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1 **Revealing chlorinated ethene transformation hotspots in a**
2 **nitrate-impacted hyporheic zone**

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15 **Abstract**

16 Hyporheic zones are increasingly thought of as natural bioreactors, capable of
17 transforming and attenuating groundwater pollutants present in diffuse baseflow. An
18 underappreciated scenario in the understanding of contaminant fate hyporheic zones is
19 the interaction between point-source trichloroethene (TCE) plumes and ubiquitous,
20 non-point source pollutants such as nitrate. This study aims to conceptualise critical
21 biogeochemical gradients in the hyporheic zone which govern the export potential of
22 these redox-sensitive pollutants from carbon-poor, oxic aquifers. Within the TCE plume
23 discharge zone, discrete vertical profiling of the upper 100 cm of sediment pore water
24 chemistry revealed an 80% increase in dissolved organic carbon (DOC) concentrations
25 and 20–60 cm thick hypoxic zones ($<2 \text{ mg O}_2 \text{ L}^{-1}$) within which most reactive transport
26 was observed. A 33% reduction of nitrate concentrations coincided with elevated pore
27 water nitrous oxide concentrations as well as the appearance of manganese and the TCE
28 metabolite *cis*-1,2-dichloroethene (cDCE). Elevated groundwater nitrate concentrations
29 ($>50 \text{ mg L}^{-1}$) create a large stoichiometric demand for bioavailable DOC in discharging
30 groundwater. With the benefit of a high-resolution grid of pore water samplers
31 investigating the shallowest 30 cm of hypoxic groundwater flow paths, we identified
32 DOC-rich hotspots associated with submerged vegetation (*Ranunculus* spp.), where low-
33 energy metabolic processes such as mineral dissolution/reduction, methanogenesis and
34 ammonification dominate. Using a chlorine index metric, we show that enhanced TCE to
35 cDCE transformation takes place within these biogeochemical hotspots, highlighting
36 their relevance for natural plume attenuation.

37 **Keywords:** hyporheic zone; terminal electron-accepting processes; chlorinated ethenes;
38 nitrate; dissolved organic carbon; natural attenuation.

39 **1 Introduction**

40 Chlorinated ethenes (CEs) such as trichloroethene (TCE) and its metabolites *cis*-1,2-
41 dichloroethene (cDCE) and vinyl chloride (VC) are among the most common volatile
42 organic contaminants (VOCs) detected in groundwater (Shapiro et al., 2004; Rivett et al.,
43 2012, Palau et al., 2014). These organohalides are prominent chemical stressors in
44 surface water ecosystems impacted by legacy industrial sources (Roy et al., 2018; Sonne
45 et al., 2018). Dissolved-phase CE plumes often continue to migrate in aquifers with
46 limited sorption capacity, and frequently discharge with groundwater baseflow to
47 streams and rivers (Weatherill et al., 2018). Back-diffusion from low permeability media
48 to groundwater means that many plumes will continue to persist well into or even
49 beyond this current century (Seyedabbasi et al., 2012). Hence, given their environmental
50 health significance and widespread abundance, it is little surprise that CEs are among
51 the most prevalent VOC class reported in surface water to date (e.g. Yamamoto, 2014;
52 Wittlingerová et al., 2016).

53 In complex landscapes with long histories of diverse land use practices, the
54 concurrence of multiple groundwater pollution incidences is increasingly common. For
55 instance, the natural attenuation and fate of point-source CE plumes which co-mingle
56 with nitrate (NO_3^-) from non-point agricultural sources has not been widely considered
57 at field scale to date (e.g. Bennet et al., 2007; Lu et al., 2017). In recent decades, global
58 groundwater NO_3^- concentrations have been steadily rising as a result of increasing
59 agricultural intensification (Gu et al., 2013; Wang et al., 2016). NO_3^- is a mobile anion
60 which only weakly adsorbs to clay minerals (Meghdadi, 2018), often resulting in high
61 loadings to vulnerable aquifers in agricultural regions (Ascott et al., 2017). This
62 groundwater excess of NO_3^- is a well-known global environmental concern linked to

63 eutrophication of receiving waters (Boyer et al., 2006) and blue-bay syndrome
64 (methaemoglobinaemia) in drinking water (Addiscott and Benjamin, 2006).

65 Before reaching surface water, these groundwater pollutants must transit the
66 aquifer-river interface or hyporheic zone (HZ) where groundwater and surface water
67 interact (Boano et al., 2014; Cardenas, 2015). The HZ offers considerable promise as a
68 passive biobarrier for in-situ 'treatment' of a broad range of organic groundwater
69 pollutants (Schaper et al., 2018; Gilevska et al., 2019) including CEs (Weatherill et al.,
70 2018). HZ sediments are often naturally rich in organic matter, resulting in hypoxic or
71 anoxic pore water conditions (Atashgahi et al., 2015) where CEs (such as TCE) and NO_3^-
72 are important terminal electron acceptors (TEAs) in dissimilatory microbial metabolism.
73 Here, fermentation of dissolved organic carbon (DOC) supplies electrons via dissolved
74 hydrogen (H_2) which drives a thermodynamic series of terminal electron-accepting
75 processes (TEAPs) (Heimann et al., 2009). Under DOC-limiting conditions, this TEAP
76 succession tends to follow an ecological succession which can be predicted by the Gibb's
77 free energy yield (ΔG°) [kJ mol^{-1}] of key redox couples: $\text{O}_2 \rightarrow \text{H}_2\text{O}$ [-238]; $\text{NO}_3^- \rightarrow \text{N}_2$ [-
78 240]; $\text{Mn}^{\text{IV}} \rightarrow \text{Mn}^{\text{II}}$; [-185]; $\text{Fe}^{\text{III}} \rightarrow \text{Fe}^{\text{II}}$ [-126]; $\text{TCE} \rightarrow \text{cDCE}$ [-121]; $\text{cDCE} \rightarrow \text{VC}$ [-101];
79 $\text{SO}_4^{2-} \rightarrow \text{HS}^-$ [-48]; $\text{HCO}_3^- \rightarrow \text{CH}_4$ [-32.8] (Wiedemeier et al., 1998; under conditions
80 reported therein).

