro efficacy studies demonstrated that a number of derivatives retained acceptable potency against CPT-1. Having the best balance of properties, **50** was selected for further evaluation. Like perhexiline, it was shown to be selectively concentrated in the myocardium and, using the Langendorff model, to be effective in improving both cardiac contractility and relaxation when challenged with high fat buffer.

## INTRODUCTION

The therapeutic potential of perhexiline **1** (PHX, Fig. 1) in the treatment of cardiovascular disorders has been known for almost 50 years<sup>1,2</sup> and the drug was introduced in 1973 for the treatment of angina,<sup>3</sup> for which it has considerable efficacy and indeed continues to be used in some centres for patients with refractory angina. Perhexiline shifts myocardial metabolism from fatty acid to carbohydrate utilisation, thereby increasing metabolic efficiency (i.e. it produces relatively more ATP for a given oxygen consumption). By increasing myocardial efficiency, perhexiline has the potential to treat a range of disorders in which impaired cardiac energetics or oxygen deficiency play a role. In addition to angina, perhexiline has been investigated and used off-label for the treatment of heart failure, giving unprecedented improvements in  $VO_2$  max, quality of life improvement and left-ventricular ejection fraction;<sup>4</sup> for acute coronary syndromes;<sup>5</sup> inoperable aortic stenosis<sup>6</sup> and hypertrophic cardiomyopathy.<sup>7</sup> The latter study has led the FDA to award orphan drug status to perhexiline. Furthermore, by its inhibitory action on fatty acid  $\beta$ -oxidation, it has been hypothesised that perhexiline may have potential in anti-neoplastic therapies to target tumour metabolism, either alone or in combination with existing therapeutic regimens (e.g., to increase susceptibility to chemotherapy).<sup>8,9</sup>



**Figure 1.** Perhexiline **1** and its enantiomers.<sup>10</sup>

However, perhexiline was removed from most markets by 1988, due to side-effects including hepatotoxicity and neurotoxicity.<sup>11</sup> The side-effects leading to withdrawal of the drug were

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3 due to sustained high plasma levels of the drug, in a small subgroup of patients, as a result  
4 of variable (apparently unpredictable) metabolism. This led to phospholipid accumulation in  
5 the liver and peripheral nerves, due to complete blockade of mitochondrial long chain fatty  
6 acid uptake. Metabolism of perhexiline, hydroxylation at the 4-positions of either of the  
7 cyclohexyl rings, is mediated primarily by CYP2D6. CYP2D6 is polymorphic, giving  
8 substantial inter-individual variation in the rate of metabolism of drugs upon which it acts. An  
9 approximate 100-fold range is seen in the rate of perhexiline clearance and the required  
10 dosage ranges from 100 mg/week to 500 mg/day.<sup>12</sup> These issues can be avoided with close  
11 patient monitoring,<sup>11</sup> but such monitoring is not practical outside well-controlled patient  
12 groups, and thus it is unlikely that perhexiline can ever again be used in a wider patient  
13 population. Recent studies investigating the potential of using pure enantiomers **1a,b** (Fig. 1)  
14 of perhexiline have suggested differences in myocardial uptake, clearance, route of  
15 metabolism and, potentially, activity and toxicity; however, it is not yet clear as to whether  
16 these data will translate to therapeutic relevance.<sup>13,14,15</sup>

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32 Perhexiline has been described as a drug of “low potency, low specificity, and low selectivity”  
33 and as having actions on a diverse range of molecular targets, some of which remain to be  
34 properly defined, which contribute to its therapeutic benefit in heart failure (HF).<sup>16</sup> However,  
35 at present, the therapeutic action of perhexiline is generally still thought to be due to  
36 inhibition of CPT-1 and to a lesser extent CPT-2.<sup>11,17</sup> CPT is an enzyme responsible for  
37 mediating mitochondrial uptake of long chain fatty acids, by coupling them to carnitine; the  
38 acylated carnitine is then taken into the cell by a carnitine acyl-carnitine translocase.  
39 Preventing free fatty acid transport across mitochondrial cell membranes causes a  
40 corresponding shift to greater carbohydrate usage and an increase in myocardial efficiency  
41 (10 - 40%; more in some studies,<sup>11</sup> in terms of ATP generated per unit of oxygen). This  
42 oxygen-sparing effect is of particular significance in the treatment of angina and heart failure,  
43 allowing cardiac output to remain constant in spite of reduced cardiac oxygen extraction.  
44 Perhexiline inhibits CPT-1 with modest potency ( $IC_{50}$  = 77  $\mu$ M and 148  $\mu$ M in rat cardiac and  
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3 liver mitochondria) and inhibits CPT-2 (79  $\mu\text{M}$ ) and fatty acid oxidation ( $\beta$ -oxidation) in  
4 hepatocytes at 22  $\mu\text{M}$ , resulting in increased lactate and glucose utilisation.<sup>18</sup> Whilst the  
5 effective contribution of CPT inhibition has also been questioned,<sup>15,16,18</sup> a further argument in  
6 support of the importance of CPT-1 inhibition as the mechanism of action, in spite of the  
7 apparent lack of potency, is the tendency for perhexiline to become concentrated in tissues,  
8 including the myocardium.<sup>18</sup> Thus the concentration in the vicinity of mitochondrial CPT-1 is  
9 thought to exceed the modest  $\text{IC}_{50}$ . This characteristic is attributed to its amphipathic  
10 nature,<sup>11</sup> though this has also been highlighted as the source of the proposed diverse  
11 actions of perhexiline<sup>16</sup> and may also complicate attempts to determine the role of  
12 stereoselectivity in the therapeutic activity of perhexiline.  
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25 In spite of the therapeutic potential and the well-understood toxicity issues, which led to its  
26 removal from clinical usage throughout most of the world, few attempts have been made to  
27 modify the chemical structure of perhexiline. To date, only a small number of reports on  
28 derivatives have been published<sup>19,20</sup> and derivatives with improved properties (relative to  
29 perhexiline) have not been reported. It is clear that compounds which have the same  
30 therapeutic profile as perhexiline, but which lack the metabolic liability associated with  
31 CYP2D6 metabolism, could be highly effective therapeutic agents for the treatment of a  
32 range of cardiovascular conditions. In the absence of complex and variable  
33 pharmacokinetics, such a compound would not be restricted to limited patient populations or  
34 to use in a clinical environment, and thus could benefit a wider patient population for a range  
35 of cardiovascular diseases. In this report we describe the successful blocking of CYP2D6-  
36 mediated metabolism by addition of fluorine atoms at the sites of metabolism and  
37 demonstrate that these compounds have broadly equivalent potency to that of perhexiline  
38 against CPT-1.  
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## 55 RESULTS AND DISCUSSION

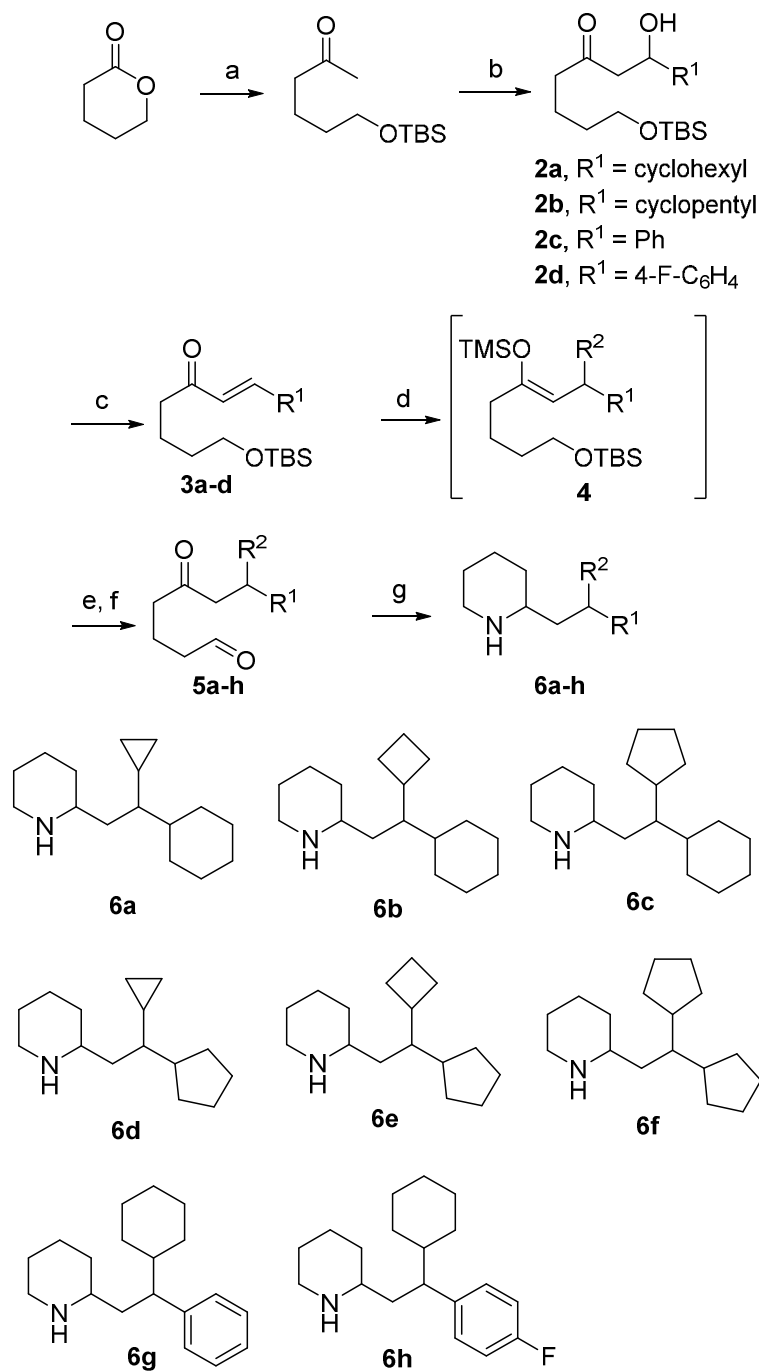
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3 **Chemistry and in vitro pharmacokinetics.** In the absence of satisfactory hypotheses  
4 concerning how perhexiline exerts its therapeutic and toxicological actions, there is no clear  
5 guidance as to how the properties could be optimised. For example, there is no evidence  
6 that a more potent CPT-1 inhibitor would necessarily be preferred. Thus, our primary aim for  
7 this programme was to develop compounds having the same activity profile as perhexiline,  
8 but lacking the CYP2D6 liability. The obvious route to achieve this was the development of  
9 compounds which were structurally as similar to perhexiline as possible, in the hope that the  
10 therapeutic effects could be decoupled from the metabolic liability. There have been  
11 surprisingly few attempts made previously to generate even simple perhexiline analogues.  
12 These have included *N*-substituted derivatives,<sup>20</sup> derivatives in which the piperidine has  
13 been replaced by other amine-bearing groups,<sup>19</sup> derivatives in which the carbon separating  
14 the two cyclohexyl groups has been substituted with a carbinol group<sup>21</sup> and the synthesis of  
15 a deuterated derivative has also been described.<sup>22</sup>