81 From a thermodynamic perspective, NO_3^- reduction (denitrification) offers
82 considerably higher free energy yields than reductive dechlorination of TCE and cDCE in
83 heterotrophic metabolism (Weatherill et al., 2018). Hence, we hypothesise that elevated
84 groundwater NO_3^- will detrimentally impact the biogeochemical transformation capacity
85 of HZs where plumes discharge through agriculturally intensive landscapes. To test this
86 hypothesis, the present study aims to conceptualise critical biogeochemical gradients

87 which govern the HZ transformation potential of a TCE plume co-mingled with non-
88 point source NO_3^- . We apply a 'bottom-up' (Lansdown et al., 2016) multi-scale
89 hydrochemical monitoring approach to: (a) resolve in-situ biogeochemical gradients
90 controlling chemical fluxes from groundwater (b) identify potential hotspots of multi-
91 pollutant transformation and (c) evaluate the impact of regionally elevated NO_3^-
92 concentrations on chlorinated ethene transformation in the HZ.

93 **2 Materials and methods**

94 **2.1 Study area characteristics**

95 The study area is located in the lowland River Tern catchment ($2^\circ 53' \text{ W}$, $52^\circ 86' \text{ N}$)
96 in Shropshire (UK) where regional land use is dominated by intensive agriculture
97 (Krause et al., 2013). The catchment is underlain by highly permeable Permo-Triassic
98 sandstones with groundwater flow through a high porosity matrix (Shepley and Streetly,
99 2007) (Fig. 1). In this tributary of the River Severn, 76% of long-term river flow is
100 derived from groundwater storage (Marsh and Hannaford, 2008). The present study
101 focuses on a 40 m river reach that has been previously identified as the discharge zone
102 of a deep-seated groundwater TCE plume (Fig. 1) (Weatherill et al., 2014). The
103 unconfined regional sandstone aquifer is unlikely to significantly retard migration of the
104 plume (Smith and Lerner, 2008).

105 **2.2 Multi-scale hydrochemical monitoring approach**

106 Depth-discrete groundwater samples were collected from four sandstone bedrock
107 boreholes (Fig. 1) in March 2012 with passive grab sampling (HydraSleeve™) under
108 ambient borehole flow conditions (Weatherill et al., 2014). A 40 m reach-scale network
109 of multi-level mini-piezometers (MP1-MP25) (see Supporting Information Appendix A;

110 Figure S1) deployed in the riverbed were sampled in early August 2012 during a
111 summer baseflow recession period (S2). Reach-scale pore water chemistry samples
112 were extracted using dedicated air-tight syringes from PTFE sample tubes with point
113 screens in contact with sediment at 20, 40, 60, 80 and 100 cm below bed level (S1)
114 (Weatherill et al., 2014). Surface water was collected as grab samples from 20 cm above
115 the bed level from the upstream, mid-point and downstream sections of MP network
116 (S1). To provide a high-resolution horizontal window on shallow pore water chemistry,
117 a rectangular grid (0.6 x 2.8 m) of pore water samplers (PW1-PW13) with point-screens
118 at 5, 10, 20 and 30 cm depth was also included (S1, S2). In-situ luminescent dissolved
119 oxygen (DO) was measured in syringe and grab samples using an optical probe (Hach-
120 Lange, UK) with a detection limit of 0.01 mg O₂ L⁻¹.

121 **2.3 Laboratory analysis**

122 VOC samples were analysed using a headspace GC-MS-SIM method described in
123 Weatherill et al. (2014). Dissolved metabolic gas samples were collected in gas-tight
124 Exetainer (Labco, UK) vials fixed by a 50 mol ZnCl solution to inhibit microbial activity.
125 Methane (CH₄) and nitrous oxide (N₂O) were determined using N₂ headspace GC-
126 FID/μECD (Agilent GC7890A series) equipped with a 1 mL sample loop. Equivalent
127 dissolved phase concentrations were estimated from headspace gas volumetric
128 concentrations using Henry's law (Comer-Warner et al., 2018). Equivalent mass
129 concentration detection limits were <0.01 μg L⁻¹ for both gases. This GC method was not
130 suitable to detect ethene, the non-chlorinated end-product of TCE reductive
131 dechlorination.

132 Nitrate (NO₃⁻), nitrite (NO₂⁻), sulfate (SO₄²⁻) and chloride (Cl⁻), were determined by
133 anion chromatography on a Dionex ICS1000 (Dionex Corporation, UK). Ammonium (as

134 N-NH₄⁺) was analysed using an automated indophenol blue LCK304 method with a
135 detection limit of 0.015 mg N L⁻¹ (Hach-Lange, Germany). Total dissolved iron (Fe),
136 manganese (Mn) and silicon (Si) were determined from inline filtered (0.45 µm) and
137 acidified samples using inductively coupled argon plasma optical emission spectroscopy
138 (ICP-OES) (Varian Vista Pro MPX) with detection limits of <0.1 mg L⁻¹ for all elements.
139 DOC samples were filtered inline by 0.7 µm pre-combusted (550 °C) glass-fibre filters to
140 pre-combusted glass sample vials acidified with HCl and analysed on a Shimadzu TOC-
141 Vcpn analyser (Shimadzu Corporation, Japan) with a detection limit of 0.2 mg C L⁻¹.

142 **2.4 Data analysis**

143 Statistical analyses were performed with SPSS v19 (IBM, USA). Variable
144 distributions were tested using the Shapiro-Wilk test of normality to determine whether
145 parametric or non-parametric statistics were appropriate. Differences in means
146 between groups were analysed using paired T-tests or one-way analysis of variance
147 (ANOVA) for normally or log-normally distributed variables. All means reported were
148 back-transformed from their logarithms. Non-parametric Mann-Whitney U-tests were
149 used to test differences in medians for datasets which were neither normally nor log-
150 normally distributed. Principal component analysis (PCA) was performed on a subset of
151 pore water samples (n = 27) from the high-resolution pore water grid (S1, S2). PCA is
152 advocated as a key line of evidence for biogeochemical interpretation of aquifer CE
153 natural attenuation potential in recent guidance literature (Tarnawski et al., 2015).

154 **3 Results and discussion**

155 **3.1 Aquifer hydrochemistry**

156 Groundwater chemistry results are presented in Table 1. cDCE, VC, Fe, Mn NH₄⁺ and

157 NO_2^- were all below their detection limits and are not shown. TCE was the only CE
158 detected in the sandstone aquifer with the highest concentrations observed at the 80 m
159 deep HGA abstraction borehole, approximately 500 m distance from the river (Fig. 1).
160 The persistently elevated TCE concentrations ($>150 \mu\text{g L}^{-1}$) at depth in this open
161 borehole are believed to be representative of a deep-seated bedrock plume (Weatherill
162 et al., 2014). The plume is inferred to have a migration pathway along a curved axis
163 between the HGA/HGO boreholes in the up-gradient aquifer and HBE/HBW boreholes
164 on the riverbank (Fig. 1) (Weatherill et al., 2014). The groundwater environment of the
165 up-gradient plume was rich in high-energy TEAs including DO (median: 5.6 mg L^{-1}) and
166 NO_3^- (median: 55.3 mg L^{-1}) with moderate concentrations of SO_4^{2-} (median: 34.2 mg L^{-1}).
167 Groundwater NO_3^- was notably elevated in the riverbank boreholes ($73\text{--}81 \text{ mg L}^{-1}$).
168 Dissolved N_2O was oversaturated by up to 35 times for all groundwater samples with a
169 median concentration of $18 \mu\text{g L}^{-1}$. The highest N_2O levels (up to $47 \mu\text{g L}^{-1}$) were
170 associated with the shallower depths of the up-gradient sandstone aquifer. DOC
171 concentrations were very low in the aquifer ($<2 \text{ mg L}^{-1}$), with highly undersaturated
172 dissolved CH_4 concentrations with respect to air equilibrium ($<0.1 \mu\text{g L}^{-1}$).

173 **3.2 Vertical chemical gradients in the hyporheic zone**

174 An overview of maximum pore water TCE and cDCE concentrations in the
175 uppermost metre of riverbed sediment are presented in Fig. 2 with all CE results
176 reported in Supporting Information Appendix B (supplementary tables). cDCE was
177 detected in 10 samples with a maximum of $6.9 \mu\text{g L}^{-1}$ at MP7 at 40 cm. VC was only
178 confirmed twice at 20 cm in MP18 ($0.5 \mu\text{g L}^{-1}$) and at MP23 ($0.7 \mu\text{g L}^{-1}$). This plume
179 spatial variability in the riverbed sediments is consistent with previous monitoring
180 results in the discharge zone (Weatherill et al., 2014).

181 Vertical depth-chemistry profiles are presented in Fig. 3 with comparison of pore
182 water median concentrations at the base (100 cm) and top (20 cm) of the riverbed
183 sediment sequence reported in Table 2. We assume that groundwater discharge through
184 the HZ is predominantly vertical and that observed chemical gradients represent bulk
185 attenuation along groundwater flow paths through riverbed pore water pathways
186 (Krause et al., 2013). TCE concentrations exhibited an overall decline during transport
187 through the HZ. High-energy TEAs (DO and NO_3^-) and SO_4^{2-} concentrations in deep (100
188 cm) riverbed pore water samples were similar to the underlying aquifer (Table 2). A
189 step change in DO concentrations from 60–40 cm to shallower sample depths with
190 hypoxic conditions (e.g. $<2 \text{ mg O}_2 \text{ L}^{-1}$) was observed for many MP locations including the
191 cross-channel TA transect and MP18-23 (Fig. 2).

192 NO_3^- concentrations declined by one third from the aquifer median during discharge
193 through the HZ. Samples from a group of mini-piezometers (MP12, MP15 and MP16)
194 exhibited exceptionally high NO_3^- concentrations ($>130 \text{ mg L}^{-1}$) which were not
195 associated with the up-gradient plume (Table 1). Overall, groundwater flow paths
196 through the HZ were not a net source of dissolved N_2O in the plume discharge zone.
197 However, six hypoxic samples from 20 to 40 cm depth had elevated N_2O concentrations
198 ($>200 \mu\text{g L}^{-1}$) including MP6, MP7, MP18 and MP23. SO_4^{2-} concentrations did not vary
199 significantly during transport through the HZ.

200 Pathways through the HZ were observed to be a significant source of DOC and CH_4
201 with marked increases from 40–20 cm for DOC and 60–40 cm for CH_4 . These samples
202 coincide with the hypoxic locations at Transect TA and MP18-MP23 and detections of
203 dissolved Mn (up to 9.6 mg L^{-1}) and occasionally Fe. Under the near-neutral
204 groundwater pH conditions (Table 1), soluble Fe and Mn are assumed to be mostly a

205 product of microbially-mediated dissolution/reduction of Fe^{III}/Mn^{IV} mineral phases
206 present in the sediment solids (Tarnawski et al., 2015).

207 The multi-level chemical profiles obtained at MP7, MP18 and MP223 (Fig. 2) were
208 co-located with 100 cm sediment core samples from Weatherill et al. (2014). These
209 cores provide a physical context for observed biogeochemical gradients below the
210 sediment-water interface (Fig. 4). At each location, discrete changes in biogeochemical
211 conditions takes place in the upper 20 to 60 cm layer of sediment, which is composed of
212 medium sands (MP18 and MP23) and silty peat (MP7). Here, large declines in DO and
213 NO₃⁻ are associated with elevated DOC, CH₄ and N₂O and the appearance of Mn and cDCE
214 (Figs. 2 and 3). Our observations suggest that these hypoxic zones in the HZ can deplete
215 high-energy TEAs from groundwater sufficiently to enable lower energy TEAPs to
216 proceed allowing limited reductive dechlorination of TCE to cDCE.

217 **3.3 High-resolution pore water sampler grid**

218 A consistently reactive hypoxic zone was selected for a follow up high-resolution
219 investigation using a rectangular grid pore water samplers (S2). The spatial variability
220 of selected parameters are presented in S3 with all results reported in Appendix B
221 (Table T1). The grid was designed to target horizontal chemical gradients in the area of
222 MP7 (Fig. 4) where the highest cDCE concentrations have been observed in previous
223 work (Weatherill et al., 2014). The riverbed within the grid area was composed of
224 medium quartz sands influenced by submerged macrophyte cover (*Ranunculus* spp.).
225 Comparison of solute concentration means between 5, 10 and 20 cm sample depths
226 using single factor ANOVA did not reveal any statistically significant 'depth effect' (p =
227 >0.05) between depth-grouped samples. These results confirm that the sampler
228 network had intercepted pore water above the critical vertical redox gradient in the HZ

229 (Figs. 3 and 4).