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As perhexiline can be hydroxylated on either cyclohexyl ring by CYP2D6, our initial focus was on the synthesis of analogues in which one or both of the cyclohexyl rings had been replaced by cycloalkyl rings of similar size (Scheme 1).

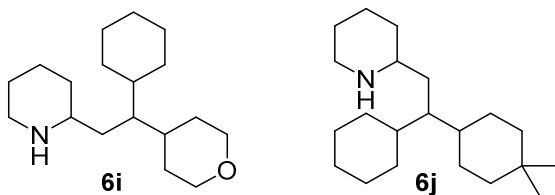
$\delta$ -Valerolactone was treated with methyl lithium and the resulting 6-hydroxy-hexan-2-one was transformed into the corresponding TBS-ether. The latter was treated with LDA and subjected to aldol reaction with aldehydes affording the  $\beta$ -hydroxy-ketones **2a-d**. The crotonic adducts **3a-d** were obtained by DBU-promoted  $\beta$ -hydroxy elimination of the intermediate mesylates. Introduction of the second group of the perhexiline-like structure was achieved by Michael reaction upon treatment of **3** with the corresponding cuprates,<sup>23</sup> then the intermediate silyl enol ethers **4** were deprotected and the primary carbinol group was oxidized with Dess-Martin periodinane (DMP) to provide the aldehydes **5a-h** (62 – 81% yields). The latter were subjected to reductive amination/cyclization<sup>24</sup> affording the small library of perhexiline analogues **6a-h**, which were isolated as mixtures of stereoisomers in good to excellent yields (78 – 91%).



**Scheme 1.** Synthesis of cycloalkyl perhexiline analogues **6a-h**. Reagents and conditions: (a) MeLi, -78 °C, then TBSCl and imidazole (93%); (b) LDA, -78 °C, then R<sup>2</sup>CHO; (c) MsCl, TEA then DBU; (d) RM (M = Mg or Li), Cu(I)X (see experimental section), TMSCl, -78 °C; (e) PTSA, MeOH-CH<sub>2</sub>Cl<sub>2</sub>; (f) DMP (62-81% over steps d-f); (g) NH<sub>4</sub>OAc, NaBH<sub>3</sub>CN, MeOH (78-91%).



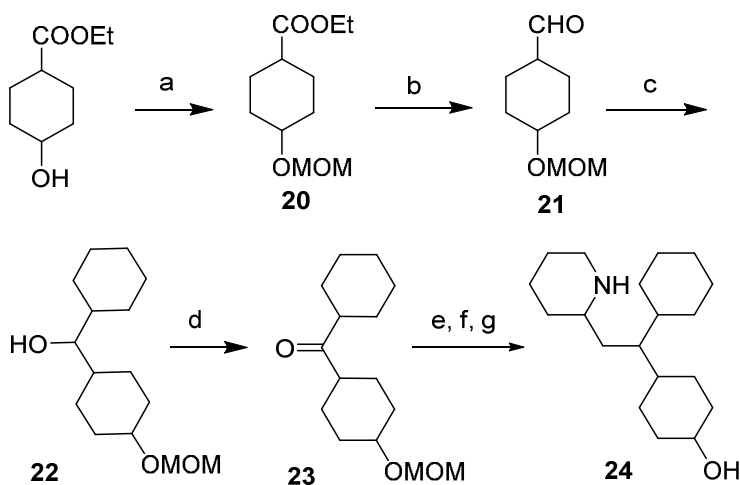
The THP analogue **6i** and the *gem*-dimethylcyclohexyl analogue **6j** (Fig. 2) were prepared following different synthetic routes (see Supporting Information for details).



**Figure 2.**

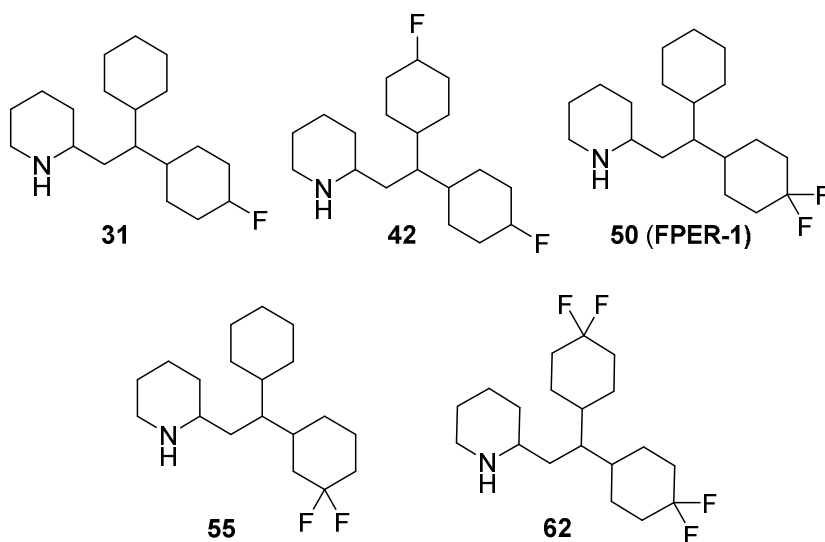
Disappointingly, in human and rat liver microsomal (HLM / RLM) stability studies, compounds **6a-j** all retained a stability profile very similar to that shown by the parent compound. We therefore turned our attention to the addition of fluorine atoms onto the cyclohexyl ring, as another potential route towards either physically blocking the site of metabolism, or to deactivating the ring towards oxidation by reduction of electron density.<sup>25</sup>

With the view of introducing fluorine by dehydroxyfluorination, we first synthesised 4-hydroxy-perhexiline **24** (Scheme 2) from ethyl 4-hydroxycyclohexane-carboxylate.



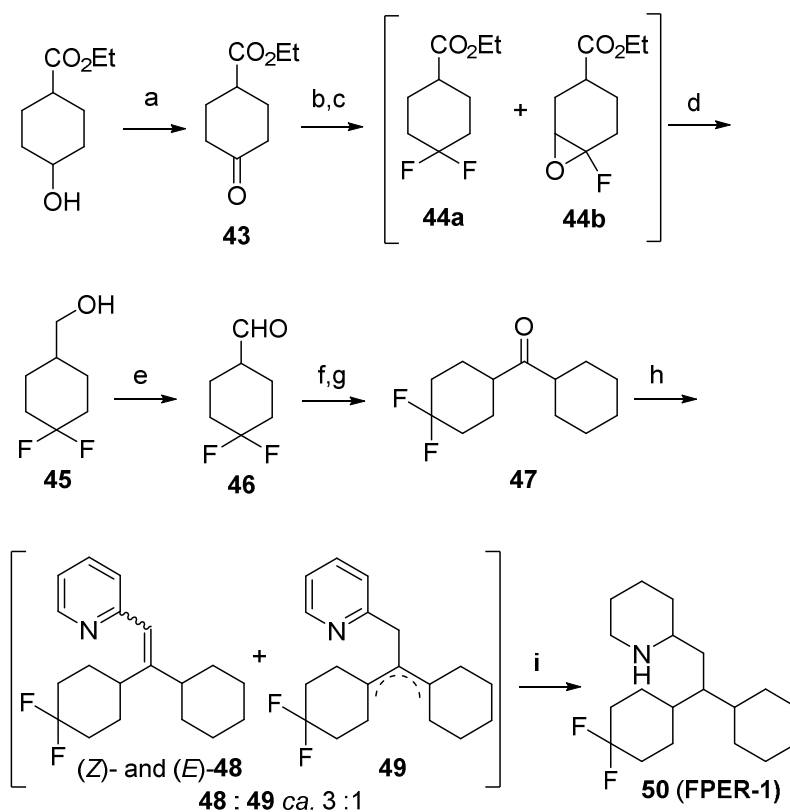
**Scheme 2.** Synthesis of 4-hydroxy-perhexiline **24**. Reagents and conditions: (a) MOMCl, DIPEA, CH<sub>2</sub>Cl<sub>2</sub> (93%); (b) Aq. NaOH, then NaBH<sub>4</sub>, THF, H<sub>2</sub>O, then Swern oxidation (87% over three steps); (c) Cyclohexyl magnesium bromide, Et<sub>2</sub>O (91%); (d) Py<sub>2</sub>Cr<sub>2</sub>O<sub>7</sub>, CH<sub>2</sub>Cl<sub>2</sub> (83%); (e) *n*BuLi, 2-picoline then SOCl<sub>2</sub>; (f) PtO<sub>2</sub>, H<sub>2</sub> (1 atm), MeOH; (g) TFA, CH<sub>2</sub>Cl<sub>2</sub> then NaOH (82% over steps e-g).