230 TCE was present in all but four samples within the network with a maximum of 29
231 $\mu\text{g L}^{-1}$ at PW7 10 cm. cDCE was detectable in 75% of samples with six locations
232 exceeding the MP network maximum of $7 \mu\text{g L}^{-1}$ (Fig. 4) reaching up $16 \mu\text{g L}^{-1}$ at PW9 5
233 cm. VC was above detection limit in 27% of samples with a maximum of $2.3 \mu\text{g L}^{-1}$ also at
234 PW9 5 cm. All PW samples had DO concentrations $\leq 2 \text{ mg O}_2 \text{ L}^{-1}$ with little variation
235 across the network. NO_3^- was more spatially variable than DO reaching a maximum of 33
236 mg L^{-1} at PW5 10 cm. This high energy TEA was below detection limit in 13% samples
237 with a further 34% showing NO_3^- less than 5 mg L^{-1} . N_2O was spatially variable with
238 concentrations exceeding $100 \mu\text{g L}^{-1}$ in six samples mostly located in the upstream
239 samplers. Very low concentrations of N_2O ($< 1 \mu\text{g L}^{-1}$) were associated with locations
240 which were also characterised by low NO_3^- concentrations in the downstream group of
241 samples. These samplers also exhibited elevated NH_4^+ ($> 1 \text{ mg N L}^{-1}$). SO_4^{2-} exhibited a
242 greater range of concentrations than NO_3^- with a maximum of 66.4 mg L^{-1} at PW1 10 cm.
243 This SO_4^{2-} concentration is comparable to the overlying river water (Table 1) and may
244 indicate surface water infiltration to the HZ at the upstream samplers.

245 Many samples from the downstream group of samplers exhibited DOC
246 concentrations considerably exceeding 10 mg L^{-1} that of river water (Table 2) and those
247 from the MP network (Table 2, Figs. 3 and 4). As a product of low energy mineral-
248 reducing TEAPs, dissolved Mn was widespread with a large number of downstream
249 samples exhibiting Mn concentrations ($> 10 \text{ mg L}^{-1}$). Fe was also present with a range of
250 $1\text{--}4 \text{ mg L}^{-1}$ and a maximum of 12.7 mg L^{-1} at PW11 30 cm. Many samples from the
251 downstream part of the network exhibited CH_4 concentrations in excess of $1000 \mu\text{g L}^{-1}$
252 or more than twice that of the MP network maximum (Fig. 3).

253 Paired two sample T-tests were used to compare spatial means from paired
254 upstream (PW1–PW5) and downstream (PW9–PW13) sample locations (Table 3).
255 Significant declines ($p = <0.05$) are observed from upstream to downstream samples for
256 TCE, NO_3^- and N_2O coupled with significant increases in DOC, Mn and CH_4 . These results
257 suggest that high and low energy TEAPs may be spatially organised at sub-metre
258 horizontal scales within shallow HZ sediments.

259 **3.4 Evaluating in-situ plume transformation extent**

260 The chlorine index (CI) is a useful metric integrating the stoichiometric mass
261 balance of parent compound to daughter products into a single number (Harkness et al.,
262 2012; Freitas et al., 2015). The CI negates the confounding effects of dilution and
263 dispersion to evaluate parent compound transformation extent:

$$264 \quad \text{CI}_{\text{TCE}} = \frac{\sum W_i C_i}{\sum C_i}$$

265 Where W_i is the number of chlorine atoms in the CE molecule and C_i is the molar
266 concentration of the CE species present. The CI approach is a cost-effective alternative to
267 compound-specific stable isotope analysis (CSIA) (e.g. Gilevska et al., 2019) for single-
268 parent compound plumes (Freitas et al., 2015) such as the one in this study. The
269 relationship between total plume mass (in nmol) and CI_{TCE} are presented in Fig. 5
270 (samples where $\text{CI}_{\text{TCE}} = 3$ are omitted for clarity). Because it was not possible to include
271 ethene, the minimum CI_{TCE} possible is 1 in this study. From this figure, it can be seen that
272 all mini-piezometer (blue squares) and many samples from the upstream plot-scale grid
273 (open circles) show similar weak dechlorination extents with CI_{TCE} values >2.5 (partial
274 TCE to cDCE transformation). A greater degree of TCE dechlorination is indicated for
275 most samples from the downstream (black circles) and some mid-point samples (green
276 circles) with a CI_{TCE} 1.6 – 2.5 (Fig. 5) where much of the plume mass is transformed to

277 cDCE with limited cDCE reduction to VC.

278 **3.5 Principal component analysis of pore water chemistry**

279 Here, we use PCA to interpret the hypoxic biogeochemical environment in which
280 reductive dechlorination occurs with concomitant TEAPs that cycle nitrogen, carbon,
281 sulfur and minerals in the HZ (Tarnawski et al., 2015). PC1 and PC2 (Fig. 6) represent
282 linear combinations of the original variables from 27 PW network samples which
283 explain 73% of the observed variance. PC1 explains 44.3% and strongly associated with
284 Mn ($r = 0.92$), Si ($r = 0.91$), CH₄ ($r = 0.84$), NH₄⁺ ($r = 0.8$) and to a lesser extent DOC ($r =$
285 0.63). PC1 is interpreted as a biogeochemical *metabolite factor* associated which is co-
286 linear with the end-products of low energy TEAPs. This factor is spatially associated
287 with most downstream and some mid-point sample locations, where pore water
288 chemistry is dominated by mineral dissolution, Mn^{IV} reduction and methanogenesis.
289 Although NH₄⁺ is strongly correlated, PC1 is unrelated to NO₃⁻ which suggests that
290 ammonification of organic matter is the dominant nitrogen cycling process rather than
291 dissimilatory nitrate reduction to ammonium (DNRA) (Rivett et al., 2008). PC1
292 negatively correlates with Cl_{TCE} ($r = -0.73$) and to a lesser degree SO₄²⁻ ($r = -0.6$). Hence,
293 in this low-energy biogeochemical environment, TCE is a favourable TEA where
294 sediment/pore water hydrochemical interactions are enhanced.