The synthesis of fluorinated perhexiline derivatives proved unexpectedly difficult and the standard late-stage fluorination approaches were not successful in introducing a fluorine onto position 4 of the cyclohexyl ring. In fact, nucleophilic attack on a 4-cyclohexyl mesylate or triflate derived from dicyclohexyl-carbinol **14** (see Supporting Information, Scheme S2) using TBAF or KF led to elimination rather than substitution, while use of DAST to effect the dehydroxyfluorination of **24** (Scheme 2) – also in *N*-protected form - led to dehydration. An alternative approach, hydrogenation of the potential 4-fluorophenyl precursor **6h** (Scheme 1) using a number of different catalysts led predominantly to concomitant reductive defluorination and gave perhexiline as the end product. Attempts to access vinyl-fluoride **27a** (See Scheme S3, Supporting Information) via Ag-mediated *ipso*-fluorodestannylation of the corresponding vinyl stannane with electrophilic fluorinating agents resulted in impractical outcomes.<sup>26</sup> Eventually, we were able to develop multi-step synthetic strategies capable of producing enough pure fluoro-perhexilines **31**, **42**, **50**, **55** and **62** (Fig. 3) for performing biological tests.



**Figure 3.** Fluoro perhexiline derivatives synthesised in this work.

As an example, the synthesis of 4,4-difluoro-perhexiline **50** (FPER-1) is shown in Scheme 3 (for detailed information on the synthesis of all fluoro-perhexilines see the Supporting Information). Commercially available ethyl 4-hydroxycyclohexane-carboxylate was first oxidized to ketone **43** and then submitted to deoxofluorination with DAST, affording an inseparable mixture of the desired *gem*-difluoro-ester **44a** and vinyl-fluoride by-product, which – for ease of separation - was oxidized with mCPBA to give the fluoro-oxirane **44b**. This mixture was then reduced with LiAlH<sub>4</sub>, then FC purification afforded the difluoro-alcohol **45**, which was transformed into the target *gem*-difluorinated molecule **50** in 5 steps.



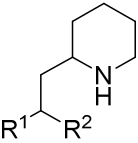
**Scheme 3.** Synthesis of 4,4-difluoro-perhexiline **50** (FPER-1). Reagents and conditions: (a) PCC (88%); (b) DAST; (c) mCPBA; (d) LiAlH<sub>4</sub> (62% over steps b-d); (e) Swern oxidation (64%); (f) Cyclohexyl magnesium chloride; (g) DMP (86% over steps f-g); (h) *n*BuLi, 2-picoline then SOCl<sub>2</sub> (88%); (i) Pd(OH)<sub>2</sub>/C, H<sub>2</sub> (1 atm), AcOH, 50 °C (94%).

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5 Racemic perhexiline **1** shows reasonable stability in the HLM assay, with a half-life of 100  
6 minutes, but with a short half-life in RLM of just 21 minutes (Table 1). The relevance of this  
7 was validated in the presence of CYP2D6-expressing Bactosomes<sup>TM</sup>, in which perhexiline  
8 was very rapidly turned over, with a half-life of just 14 minutes. As these rat liver microsomes  
9 are prepared from Sprague-Dawley rats, regarded as a model for CYP2D6 extensive  
10 metabolisers,<sup>12</sup> this is a logical correlation. The closest analogue of perhexiline **6c**, in which  
11 one cyclohexyl had been replaced by a cyclopentyl ring, showed near identical stability to  
12 the parent compound in both the RLM and CYP2D6 assays. As shown in Table 1, none of  
13 these unsubstituted cycloalkyl analogues **6a-j** showed significantly improved RLM stability  
14 compared to perhexiline **1**. The improved stability of "(+)-perhexiline" (absolute  
15 stereochemistry not defined at that stage) has previously been proposed.<sup>27</sup> The results  
16 shown in Table 1 may be taken to suggest that (*R*)-(+)-perhexiline **1a** was more stable than  
17 (*S*)-(-)-perhexiline **1b** in the HLM assay (98 and 65 min, respectively), whilst the two  
18 enantiomers had the same stability in the RLM assay (10 and 11 min, respectively).  
19 However, in view of the apparent lower stability of both of the two enantiomers in  
20 comparison with the racemic parent compound **1** in the RLM assay (21 min; a highly  
21 unlikely, if not impossible occurrence) we do not feel that this assay can be used to draw any  
22 conclusions concerning relatively small differences in potency. However, we were able to  
23 conclude that replacement of one or both cyclohexyl groups with other cycloalkyl groups did  
24 not remove the CYP2D6 liability. Other modifications, including the addition of a *gem*-  
25 dimethyl onto one of the cyclohexyl rings (**6j**), or replacement of one of the rings with a 4-  
26 tetrahydropyranyl (**6i**), with a phenyl (**6g**), or with a 4-fluorophenyl (**6h**), did not lead to  
27 significant improvements in the PK profile. The 4-tetrahydropyran derivative **6i** gave an  
28 apparent gain in RLM stability, but this was not reflected by any improvement in stability to  
29 CYP2D6. These results suggested that it would be necessary to either replace both  
30 cyclohexyl rings, or to more strongly deactivate a remaining cyclohexyl ring in order to  
31 remove the metabolic liability. Disappointingly, and unexpectedly, the addition of one  
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3 fluorine (compound **31**) gave no improvement in stability and, equally surprisingly, the  
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5 addition of one fluorine on each cyclohexyl (compound **42**) gave a compound which was  
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7 chemically unstable on storage in pure form. All other compounds showed no decomposition  
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9 on incubation for 1 h in pH 7.4 PBS at 37 °C. It is not clear as to what caused this instability,  
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11 but this compound was not further evaluated. At the other end of the scale, as predicted, the  
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13 tetrafluoro derivative **62**, in which all of the potential sites for metabolism, as identified by  
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15 previous studies<sup>28</sup> had been blocked, was extremely stable. Perhaps most interestingly, the  
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17 addition of a 4,4-difluoro group onto just one of the cyclohexyl rings (compound **50**) also  
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19 gave a very substantial increase in stability over the parent compound, and stability in the  
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21 HLM, RLM and CYP2D6 assays had increased to >500, 72 and 104 min respectively (Table  
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23 1). Our hypothesis, aimed at leaving as much of the parent structure intact as possible, was  
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25 that one, but perhaps not both, of the cyclohexyl rings formed an essential binding  
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27 interaction with the target. Thus a derivative might be more likely to retain activity if only one  
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29 ring had been modified. With this in mind, we also prepared the 3,3-difluorocyclohexyl  
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31 derivative **55**, feeling that the reduction in electron density from a more distant group could  
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33 be sufficient to reduce metabolism, whilst at the same time perhaps interfering less with a  
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35 binding interaction. However, this compound did not show the same improvement in stability.  
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37 These data show that certain fluorinated perhexiline analogues are much less heavily  
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39 metabolised by CYP2D6 than perhexiline and thus should avoid the inter-individual  
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41 variability which compromises perhexiline therapy. Therefore, these compounds are  
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43 expected to show more reliable and predictable pharmacokinetics and hence may also avoid  
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45 the toxicity issues found with perhexiline.  
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51 **Table 1.** Human and rat liver microsomal (HLM / RLM) stability studies.

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	R <sup>1</sup>	R <sup>2</sup>	HLM t <sub>1/2</sub> (min)	RLM t <sub>1/2</sub> (min)	CYP2D6 t <sub>1/2</sub> (min)

			± SEM	± SEM	± SEM
(±)-PHX <b>1</b>	Cyclohexyl	Cyclohexyl	100 ± 1	21 ± 1	14 ± 3
(+)-(R)-PHX <b>1a</b>	Cyclohexyl	Cyclohexyl	98 ± 25	10 ± 0.8	-
(-)-(S)-PHX <b>1b</b>	Cyclohexyl	Cyclohexyl	65 ± 9	11 ± 0.4	-
<b>6a</b>	Cyclohexyl	Cyclopropyl	124 ± 18	3 ± 0.5	-
<b>6b</b>	Cyclohexyl	Cyclobutyl	>200	10 ± 0.6	-
<b>6c</b>	Cyclohexyl	Cyclopentyl	>150	19 ± 2	16 ± 1
<b>6d</b>	Cyclopentyl	Cyclopropyl	27 ± 2	3 ± 0.04	-
<b>6e</b>	Cyclopentyl	Cyclobutyl	51 ± 5	7 ± 0.5	-
<b>6f</b>	Cyclopentyl	Cyclopentyl	50 ± 4	14 ± 1	-
<b>6g</b>	Cyclohexyl	Phenyl	56 ± 7	4 ± 0.3	-
<b>6h</b>	Cyclohexyl	4-Fluorophenyl	79 ± 8	7 ± 0.2	-
<b>6i</b>	Cyclohexyl	4-Tetrahydro pyranyl	84 ± 21	31 ± 1	7 ± 0.8
<b>6j</b>	Cyclohexyl	4,4-Dimethyl cyclohexyl	77 ± 11	20 ± 1	-
<b>31</b>	Cyclohexyl	4-Fluoro cyclohexyl	65 ± 7	11 ± 0.9	16 ± 0.5
<b>50</b>	Cyclohexyl	4,4-Difluoro cyclohexyl	>500	72 ± 6	104 ± 17
<b>55</b>	Cyclohexyl	3,3-Difluoro cyclohexyl	>150	35 ± 1	-
<b>42</b>	4-Fluoro cyclohexyl	4-Fluoro cyclohexyl	-	-	-
<b>62</b>	4,4-Difluoro cyclohexyl	4,4-Difluoro cyclohexyl	>4000	>1500	>150

**Biology:** We compared our compounds to perhexiline **1** in a suitable efficacy model in which perhexiline has shown potency. Compound selection was guided by first determining *in vitro* inhibitory potency against the most widely accepted target for perhexiline, CPT-1 (Table 2). Inhibitory potency against rat mitochondrial CPT-1 *in vitro* was measured by determining the ability of derivatives to inhibit the conversion of palmitoyl CoA and L-carnitine into palmitoylcarnitine, measuring the disappearance of palmitoyl CoA spectrophotometrically, as described previously.<sup>29</sup> In this assay the ~50% of the CPT activity that is sensitive to inhibition by malonyl CoA is defined as CPT-1,<sup>29</sup> and under the conditions of the assay we obtained an IC<sub>50</sub> of 1.2 μM for malonyl CoA. Since all tested inhibitors also showed the same maximal inhibition of activity as malonyl CoA, we assumed that this also represents inhibition of CPT-1.

We found that the *gem*-difluoro derivative **50** showed only a small loss in inhibitory potency (45 μM, *P*<0.05) compared to racemic perhexiline **1** (11 μM) and its (*S*) enantiomer **1b** (14 μM). The apparent increased potency of the (*S*) enantiomer **1b** matches the published

suggestion that “(-)-perhexiline” might have better therapeutic properties.<sup>14</sup> The (*R*) enantiomer **1a** shows a significant reduction in potency compared to racemic perhexiline **1** (32  $\mu\text{M}$ ,  $P < 0.05$ ) and approaches significance when compared to the (*S*) enantiomer **1b**. This compares to a literature value of 77  $\mu\text{M}$  for perhexiline **1**.<sup>17</sup> However, the difficulties in obtaining precise and consistent data for CPT-1 inhibition have been noted and discussed by Ceccarelli.<sup>18</sup> The exceptionally stable tetrafluoro derivative **62** had lost some of its activity against CPT-1 (88  $\mu\text{M}$ ,  $P < 0.05$  compared to **50**). Replacement of one cyclohexyl with a cyclopentyl in compound **6c** had no impact on potency (11  $\mu\text{M}$ ), whilst replacement of both cyclohexyls with cyclopentyl groups, in compound **6f**, gave a non-significant reduction in potency (19  $\mu\text{M}$ ). However, the hypothesis that retaining one intact cyclohexyl might be sufficient to retain activity, whilst a wide range of modifications on the other might be tolerated, was disproved by the loss of potency seen with the 3,3-difluoro **55** (215  $\mu\text{M}$ ) and tetrahydropyran **6i** (715  $\mu\text{M}$ ) derivatives. We also investigated the stereoisomeric mixture of 4-hydroxy derivatives **24** to study the possibility that the activity of perhexiline was in-part mediated through a metabolite,<sup>28</sup> but instead found it to be very poorly active (475  $\mu\text{M}$ ). Having the best balance of pharmacokinetic and inhibitory properties, plus relative ease of synthesis, we selected the 4,4-*gem*-difluoro **50** derivative to take forward for *in vivo* PK and efficacy study.

**Table 2.** CPT-1 inhibitory activities.

	R <sup>1</sup>	R <sup>2</sup>	IC <sub>50</sub> for CPT-1 inhibition ( $\mu\text{M}$ )*
(±)-PHX <b>1</b>	Cyclohexyl	Cyclohexyl	11 (7 - 17)
(+)-( <i>R</i> )-PHX <b>1a</b>	Cyclohexyl	Cyclohexyl	32 (24 - 41)
(-)-( <i>S</i> )-PHX <b>1b</b>	Cyclohexyl	Cyclohexyl	14 (8 - 25)
<b>6c</b>	Cyclohexyl	Cyclopentyl	11 (5 - 24)
<b>6f</b>	Cyclopentyl	Cyclopentyl	19 (7 - 40)
<b>6i</b>	Cyclohexyl	4-Tetrahydropyranyl	715 (510 - 1050)
<b>50</b>	Cyclohexyl	4,4-Difluorocyclohexyl	45 (34 - 60)
<b>55</b>	Cyclohexyl	3,3-Difluorocyclohexyl	215 (160 - 290)
<b>62</b>	4,4-Difluorocyclohexyl	4,4-Difluorocyclohexyl	88 (61 - 127)
<b>24</b>	Cyclohexyl	4-Hydroxycyclohexyl	475 (290 - 745)

\* mean and pooled 95% confidence interval of two independent experiments.

**In vivo pharmacokinetics:** It has been proposed that the unexplained activity of perhexiline **1**, in spite of modest *in vitro* potency against all proposed targets, is due to selective accumulation in the myocardium. Thus, in a continued attempt to ensure that our compounds mirrored perhexiline in as many aspects as possible, we conducted studies to determine the *in vivo* stability and myocardial exposure. The distribution and stability of **50**, following daily oral dosing, were compared to perhexiline in male BALB/c mice (samples taken at 1, 4, 8 and 24 h following dosing; n = 3 per timepoint – see supplementary information for full details). Plasma and myocardial concentrations, and area under the curve, following 10 mg/kg oral gavage of **50** and perhexiline were investigated. In the mouse model, perhexiline **1** is very rapidly metabolised and thus no direct comparisons can be drawn between the absolute concentrations of the two compounds. However, Table 3 shows that there is a significant accumulation of perhexiline in the myocardium (5.7-fold at 1 h, 4.9-fold at 4 h and 4.8-fold at 8 h, as defined by drug concentration; 2.4-fold as defined by AUC), whilst **50** shows an even larger selective accumulation (10.7-fold at 1 h, 10.9-fold at 4 h and 9.5-fold at 8 h, as defined by drug concentration; 8.6-fold as defined by AUC).

**Table 3. In vivo pharmacokinetics in a mouse model**

		<b>50</b> (10 mg/kg)		Perhexiline (10 mg/kg)	
		Plasma	Myocardium	Plasma	Myocardium
Conc (ng/mL)	1 h	346.5 ± 257.4	3726.8 ± 2808.5	116.4 ± 39.2	664.5 ± 157.5 <sup>#</sup>
	4 h	171.9 ± 44.1 <sup>*</sup>	1877.3 ± 606.1 <sup>*#</sup>	28.3 ± 8.8	125.8 ± 62.5
	8 h	191.8 ± 48.0 <sup>*</sup>	1826.7 ± 498.7 <sup>*#</sup>	6.0 ± 5.6	29.2 ± 25.2
AUC 0-24 (hr*ng/mL)		3763 ± 1501 <sup>*</sup>	32360 ± 7030 <sup>**##</sup>	876 ± 123	2114 ± 265 <sup>##</sup>

<sup>\*</sup>*P*<0.05 relative to perhexiline; <sup>\*\*</sup>*P*<0.01 relative to perhexiline; <sup>#</sup>*P*<0.05 relative to plasma; <sup>##</sup>*P*<0.01 relative to plasma (Student's t-test).

Further studies were conducted, showing that selective myocardial accumulation was also seen in male Sprague-Dawley rats (used as a model for CYP2D6-mediated metabolism)<sup>12</sup> (Table 4). The distribution and stability of **50** following daily oral dosing were compared to



perhexiline **1** (plasma samples taken at 1, 2, 4, 8 and 24 h following dosing; n = 3 per timepoint – see Supporting Information for full details). Again perhexiline is metabolised very quickly, which is not representative of the situation in humans, but similar myocardium : plasma ratios can be seen in comparing perhexiline **1** (19-fold at 1 h and 11.5-fold at 8 h, as defined by drug concentration) and **50** (15.3-fold at 1 h and 7.2-fold at 8 h, as defined by drug concentration).

**Table 4. *In vivo* pharmacokinetics in a rat model**

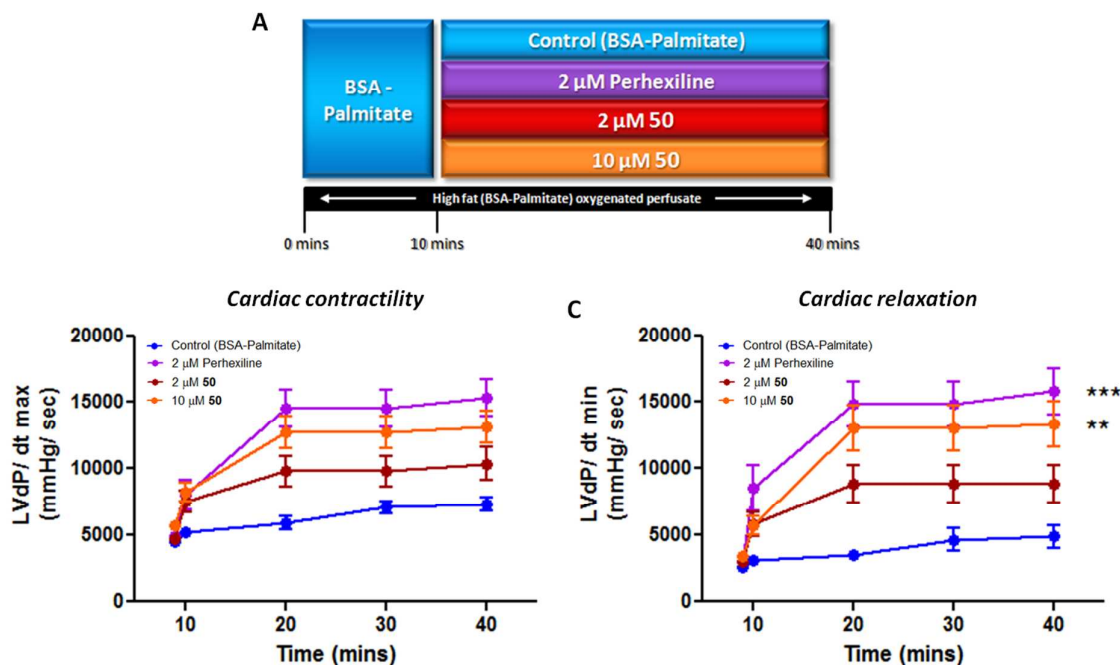
		<b>50</b> (10 mg/kg)		Perhexiline (10 mg/kg)	
		Plasma	Myocardium	Plasma	Myocardium
Conc (ng/mL)	1 h	386.3 ± 28.4*	5913.4 ± 1350.2 <sup>#</sup>	99.9 ± 43.5	1901.2 ± 887.1 <sup>#</sup>
	2 h	375.4 ± 139.7*	-	112.9 ± 32.6	-
	4 h	297.1 ± 145.2	-	97.44 ± 24.3	-
	8 h	178.6 ± 93.4	1285.9 ± 822.8	31.4 ± 5.7	361.2 ± 104.7 <sup>#</sup>

\**P*<0.05 relative to perhexiline; <sup>#</sup>*P*<0.05 relative to plasma (Student's t-test)

Thus, assuming that this accumulation is responsible for the therapeutic effect, rather than presenting a toxicological liability, **50** appears to match our requirements as an optimised (in terms of CYP2D6 liability) orally-bioavailable analogue of perhexiline, suitable for efficacy studies.