295 PC2 accounts for 28.8% of the variance observed and is strongly associated with
296 oxidised nitrogen species including NO₂⁻ and N₂O ($r = 0.88$) and NO₃⁻ ($r = 0.76$). Given the
297 depleted NO₃⁻ concentrations and hypoxic pore water conditions present, NO₂⁻ and N₂O
298 are likely to be intermediates of in-situ denitrification (Rivett et al., 2008; McAleer et al.,
299 2017). NH₄⁺ only very weakly correlates with this factor ($r = 0.34$) which suggests that
300 DNRA may play a limited role in this environment. Hence, PC2 is interpreted as a

301 *denitrification factor* which is spatially associated with most of the upstream and some
302 mid-point samples. This factor is positively correlated with CI_{TCE} ($r = 0.47$) and
303 negatively correlated with DOC ($r = -0.5$). PC2 represents active NO_3^- reduction as a high
304 energy TEAP which is spatially independent of low energy carbon, nitrogen and mineral
305 cycling in the HZ.

306 **4 Discussion**

307 **4.1 Large scale hydrochemical gradients in the hyporheic zone**

308 Reach-scale observations from multi-level profiles within the top metre of bed
309 sediments suggest that the HZ is an important sink for high-energy groundwater TEAs
310 (Table 1) with bulk attenuation of 71% for DO and 33% for NO_3^- (Table 2). Pore water
311 TCE concentrations were observed to decline by 44% in the plume discharge zone
312 (Table 2). The deeper riverbed is characterised by discontinuous lenses of peat, clay and
313 silt overlying the bedrock aquifer which may be up to several metres thick (Weatherill et
314 al., 2014). From Fig. 5, clear differences can be seen in the vertical profiles of TCE where
315 peat is absent (MP18) and present (MP7). Given that metabolites (cDCE and VC) were
316 detected in just 9.5% of MP network samples, much of the observed pore water
317 attenuation may be attributed to non-reactive sorptive-diffusive transport processes
318 (e.g. Rivett et al., 2019).

319 Before reaching surface water, chemical fluxes from groundwater must transit
320 variable thickness bedforms composed of well-sorted quartz sands with porosities of
321 20% (Weatherill et al., 2014; Weatherill, 2015). Heat flow experiments have shown that
322 this pore water domain is in close hydraulic continuity with the overlying surface water
323 column (Angermann et al., 2012) due to the presence of the flow-confining structures

324 beneath (Gomez-Velez et al., 2014). Aquifer and river water Cl^- concentrations did not
325 contrast significantly (Fig. 3) and hence Cl^- could not be used as a conservative tracer to
326 delineate hyporheic exchange zone extent (e.g. Freitas et al., 2015). On the other hand,
327 SO_4^{2-} concentrations differed considerably (Fig. 3). Nonetheless, no evidence of surface
328 water mixing to a depth of 20 cm can be seen in the SO_4^{2-} profiles (Fig. 3) under baseflow
329 conditions. This precludes dilution from surface water infiltration as a factor in the
330 observed TCE attenuation under baseflow conditions in the MP network (e.g. Hamonts
331 et al., 2012).

332 Our findings suggest that in-situ reactive transport takes place within discrete
333 hypoxic envelopes associated with the TA transect MP locations (40–60 cm thick) and
334 MP18–MP23 where it is thinner (20 cm). This biogeochemically active zone is mostly
335 associated with the quartz sands layer and some peat-rich lenses in the underlying
336 cohesive deposits (Fig. 4). At MP6, MP7, MP18 and MP23, step-change declines in NO_3^-
337 are associated with greater than 10-fold increases in pore water N_2O concentrations
338 ($>200 \mu\text{g L}^{-1}$) (Figs. 3 and 4). N_2O production in the HZ is thought to be indicative of
339 partial NO_3^- reduction where pore water residence times are insufficient to allow
340 complete transformation to N_2 (Quick et al., 2016). These hypoxic zones are also
341 associated with an 80% increase in pore water DOC concentrations and a greater than
342 20-fold increase in CH_4 from the aquifer background (Table 2). The presence of CH_4 and
343 Mn indicate that low-energy TEAPs take place where high energy TEAs are locally
344 depleted (e.g. $\text{DO} < 2 \text{ mg O}_2 \text{ L}^{-1}$; $\text{NO}_3^- < 4 \text{ mg L}^{-1}$) (Fig. 4). These conditions appear to be
345 capable of supporting partial reductive dechlorination of TCE to cDCE (e.g. $\text{CI}_{\text{TCE}} > 2.5$)
346 (Fig. 5)..

347 **4.2 Evidence of multi-pollutant transformation hotspots**

348 Evidence of discrete horizontal organisation of dissimilatory TEAPs at sub-metre
349 scales was observed within the rectangular grid (Table 3). At the time of sampling, much
350 of the riverbed at the TA transect of the MP network as well as parts of the wider river
351 corridor was occluded by submerged *Ranunculus* spp. stands. In most upstream PW
352 network samples, active denitrification is suggested by depleted NO_3^- concentrations
353 ($<4\text{--}30\text{ mg L}^{-1}$) with elevated N_2O ($>100\text{ }\mu\text{g L}^{-1}$) and NO_2^- ($0.4\text{--}2\text{ mg L}^{-1}$). In some
354 downstream and some mid-point samples, oxidised nitrogen species are mostly absent
355 and the pore water chemistry is dominated by the metabolites of low-energy TEAPs
356 including mineral dissolution/reduction, methanogenesis and ammonification (Fig. 6).
357 From a thermodynamic perspective, this low-energy metabolic regime presents a
358 favourable environment for plume transformation which is demonstrated by the low
359 chlorine indices observed in the sub-group of PW samples where $\text{CI}_{\text{TCE}} = 1.6\text{--}2.5$ (Fig. 5).
360 Overall, the reductive dechlorination potential of the hyporheic zone appears to be
361 limited to cDCE production with only minor concentrations of VC detected which is
362 consistent with previous results (Weatherill et al., 2014). This incomplete natural
363 attenuation may be ascribed to the presence of residual DO concentrations, a lack of
364 suitable cDCE dechlorinator populations or competition for reducing equivalents by
365 metal-reducing bacteria (Chambon et al., 2013; Paul et al., 2016). The presence of both
366 NH_4^+ and CH_4 in hypoxic samples may favour alternative cometabolic pathways for cDCE
367 and VC mineralisation at the interface between hypoxic and oxic flow paths (Mattes et
368 al., 2010). CSIA and molecular microbial techniques may offer additional lines of
369 evidence to elucidate metabolic pathways and the activities of specific degraders (Badin
370 et al., 2016; Ottosen et al., 2019) and are advocated for future work.