***Ex vivo* efficacy studies:** The *ex vivo* Langendorff murine model was used to assess the effects of perhexiline **1** and analogue **50** on indicators of cardiac contractility (LVdP/dt max) and relaxation (LVdP/dt min) following perfusion with a high fat buffer. In recent years it has been shown that HF can be characterised by severe metabolic disturbances, intrinsic to the myocardium, which include fatty acid overload and build-up within the cardiomyocyte cytosol and mitochondria.<sup>30</sup> As such, energy production within the cardiomyocyte is impaired, which in turn disrupts cardiac function and contractile activity over time. By perfusing hearts with a high fat buffer, we can simulate the fatty acid overload observed during HF and can measure

the effects of perhexiline **1** and compound **50** on cardiac function, such as LVdP/dt max and min, under these conditions.<sup>31</sup>



**Figure 2.** Perhexiline **1** and 4,4-*gem*-difluoro-perhexiline **50** improve cardiac contractility and relaxation in a high fat perfusate. **A:** The Langendorff perfusion protocol. Hearts were stabilised for 10 minutes with BSA-Palmitate and then perfused for 30 minutes with either 1) Control (BSA-Palmitate), 2) 2 μM perhexiline, 3) 2 μM **50** or 4) 10 μM **50**. Cardiac contractility (LVdP/dt max) and cardiac relaxation (LVdP/dt min) were calculated from the left ventricular pressure. **B:** LVdP/dt max. **C:** LVdP/dt min. Data are presented as mean ± SEM; n = 12-19 mice. \*\* $P < 0.01$  vs Control (BSA-Palmitate); \*\*\* $P < 0.001$  vs Control (BSA-palmitate) (1-way ANOVA, post-hoc Bonferroni test).

The results demonstrate that 2 μM perhexiline **1** shows a strong trend for enhancing LVdP/dt max (2.1 fold increase when compared to the control; n=16;  $P=0.07$ ) and is highly effective for enhancing LVdP/dt min (3.2 fold increase when compared to the control; n=16;  $P < 0.001$ ) in the presence of high fat. At a concentration of 10 μM, like perhexiline, **50** showed a similar trend to increasing LVdP/dt max (1.8 fold increase when compared to the control; n=16

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3  $P=0.07$ ) and was almost as effective as perhexiline in increasing LVdP/dt min (2.7 fold  
4 increase when compared to the control;  $n=16$ ;  $P<0.01$ ). These results also demonstrate that  
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7 2  $\mu\text{M}$  perhexiline and 10  $\mu\text{M}$  **50** have significant lusitropic effects, a feature that current  
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9 pharmacological treatments for HF lack. It can be noted that the apparent 5-fold loss of  
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11 potency of **50** compared to perhexiline is in accord with the relative potencies vs CPT-1;  
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13 however - as it cannot be assumed that CPT-1 is the only target, or even the major target, or  
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15 that the ratio of potencies for perhexiline and **50** vs other targets will remain the same – the  
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17 importance of this observation cannot be determined. At present, most drug therapies for  
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19 HF only improve systolic function such as cardiac contractility, and therefore only benefit  
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21 patients with systolic HF.<sup>32</sup> As such, perhexiline and **50** show promising potential for use in  
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23 diastolic HF, a condition in which cardiac relaxation is impaired, as opposed to contractility,  
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25 and where there are currently no well-defined existing treatments.<sup>32-34</sup> Furthermore, **50** at 2  
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27  $\mu\text{M}$  shows a small improvement on both cardiac parameters with clear evidence of dose  
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29 dependency.  
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## 32 33 CONCLUSION

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35 In summary, these results reveal that 4,4-*gem*-difluoro-perhexiline **50** (FPER-1) holds a  
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37 similar therapeutic profile to perhexiline **1** and thus **50**, and possibly other analogues of  
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39 perhexiline, may have the same potential for the treatment of HF and cardiovascular  
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41 diseases, but without the metabolic liability of the parent compound. As a consequence,  
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43 compounds with the properties exemplified by **50** are expected to be suitable for use across  
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45 a broader spectrum of the patient population and not to require additional monitoring of drug  
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47 plasma concentrations or patient phenotyping, both of which limit drug use and convenience.  
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49 However, it is also clear from these studies that elucidation of the mechanism of action of  
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51 perhexiline will be required before it will be possible to conduct an informed drug discovery  
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53 programme to generate a perhexiline-like drug for clinical development.  
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## 56 57 58 EXPERIMENTAL SECTION

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3 **General Methods:** All reactions were carried out under a nitrogen or argon atmosphere with  
4 dry solvents under anhydrous conditions, unless otherwise mentioned. Solvents and  
5 reagents: anhydrous tetrahydrofuran (THF), diethyl ether (Et<sub>2</sub>O), methylene chloride  
6 (CH<sub>2</sub>Cl<sub>2</sub>) and toluene were purchased from commercial suppliers. Reagents were purchased  
7 at the highest commercial quality, used without further purification and handled in  
8 accordance with COSHH regulations. Chromatography: flash chromatography (FC) was  
9 carried out on silica gel (Merck Silica gel Si 60, 40-63 μm) according to the method  
10 described by Still.<sup>35</sup> Thin-layer chromatography (TLC) was carried out on glass-based 0.25  
11 mm Merck silica gel plates (60F-254) which were developed with UV irradiation (254 nm and  
12 365 nm), an aqueous solution of KMnO<sub>4</sub> and an ethanol solution of ammonium molybdate,  
13 and heat as developing agents. <sup>1</sup>H NMR spectra: these were recorded at 400 MHz on a  
14 Bruker ADVANCEIII 400 instrument. Chemical shifts (δ<sub>H</sub>) are given in parts per million (ppm)  
15 as referenced to the appropriate residual solvent peak. <sup>13</sup>C NMR spectra: these were  
16 recorded at 101 MHz on a Bruker ADVANCE III 400 instrument. Chemical shifts (δ<sub>C</sub>) are  
17 given in parts per million (ppm) as referenced to CHCl<sub>3</sub>. <sup>19</sup>F NMR spectra: these were  
18 recorded at 376 MHz on a Bruker ADVANCE III 400 instrument without <sup>1</sup>H decoupling.  
19 Chemical shifts (δ<sub>C</sub>) are given in parts per million (ppm) as referenced to CFC<sub>3</sub> as 0 ppm. All  
20 NMR spectra were recorded at 298K unless otherwise stated. The following abbreviations  
21 were used to explain the multiplicities: s = singlet, d = doublet, t = triplet, q = quartet, quint =  
22 quintet, m = multiplet, br = broad. Purities of all tested compounds were determined by  
23 reverse phase HPLC-MS using an Agilent 1200 series chromatograph system coupled with  
24 an Agilent G6120 signal quadrupole detector equipped with an electrospray ionisation  
25 source and detection in positive mode. HPLC conditions for purity analysis: Zorbax Eclipse  
26 Plus C18 column, 4.6 × 100 mm, 3.5 μm; mobile phase: 75% MeOH / 25% H<sub>2</sub>O with 0.1%  
27 formic acid; flow rate: 1 mL/min. All tested compounds were found to be at least 96% purity.  
28 HRMS: analyses were performed using an LTQ Orbitrap XL MS spectrometer.  
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3 *Representative procedures for carbocyclic perhexiline analogues.*

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5 **6-((*tert*-Butyldimethylsilyl)oxy)hexan-2-one**

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8 A solution of MeLi (14.8 mL, 23.71 mmol, 1.6 M in Et<sub>2</sub>O) was added at -78 °C to an Et<sub>2</sub>O (10  
9 mL) solution of δ-valerolactone (2.158 g, 21.55 mmol). The resulting reaction mixture was  
10 stirred for 1 h at -78 °C before quenching with NH<sub>4</sub>Cl (10 mL, sat. aq.) and dilution with  
11 EtOAc (20 mL). The layers were separated and the organic layer was washed with brine (20  
12 mL), dried (Na<sub>2</sub>SO<sub>4</sub>) and concentrated under reduced pressure to give the crude hydroxy  
13 ketone, which was used directly without further purification.