371 DOC concentrations exceeding that of surface water and the underlying pore water
372 (e.g. $>10 \text{ mg L}^{-1}$) were observed in many PW samples from the PW network. This is
373 consistent with other reports of vegetated riverbeds as important sources of pore water
374 DOC with intensified hotspots of carbon cycling in the HZ as a result (e.g. Trimmer et al.,
375 2009; Ullah et al., 2014). Submerged macrophytes are thought to modify their local
376 aquatic environment by reducing river flow velocities resulting in increased fine
377 sediment deposition in spring and summer months (Sand-Jensen, 1998; Heppel et al.,
378 2009) with a consequent reduction in bed permeability and an increase in contaminant
379 exposure times in reactive zones (Oldham et al., 2013). Our results suggest that the
380 hypoxic sediments at the mid and down-stream parts of the network were influenced by
381 organic-rich fine sediment derived from allochthonous surface water catchment sources
382 (Ballantine et al., 2008), resulting in elevated in pore water DOC, Mn, Si and NH_4^+
383 concentrations. In addition, delivery of root exudate DOC in the rooting zone which cuts
384 across the HZ may play a role in enhanced biogeochemical cycling in vegetated
385 sediments (Ullah et al., 2014). Our observations show that both N_2O and CH_4 are useful
386 indicators for hotspots of high and low-energy TEAP activity respectively in the HZ.

387 **4.3 Implications of elevated nitrate on TCE transformation potential**

388 The large metabolic energy gain from reduction of NO_3^- implies that when DOC is in
389 excess, denitrification will take place in preference to lower energy reductive
390 dechlorination reactions in the HZ. Liu et al. (2015) demonstrated that NO_3^- additions
391 had an inhibitory effect on low energy TEAPs (SO_4^{2-} reduction and methanogenesis) in
392 sewer sediments. Denitrifying bacteria can maintain pore water H_2 thresholds (<0.1
393 nmol) far lower than necessary for reductive dechlorination (Weatherill et al., 2018).
394 However, considerable overlap in H_2 thresholds are observed experimental studies

395 under metal-reducing conditions with mineral bioavailability playing an important role
396 (Paul et al., 2016). Our findings suggest that reduction of NO_3^- and TCE to cDCE
397 dechlorination occur independently of one another which is supported by the PCA
398 results in Fig. 6.

399 Egli et al. (2010) suggest a minimum DOC concentration of 2 mg L^{-1} is required to
400 initiate catabolic genes involved in microbial contaminant transformation. Throughout
401 the up-gradient aquifer, DOC concentrations were at or below this threshold (Table 1).
402 The DOC resources required to reduce high-energy groundwater TEAs (DO and NO_3^-)
403 are estimated using the following stoichiometry (Tesoriero and Puckett, 2011):



406 With this stoichiometry, it is possible to estimate the regional DOC demand (mg C L^{-1})
407 imposed by high-energy TEAPs at overlapping spatial scales within the monitoring
408 networks. Fig. 7 plots observed DO concentration against high-energy DOC demand from
409 DO/ NO_3^- concentrations recorded in samples. Given the limited solubility of O_2 in
410 groundwater, the high-energy DOC demand is dominated by the elevated NO_3^-
411 background (Table 1, Fig. 3). By plotting observed DOC concentration for each sample
412 (denoted by the black diamonds) with the calculated DOC demand, it is possible to
413 identify locations in the HZ where DOC is in excess (Fig. 7). NO_3^- -DOC stoichiometric
414 relationships exert important controls on groundwater redox conditions that govern
415 nitrogen export patterns at landscape scale (Taylor and Townsend, 2012; Helton et al.,
416 2015). It is clear that the DOC required to reduce elevated groundwater NO_3^- produces
417 an electron donor-limited environment except for a subset of hypoxic PW samples. In
418 these locations, lower-energy metabolism is favoured including reductive dechlorination

419 of TCE to cDCE. Therefore, elevated NO_3^- levels are likely to impose a critical redox
420 'buffer' (e.g. Diem et al., 2013) which must first be overcome when CE plumes traverse
421 intensive agricultural regions underlain by vulnerable oxic groundwater systems.

422 **5 Conclusions**

423 With the application of a multi-scale 'bottom-up' monitoring approach, we have
424 shown that biogeochemical processes in the HZ can locally modify chemical fluxes and
425 redox conditions in discharging groundwater. Biogeochemically active hypoxic
426 pathways through the uppermost 20–60 cm of sediment pore water are capable of
427 reducing bulk groundwater discharge of NO_3^- by one third. In these zones, active
428 denitrification is indicated by depleted NO_3^- and elevated dissolved N_2O concentrations.
429 Using the chlorine index of TCE as a metric to evaluate in-situ plume transformation, we
430 have shown that enhanced dechlorination of TCE to cDCE occurs locally within DOC-rich
431 hotspots where pore water chemistry is dominated by the end-products of low-energy
432 microbial metabolism (Mn , CH_4 and NH_4).

433 Our results highlight the importance of DOC-rich pore water associated with
434 vegetated riverbeds for seasonal ecosystem service provision in the passive 'treatment'
435 of groundwater pollutants discharging from carbon-poor oxic aquifers. This local
436 electron donor excess is capable of overcoming the large stoichiometric demands for
437 carbon resources posed by high background concentrations of NO_3^- . Here, an ecological
438 succession of TEAPs is enabled where reductive dechlorination of CE species becomes
439 thermodynamically favoured. Below a critical DOC threshold, reduction of elevated
440 groundwater NO_3^- inhibits the transformation of CEs in dissimilatory microbial
441 metabolism. The inclusion pore water N_2O and CH_4 in combination with DO and

442 traditional redox indicators to identify respective high and low energy microbial
443 metabolic regimes is advocated further.

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621 **Tables and figure captions**

622 **Table 1:** Permo-Triassic sandstone groundwater hydrochemistry. Refer to Figure 1 for
 623 location of monitoring boreholes.

Borehole ID	pH	EC	DO	TCE	NO₃	SO₄	Cl	DOC	CH₄	N₂O
	pH units	μS cm ⁻¹	mg O ₂ L ⁻¹	μg L ⁻¹	mg L ⁻¹	mg L ⁻¹	mg L ⁻¹	mg C L ⁻¹	μg L ⁻¹	μg L ⁻¹
HGA 35 m	7.6	623	5.1	89	66.2	24.8	32.6	2.19	0.08	44.3
HGA 50 m	7.6	556	6.6	137	53.1	23.1	33.0	1.64	0.09	16.1
HGA 80 m	7.5	545	6.2	162	66.7	20	34.2	1.35	0.08	18.1
HGO 20m	7.8	498	5.6	48.8	55.3	21.2	37.9	2.01	0.09	28.3
HGO 40m	7.7	502	5	74.4	52.8	20.5	36.0	1.58	0.08	20
HBE	7.1	753	6.4	26.9	81.1	28.4	29.6	2.14	0.07	5.9
HBW	7.3	661	6.8	9.38	73	22	37.5	1.94	0.09	17.2

624

625

626 **Table 2:** Comparison of median solute concentrations using Mann-Witney U-tests from
 627 selected mini-piezometer samples at the bottom (100 cm) and top (20 cm) of the
 628 hyporheic zone (n = 16) including the aquifer (n = 7) and river (n = 3) for comparison.
 629 Significant changes are expressed as percentages.

Parameter	units	Aquifer	MP _{100cm}	MP _{20cm}	River	Sig. change	p
TCE	µg L ⁻¹	74.4	21.1	12.0	<0.01	-43%	0.01
DO	mg O ₂ L ⁻¹	6.3	6.4	1.9	9.6	-71%	<0.01
NO₃	mg L ⁻¹	60.8	60.3	40.1	37.4	-33%	<0.01
SO₄	mg L ⁻¹	22.6	25.4	28.1	61.0	-	>0.05
Cl	mg L ⁻¹	34.2	32.3	31.6	34.3	-	>0.05
DOC	mg C L ⁻¹	1.98	2.6	4.7	7.96	+80%	<0.01
N₂O	µg L ⁻¹	17.7	18.3	17.6	3.2	-	>0.05
CH₄	µg L ⁻¹	0.08	0.17	2.86	8.22	+1582%	<0.01

630

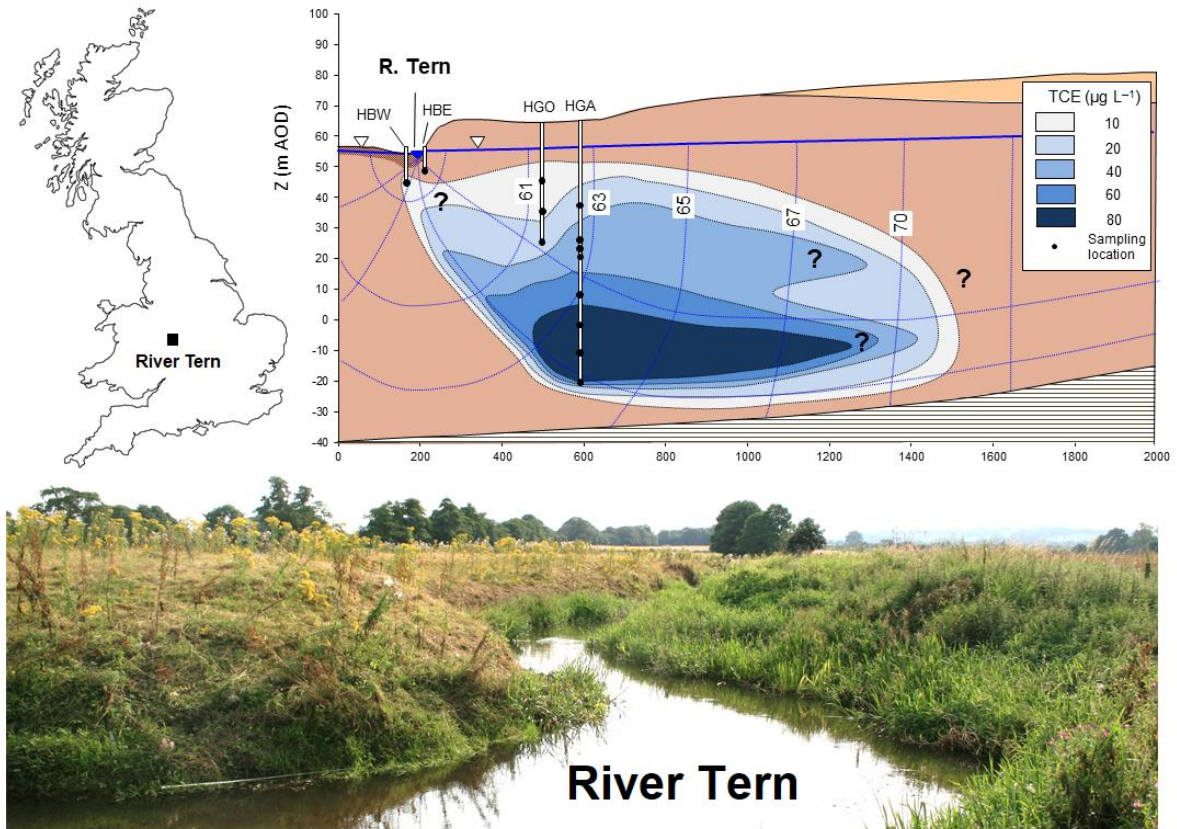
631

632 **Table 3:** Comparison of selected mean solute concentrations at paired
 633 upstream/downstream samples of the pore water grid using paired T-tests (n = 14)
 634 (standard deviations in parentheses). Significant changes are expressed as percentages.