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21 TBSCl (3.92 g, 26.0 mmol) and imidazole (2.31 g, 33.93 mmol) were added at 0 °C to a  
22 CH<sub>2</sub>Cl<sub>2</sub> (10 mL) solution of the crude hydroxy ketone. The resulting mixture was allowed to  
23 warm to 23 °C and stirred for 3 h prior to quenching with a saturated solution of NaHCO<sub>3</sub> (10  
24 mL) and dilution with CH<sub>2</sub>Cl<sub>2</sub> (10 mL). The layers were separated, and the aqueous layer  
25 was extracted with CH<sub>2</sub>Cl<sub>2</sub> (2 × 5 mL). The combined organic layers were dried over Na<sub>2</sub>SO<sub>4</sub>  
26 and concentrated under reduced pressure. FC (silica gel, hexanes:EtOAc 5:1) afforded 6-  
27 ((*tert*-Butyldimethylsilyl)oxy)hexan-2-one (4.64 g, 93% for two steps) as a colourless oil. <sup>1</sup>H  
28 NMR (400 MHz, CDCl<sub>3</sub>) δ = 3.60 (t, *J* = 6.3 Hz, 2H), 2.44 (t, *J* = 7.3 Hz, 2H), 2.12 (s, 3H),  
29 1.69 – 1.58 (m, 2H), 1.56 – 1.45 (m, 2H), 0.88 (s, 9H), 0.03 (s, 6H); <sup>13</sup>C NMR (101 MHz,  
30 CDCl<sub>3</sub>) δ = 209.1, 62.8, 43.5, 32.2, 29.8, 25.9, 20.3, 18.3, -5.3. *m/z* (ESI) 231 [M+H]<sup>+</sup>.

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44 **7-((*tert*-Butyldimethylsilyl)oxy)-1-cyclohexyl-1-hydroxyheptan-3-one (2a)**

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46 *Method A:* a solution of freshly prepared LDA (3.80 mL, 3.80 mmol, 1.0 M in THF/hexanes)  
47 was added at -78 °C to a THF (5 mL) solution of 6-((*tert*-butyldimethylsilyl)oxy)hexan-2-one  
48 (730 mg, 3.17 mmol). The resulting mixture was stirred for 1 h at -78 °C before  
49 cyclohexanecarboxaldehyde (426 mg, 460 μL, 3.80 mmol) was added. The reaction mixture  
50 was stirred at -78 °C for further 30 min before quenching with NH<sub>4</sub>Cl (10 mL, sat. aq.) and  
51 dilution with EtOAc (10 mL). The layers were separated and the organic layer was washed  
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3 with brine (20 mL), dried (Na<sub>2</sub>SO<sub>4</sub>) and concentrated under reduced pressure. FC (silica gel,  
4 hexanes:EtOAc 2:1) furnished the aldol adduct **2a** (955 mg, 88 %) as a pale yellow oil. <sup>1</sup>H  
5 NMR (400 MHz, CDCl<sub>3</sub>) δ = 3.85 – 3.73 (m, 1H), 3.60 (t, *J* = 6.2 Hz, 2H), 3.04 (brs, 1H), 2.67  
6 – 2.39 (m, 4H), 1.89 – 1.78 (m, 1H), 1.78 – 1.70 (m, 2H), 1.69 – 1.57 (m, 4H), 1.54 – 1.47  
7 (m, 2H), 1.41 – 1.29 (m, 1H), 1.29 – 1.09 (m, 3H), 1.09 – 0.92 (m, 2H), 0.88 (s, 9H), 0.04 (s,  
8 6H); <sup>13</sup>C NMR (101 MHz, CDCl<sub>3</sub>) δ = 212.6, 71.7, 62.7, 46.1, 43.4, 43.0, 32.1, 28.8, 28.3,  
9 26.4, 26.2, 26.1, 25.9, 20.1, 18.3, -5.3. *m/z* (ESI) 343 [M+H]<sup>+</sup>.  
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### 20 **(*E*)-7-((*tert*-Butyldimethylsilyl)oxy)-1-cyclohexylhept-1-en-3-one (3a)**

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22 *Method B:* Et<sub>3</sub>N (540 μL, 3.87 mmol) and MsCl (150 μL, 1.94 mmol) were added at 0 °C to a  
23 CH<sub>2</sub>Cl<sub>2</sub> solution of aldol adduct **2a** (440 mg, 1.28 mmol). The solution was stirred for 1 h at 0  
24 °C followed by addition of DBU (290 μL, 1.94 mmol). The resulting mixture was warmed to  
25 23 °C and stirred for further 2 h prior to quenching with NH<sub>4</sub>Cl (10 mL, sat. aq.) and dilution  
26 with EtOAc (20 mL). The layers were separated and the organic layer was washed with brine  
27 (20 mL), dried (Na<sub>2</sub>SO<sub>4</sub>) and concentrated under reduced pressure. FC (silica gel,  
28 hexanes:EtOAc 5:1) afforded the enone **3a** (410 mg, 98%) as a pale yellow oil. <sup>1</sup>H NMR (400  
29 MHz, CDCl<sub>3</sub>) δ = 6.76 (dd, *J* = 16.0, 6.8 Hz, 1H), 6.05 (dd, *J* = 16.0, 1.4 Hz, 1H), 3.63 (t, *J* =  
30 6.2 Hz, 2H), 2.57 (t, *J* = 7.3 Hz, 2H), 2.19 – 2.08 (m, 1H), 1.81 – 1.73 (m, 4H), 1.73 – 1.60  
31 (m, 3H), 1.60 – 1.49 (m, 3H), 1.36 – 1.26 (m, 2H), 1.22 – 1.09 (m, 2H), 0.89 (m, 9H), 0.04  
32 (m, 6H); <sup>13</sup>C NMR (101 MHz, CDCl<sub>3</sub>) δ = 201.1, 152.1, 127.8, 62.9, 40.6, 39.8, 32.3, 31.8,  
33 25.95, 25.92, 25.7, 20.8, 18.3, -5.3. *m/z* (ESI) 325 [M+H]<sup>+</sup>.  
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### 50 **7-Cyclohexyl-7-cyclopropyl-5-oxoheptanal (5a)**

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52 *Method C:*<sup>23</sup> a solution of cyclopropylmagnesium bromide (1.67 mL, 1.67 mmol, 1.0 M in 2-  
53 MeTHF) was added at -78 °C to a suspension of CuI (159 mg, 0.835 mmol) in THF (2 mL).  
54 The resulting mixture was allowed to warm to 0 °C and stirred for further 1 h to give the  
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organocuprate reagent. The solution was then cooled to -78 °C followed by addition of a THF solution (0.5 mL) of enone **3a** (180 mg, 0.56 mmol) and TMSCl (142  $\mu$ L, 1.12 mmol). The reaction mixture was stirred at -78 °C for 1 h before quenching with a MeOH solution of PTSA (2 N, 5 mL, 10 mmol). The solution was stirred at 23 °C for 2 h prior to dilution with EtOAc (20 mL). The layers were separated and the organic layer was washed with brine (10 mL), dried (Na<sub>2</sub>SO<sub>4</sub>) and concentrated under reduced pressure. The mixture was passed through a plug of silica gel (EtOAc) to give the corresponding alcohol, which was used without further purification.

Dess–Martin periodinane (285 mg, 0.67 mmol) was added at 23 °C to a CH<sub>2</sub>Cl<sub>2</sub> solution (5 mL) of the alcohol above. The reaction mixture was stirred for 6 h before quenching with Na<sub>2</sub>SO<sub>3</sub> (5 mL, sat. aq.). The resulting mixture was extracted with CH<sub>2</sub>Cl<sub>2</sub> (3  $\times$  5 mL), combined organic layers were dried over Na<sub>2</sub>SO<sub>4</sub> and concentrated under reduced pressure. The residue was purified by FC (silica gel, hexanes:EtOAc 3:1) afforded the dicarbonyl compound **5a** (101 mg, 73 % for three steps). <sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>)  $\delta$  = 9.82 (t, *J* = 1.5 Hz, 1H), 2.59 – 2.39 (m, 5H), 1.78 – 1.42 (m, 8H), 1.42 – 1.26 (m, 1H), 1.26 – 0.91 (m, 6H), 0.68 – 0.49 (m, 1H), 0.49 – 0.37 (m, 1H), 0.37 – 0.22 (m, 1H), 0.12 (tt, *J* = 14.5, 7.2 Hz, 1H), -0.09 (td, *J* = 9.3, 5.3 Hz, 1H); <sup>13</sup>C NMR (101 MHz, CDCl<sub>3</sub>)  $\delta$  = 212.1, 201.3, 46.0, 45.5, 43.4, 42.8, 32.1, 30.6, 30.1, 26.8, 26.74, 26.68, 19.8, 14.21 5.6, 3.4. *m/z* (ESI) 251 [M+H]<sup>+</sup>.

### 2-(2-Cyclohexyl-2-cyclopropylethyl)piperidine (**6a**)

*Method D:*<sup>24</sup> NH<sub>4</sub>OAc (53 mg, 0.69 mmol), NaBCNH<sub>3</sub> (41 mg, 0.65 mmol) and 3 Å molecular sieves (100 mg) were added at 23 °C to a MeOH solution (2 mL) of dicarbonyl compound **5a** (78 mg, 0.31 mmol). The resulting mixture was stirred at 23 °C for 16 h before filtration through a pad of Celite<sup>®</sup>. The filtrate was concentrated under reduced pressure. The residue was purified by FC (silica gel, MeOH:CH<sub>2</sub>Cl<sub>2</sub> 0:100  $\rightarrow$  10:90) to afford compound **6a** (58 mg, 78 %, mixture of diastereoisomers, purity > 97%) as a pale yellow oil. NMR spectra were recorded by using the HCl salt of the compound, which was prepared by treating **6a** (10 mg)

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3 with HCl (200  $\mu$ L, 1.25 M in MeOH) followed by concentration under reduced pressure.  $^1\text{H}$   
4 NMR (400 MHz,  $\text{CDCl}_3$ )  $\delta$  = 9.24 (brs, 2H), 3.51 – 3.38 (m, 1H), 3.27 (dd,  $J$  = 13.1, 7.0 Hz,  
5 0.5H), 3.21 – 3.07 (m, 0.5H), 2.83 (dtd,  $J$  = 15.9, 12.7, 3.2 Hz, 1H), 2.22 – 1.53 (m, 12H),  
6 1.53 – 0.96 (m, 7H), 0.85 (ddd,  $J$  = 16.2, 13.6, 6.8 Hz, 0.5H), 0.72 – 0.37 (m, 7.5H), 0.24 –  
7 0.07 (m, 3H), 0.29 – 0.08 (m, 1.5H), 0.06 – -0.03 (m, 0.5 H);  $^{13}\text{C}$  NMR (101 MHz,  $\text{CDCl}_3$ )  $\delta$  =  
8 55.9, 55.6, 44.8, 44.7, 44.5, 44.29, 42.9, 42.0, 36.40, 36.35, 30.9, 30.7, 30.0, 29.0, 28.4,  
9 27.1, 26.94, 26.87, 26.75, 26.72, 22.5, 22.4, 22.3, 22.2, 13.61, 13.56, 4.9, 4.5, 4.0, 3.7.  $m/z$   
10 (ESI) 236  $[\text{M}+\text{H}]^+$ ; HRMS (ESI) calcd for  $\text{C}_{16}\text{H}_{30}\text{N}^+$   $[\text{M}+\text{H}]^+$  236.2373, found 236.2369.  
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### 22 **Ethyl 4-(methoxymethoxy)cyclohexanecarboxylate (20)**

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24 DIPEA (992  $\mu$ L, 5.8 mmol) and MOMCl (330  $\mu$ L, 4.35 mmol) were added to a  $\text{CH}_2\text{Cl}_2$  (7 mL)  
25 solution of ethyl 4-hydroxycyclohexanecarboxylate (500 mg, 2.9 mmol) cooled to 0  $^\circ\text{C}$ . After  
26 stirring at r.t. for 6 h  $\text{H}_2\text{O}$  (5 mL) was added, the layers were separated and the organic  
27 phase was dried ( $\text{Na}_2\text{SO}_4$ ). The solvent was evaporated and the crude was purified by FC  
28 (hexane:EtOAc 8:2) to give **20** (590 mg, 93% yield) as a colorless oil.  $^1\text{H}$  NMR (400 MHz,  
29  $\text{CDCl}_3$ )  $\delta$  = 4.47 (s, 1 H), 3.94 (q, 2 H,  $J$  = 7.0 Hz, 2 H), 3.58-3.51 (m, 1H), 3.17 (s, 3 H), 2.23  
30 – 2.21 (m, 1 H), 1.93-1.78 (m, 2H), 1.76-1.65 (m, 2H), 1.63-1.52 (m, 2H), 1.51-1.41 (m, 3H),  
31 1.07 (t,  $J$  = 7.0 Hz, 3 H);  $^{13}\text{C}$  NMR (101 MHz,  $\text{CDCl}_3$ )  $\delta$  = 175.0, 94.5, 74.7, 60.0, 54.9, 42.2,  
32 29.6, 23.8, 14.0.  $m/z$  (ESI) 217  $[\text{M}+\text{H}]^+$   
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### 46 **4-(Methoxymethoxy)cyclohexanecarbaldehyde (21)**

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48 NaOH (210 mg, 5.24 mmol) was added to a solution of **20** (570 mg, 2.62 mmol) in a  
49 THF: $\text{H}_2\text{O}$  4:1 mixture (10 mL). The reaction was stirred overnight, then  $\text{H}_2\text{O}$  (5 mL) was  
50 added. The layers were separated, the aqueous phase was acidified with 1N HCl until pH 1  
51 was reached and then extracted with EtOAc (10 mL). The organic phase was dried  
52 ( $\text{Na}_2\text{SO}_4$ ), filtered and the solvent was removed under reduced pressure to give the  
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3 corresponding acid (413 mg, 83% yield) as a white solid. Et<sub>3</sub>N (250 μL, 1.80 mmol) and ethyl  
4 chloroformate (172 μL, 1.80 mmol) were added to a dry THF (5 mL) solution of acid (300  
5 mg, 1.6 mmol) cooled to 0 °C. The resulting suspension was stirred at the same temperature  
6 for 1h and then filtered through Celite. The filtrate was then added to a solution of NaBH<sub>4</sub>  
7 (116 mg, 3.1 mmol) in water (2 mL) cooled to 0 °C. The reaction mixture was stirred at the  
8 same temperature for 30 min, quenched with H<sub>2</sub>O (5 ml) and extracted with EtOAc (10 mL).  
9 The solvent was removed under reduced pressure and the crude was purified by FC (Hex:  
10 EtOAc 1:1) to give 298 mg of the corresponding alcohol (268 mg, 96% yield) as a colourless  
11 oil. A solution of (COCl)<sub>2</sub> (133 μL, 1.5 mmol) in CH<sub>2</sub>Cl<sub>2</sub> (3 mL) was added dropwise to a  
12 solution of DMSO (204 μL, 2.87 mmol) in CH<sub>2</sub>Cl<sub>2</sub> (5 mL), cooled to -78 °C. The reaction was  
13 stirred at the same temperature for 15 min, then a solution of the alcohol above (200 mg,  
14 1.15 mmol) in CH<sub>2</sub>Cl<sub>2</sub> (5 mL) was added. After stirring for 1 h, Et<sub>3</sub>N (800 μL, 5.75 mmol) was  
15 added. After 30 min the reaction was warmed to r.t, quenched with 1N HCl (aq) and  
16 extracted with EtOAc (10 mL). The solvent was dried (Na<sub>2</sub>SO<sub>4</sub>), filtered and concentrated  
17 under reduced pressure. The crude was purified by FC (hexane:EtOAc 8:2) to give **58** (165  
18 mg, 83%) as a yellow oil. <sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>) δ = 9.51 (s, 1 H), 4.52 (s, 2 H), 3.66-  
19 3.55 (m, 1H), 3.23 (s, 3 H), 2.20 – 2.23 (m, 1 H), 1.98-1.85 (m, 2H), 1.82-1.70 (m, 1H), 1.67-  
20 1.40 (m, 4H), 1.30 – 1.21 (m, 1H); <sup>13</sup>C NMR (101 MHz, CDCl<sub>3</sub>) δ = 204.2, 94.6, 74.6, 71.9,  
21 55.1, 48.2, 31.1, 23.7, 21.2. *m/z* (ESI) 173 [M+H<sup>+</sup>].  
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#### 45 **Cyclohexyl(4-(methoxymethoxy)cyclohexyl)methanol (22)**

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47 A small crystal of iodine was added to a suspension of magnesium shavings (63 mg, 2.58  
48 mmol) in Et<sub>2</sub>O (2 mL, nitrogen atmosphere). Bromocyclohexane (317 μL, 2.58 mmol) was  
49 then added slowly (over 5 min), so as to maintain a gentle reflux. The mixture was heated to  
50 reflux for 30 min. After cooling to r.t, aldehyde **21** (370 mg, 2.15 mmol) in Et<sub>2</sub>O (2 mL, ) was  
51 slowly added. After stirring at r.t overnight, NH<sub>4</sub>Cl (5 mL, sat. aq.) was added and the  
52 reaction was extracted with EtOAc (10 mL). The organic layer was dried (Na<sub>2</sub>SO<sub>4</sub>), filtered  
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3 and concentrated under reduced pressure. The crude was purified by FC (hexane:EtOAc  
4 7:3) to give **22** (233 mg, 91%) as a yellow oil. <sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>) δ = 4.59 (s, 2 H),  
5 3.77 - 3.69 (m, 1H), 3.28 (s, 3H), 3.10-2.93 (m, 1H), 2.07- 1.92 (m, 1H), 1.88-1.77 (m, 1H),  
6 1.74-1.73 (m, 3H), 1.62-1.26 (m, 8H), 1.25- 0.87 (m, 8H); <sup>13</sup>C NMR (101 MHz, CDCl<sub>3</sub>) δ =  
7 94.5, 79.7, 75.9, 71.3, 55.1, 40.2, 39.8, 39.0, 32.4, 30.2, 29.9, 28.0, 26.5, 24.0, 21.8. *m/z*  
8 (ESI) 257 [M+H<sup>+</sup>]  
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### 18 **Cyclohexyl(4-(methoxymethoxy)cyclohexyl)methanone (23)**

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20 PCC (846 mg, 2.25 mmol) was added to a solution of **22** (470 mg, 1.87 mmol) in CH<sub>2</sub>Cl<sub>2</sub> (40  
21 mL). The reaction was stirred overnight, then filtered through Celite. The filtrate was  
22 concentrated under reduced pressure and purified by FC (hexane:EtOAc 9:1) to give **23**  
23 (395 mg, 83%) as a colourless oil. <sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>) δ = 4.56 (s, 2 H), 3.75 - 3.64  
24 (m, 1H), 3.27 (s, 3H), 2.50-2.35 (m, 2H), 1.84-1.74 (m, 2H), 1.73-1.62 (m, 6H), 1.61-1.35 (m,  
25 5H), 1.33- 1.05 (m, 5H); <sup>13</sup>C NMR (101 MHz, CDCl<sub>3</sub>) δ = 216.4, 94.6, 71.0, 55.2, 48.7, 48.0,  
26 29.8, 28.7, 25.8, 25.7, 23.1. *m/z* (ESI) 277.2 [M+Na<sup>+</sup>]  
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### 38 **4-(1-Cyclohexyl-2-(piperidin-2-yl)ethyl)cyclohexanol (24)**

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40 A solution of *n*BuLi (680 μL, 1.7 mmol 2.5 M in hexanes) was added dropwise to a stirred  
41 solution of 2-picoline (168 μL, 1.7 mmol) in THF (4 mL) at -78 °C. The resulting red solution  
42 was stirred at that temperature for 10 min before being warmed to -10 °C for 20 min, after  
43 which the mixture was cooled to -78 °C and a THF (2 mL) solution of **23** (395 mg, 1.55  
44 mmol) was added. The resulting mixture was stirred at that temperature for 30 min before  
45 SOCl<sub>2</sub> was added (186 μL, 2.55 mmol). The milky reaction mixture was warmed to 0 °C for  
46 30 min prior to quenching with NH<sub>4</sub>Cl (10 mL, sat. aq.) and dilution with EtOAc (10 mL). The  
47 layers were separated and the organic layer was washed with brine (20 mL), dried (Na<sub>2</sub>SO<sub>4</sub>)  
48 and concentrated under reduced pressure. FC (hexane:EtOAc 8:2) gave the intermediate  
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3 unsaturated 2-pyridines (442 mg, 75%) as a yellow oil. To a solution of these 2-pyridines  
4 (210 mg, 0.63 mmol) in MeOH (7 mL) a catalytic amount of PtO<sub>2</sub> was added. The reaction  
5 was stirred under a hydrogen atmosphere for 24 h, filtered through a Celite<sup>®</sup> pad and  
6 concentrated under reduced pressure. The crude was purified by FC (CH<sub>2</sub>Cl<sub>2</sub>:MeOH 9:1) to  
7 give 2-(2-cyclohexyl-2-(4-(methoxymethoxy)cyclohexyl)ethyl)piperidine (180 mg, 84%) as a  
8 yellow oil.

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15 TFA (120 μL, 1.55 mmol) was added to a CH<sub>2</sub>Cl<sub>2</sub> (4 mL) solution of 2-(2-cyclohexyl-2-(4-  
16 (methoxymethoxy)cyclohexyl)ethyl)piperidine (175 mg 0.51 mmol). After stirring for 4 h the  
17 solvent was removed under reduced pressure and the crude dissolved in THF (3 mL). 1*N*  
18 aq. NaOH (1.5 mL) was added and the reaction was stirred for 1 h. H<sub>2</sub>O (5 mL) and EtOAc  
19 (10 mL) were added and the layers were separated. The organic layer was dried (Na<sub>2</sub>SO<sub>4</sub>)  
20 and concentrated under reduced pressure to give **24** (130 mg, 86%, purity > 96%) as a  
21 yellow amorphous solid. <sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>) δ = 3.93-3.83 (m, 1H), 2.98 (d, *J* = 12.1  
22 Hz, 1H), 2.53 (td, *J* = 9.6 and 2.6 Hz, 1H), 2.39 – 2.27 (m, 1 H), 2.26 – 2.22 (m, 1 H), 1.68-  
23 0.81 (m, 30 H); <sup>13</sup>C NMR (101 MHz, CDCl<sub>3</sub>) δ = 66.2, 57.0, 47.2, 44.5, 39.8, 39.4, 36.4, 33.1,  
24 32.9, 32.9, 31.9, 29.6, 27.0, 26.8, 26.7, 26.3, 25.2, 24.9, 23.3. *m/z* (ESI) 294 [M+H<sup>+</sup>]  
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### 38 ***In vitro* CPT-1 Inhibition Assay:**<sup>29</sup>

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40 Preparation of Rat Heart Mitochondrial Suspension (MS): All of the steps below were  
41 performed on ice or at 4 °C. Three male Wistar rats (8 weeks of age) were sacrificed by  
42 cervical dislocation, hearts were excised, rinsed twice with ice-cold Homogenization Buffer  
43 (HB: 10 mM Tris-HCl, 250 mM sucrose, 1 mM EDTA, pH 7.4), blotted, weighed, and finely  
44 minced with scissors for 5 min. Minced hearts were homogenized in 5 volumes of HB (by  
45 tissue weight) using a Polytron homogenizer. The homogenate was centrifuged for 10 min at  
46 600 x g, and the supernatant was removed and saved. The pellet was re-suspended in HB,  
47 homogenized again, spun for 10 min at 600 x g, and the supernatants from the first and  
48 second homogenizations were combined. The combined supernatant was centrifuged for 10  
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3 min at 7000 x g, the supernatant was discarded, the pellet was re-suspended in HB to the  
4 original homogenization volume and centrifuged again for 10 min at 7000 x g. The  
5 supernatant was discarded and the pellet of washed mitochondria was re-suspended in HB  
6 to 0.5 volumes relative to original heart weight. Protein content was determined using the  
7 Bradford protein assay, the solution was diluted with HB to a protein concentration of 10  
8 mg/mL, and small individual aliquots of this mitochondrial suspension (MS) were shock-  
9 frozen in LN<sub>2</sub> and stored at -80 °C. Aliquots of MS were thawed and diluted with HB to 2  
10 mg/ml protein just prior to use in CPT-1 assays.

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12 Each reaction (final volume 200 μL, pH 6.8) contained 0.04 mg of MS protein, 25 mM Tris-  
13 HCl, 150 mM sucrose, 1 mM EDTA, 60 mM KCl, 1.2 mg/mL lipid-free BSA, 0.1 mM 4,4'-  
14 dithiodipyridine, 0.4 mM L-carnitine, 0.04 mM palmitoyl CoA and varying concentrations of  
15 inhibitor. Reactions were performed in a quartz glass microcuvette, and inhibitors were pre-  
16 incubated for 3 min at 37 °C with all assay components except for L-carnitine and palmitoyl  
17 CoA. Each inhibitor was dried down from a CH<sub>2</sub>Cl<sub>2</sub> stock, redissolved in the assay buffer  
18 described above, and appropriate serial dilutions were made. A solution of L-carnitine and  
19 palmitoyl CoA (in water) was added to initiate the enzyme reaction, and the cuvette was  
20 placed in a spectrophotometer programmed in kinetic mode to read the increase in  
21 absorbance at 324 nm for 3.5 min. The slopes of the kinetic curves were calculated and  
22 compared to the mean of positive controls generated in the absence of any inhibitor (defined  
23 as 100 % activity), to determine the % control activity for each inhibitor concentration.  
24 Controls and each inhibitor concentration were analyzed with 5 replicates, and mean +/- SD  
25 were calculated for each point. IC<sub>50</sub> estimates and 95% confidence intervals for the malonyl  
26 CoA-inhibitable component of the CPT activity were determined following subtraction of the  
27 non-inhibited activity by non-linear regression of inhibition curves using GraphPad Prism 5  
28 (GraphPad Software Inc), and each value listed in Table 2 represents the mean IC<sub>50</sub> value  
29 and pooled confidence interval from two independent experiments using each inhibitor.  
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**Ex vivo Langendorff Model:**

Briefly, C57BL/6 (25 - 30 g) murine hearts were isolated and mounted onto the Langendorff perfusion system and retrogradely perfused with a modified Krebs buffer consisting of 1.2 mM sodium palmitate pre-conjugated to 3% bovine serum albumin (BSA).<sup>36</sup> The buffer was perfused at a constant pressure of 80 mmHg at 37 °C and continuously gassed with 95% O<sub>2</sub> / 5% CO<sub>2</sub>. Following 10 min stabilisation in control (BSA-Palmitate) Krebs buffer, hearts were perfused for 30 min with 1) control (BSA-Palmitate), 2) 2 µM perhexiline, 3) 2 µM **50**, or 4) 10 µM **50**. The maximal and minimal change in left ventricular pressure over time (LVdP/dt max and min, respectively) were calculated from the left ventricular pressure recorded throughout the 40 minutes of perfusion. All data are expressed as mean ± SEM and statistical comparison between groups were performed using 1-way ANOVA, followed by the post-hoc Bonferroni test. Values of  $P < 0.05$  were deemed as significant.

**Pharmacokinetics assays:**

Routine in vitro stability studies were conducted by Cyprotex Ltd (Macclesfield, UK) and in vivo pharmacokinetics were conducted by Sai Life (Pune, India) using standard methodology.

## ASSOCIATED CONTENT

The Supporting Information is available free of charge on the ACS Publications website at DOI: XXXXX. Experimental procedures, characterization of all intermediates and target compounds, and copies of NMR spectra of compounds **6a-j**, **31**, **42**, **50**, **55**, **62**.

## AUTHOR INFORMATION

Corresponding authors:

\*E-mail: [m.zanda@abdn.ac.uk](mailto:m.zanda@abdn.ac.uk); telephone: +44 (0)1224 437513

\*E-mail: [i.greig@abdn.ac.uk](mailto:i.greig@abdn.ac.uk); telephone: +44 (0)1224 437370

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5 *Notes:*

6  
7 M.P.F., M.Z. and I.R.G. are co-founders of Signal Pharma Limited that owns the I.P. of some  
8  
9 of the molecules described in this article.

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23 ABBREVIATIONS USED

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25 CYP2D6, cytochrome P450 2D6; CPT-1, carnitine palmitoyltransferase-1; CPT-2, carnitine  
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27 palmitoyltransferase-2; HF, heart failure; DMP, Dess-Martin periodinane; HLM, human liver  
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29 microsomes; RLM, rat liver microsomes; LVdP, Left Ventricular developed Pressure; SEM,  
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31 Standard Error of Mean.

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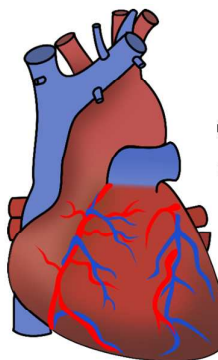
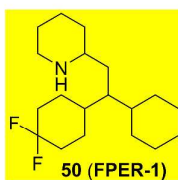


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## TOC

- Significantly increased *in vitro* metabolic stability
- Good *in vitro* potency
- Selective concentration in the myocardium
- Improvements in cardiac contractility and relaxation



*Perhexiline and  
FPER-1 effect*

Carbohydrate  
consumption ↑

Fatty acid  
consumption ↓

Myocardial  
efficiency ↑