Parameter	units	PW_{upstream}	PW_{downstream}	Sig. change	p
NO₃	mg L ⁻¹	14.1 (2.03)	5.1 (1.47)	-64%	<0.05
SO₄	mg L ⁻¹	38.8 (1.38)	26.5 (1.53)	-32%	0.032
Cl	mg L ⁻¹	29.9 (1.11)	32.6 (1.06)	-	>0.05
Mn	mg L ⁻¹	2.07 (1.89)	6.11 (3.07)	+195%	<0.05
DOC	mg C L ⁻¹	4.53 (1.38)	6.89 (1.58)	+32%	<0.05
N₂O	µg L ⁻¹	31.5 (5.43)	6.93 (4.45)	-78%	<0.05
CH₄	µg L ⁻¹	11.6 (2.90)	266.1 (2.15)	+2196%	<0.05

635

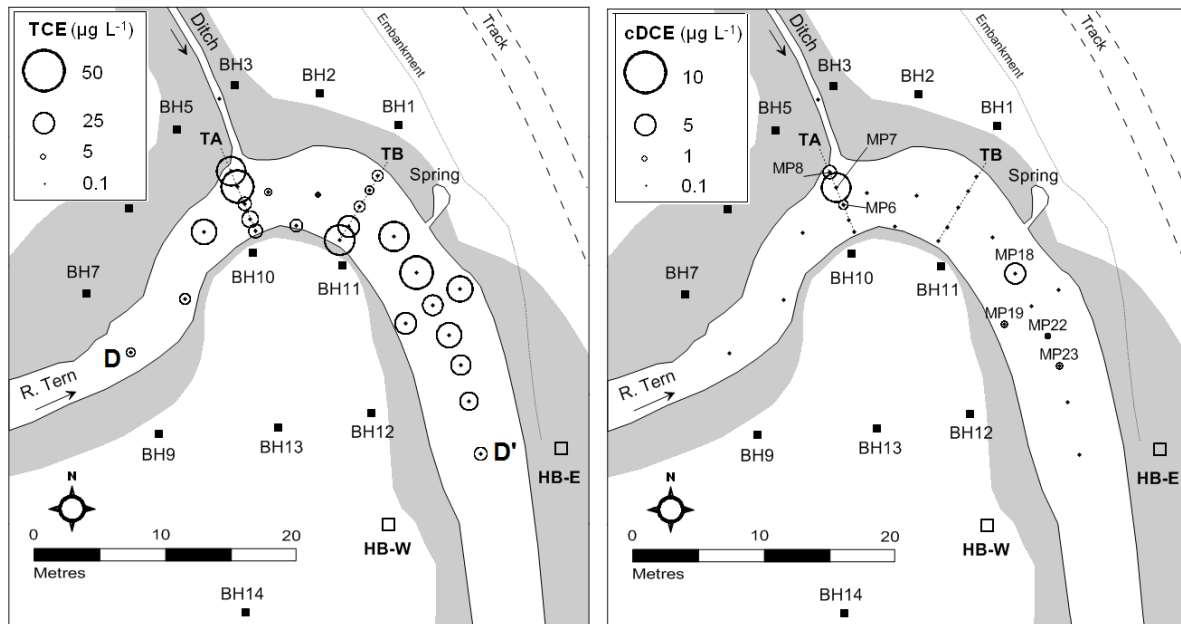
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637

638 **Figure 1:** Study area location and conceptual site model of regional aquifer and TCE
 639 plume and its inferred discharge zone at the River Tern (adapted from Weatherill et al.,
 640 2014). Black dots denote sampling depths in open boreholes using passive methods.

641

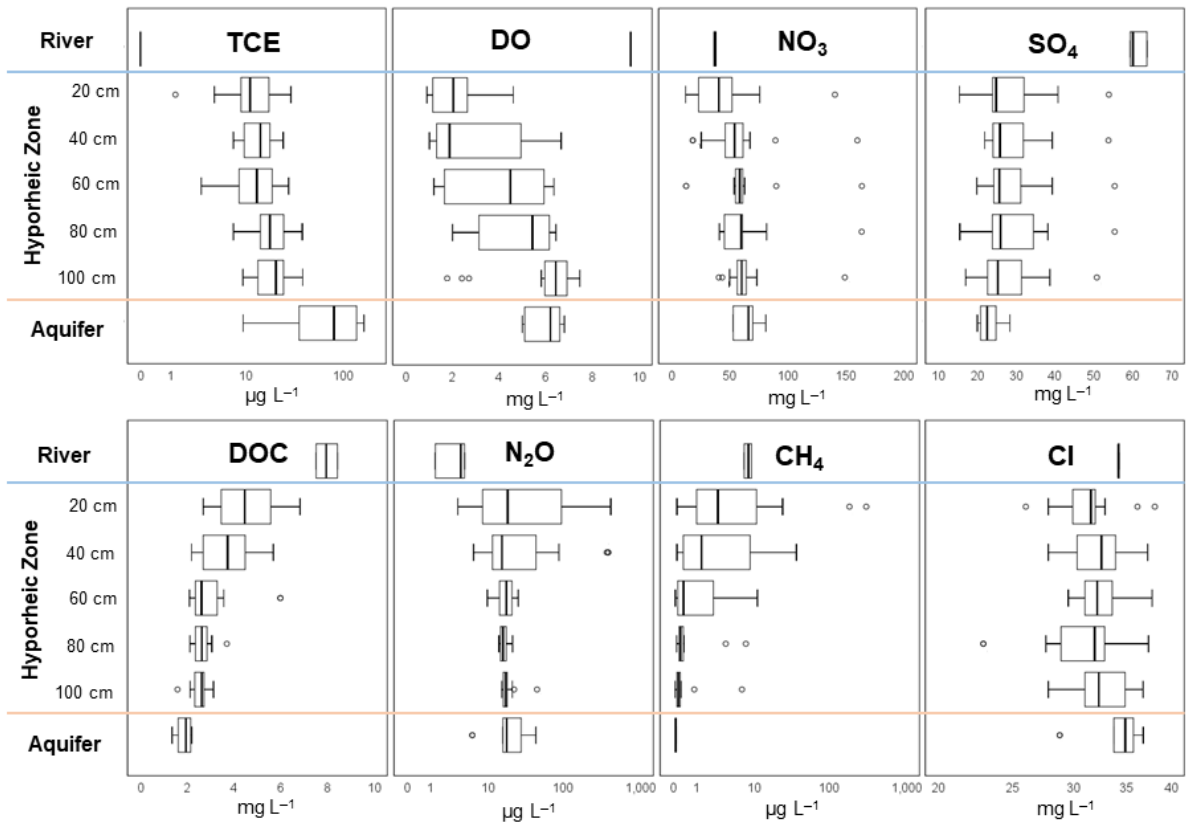


642

643 **Figure 2:** Spatial variability of maximum TCE and cDCE at mini-piezometer locations in
 644 the plume discharge zone in August 2012. Cross-channel plume concentration gradients
 645 are observed at the TA and TB mini-piezometer transects with the highest TCE
 646 concentrations occurring downstream of the TB transect. In-situ reductive
 647 dechlorination is indicated by cDCE detections with the highest concentrations at MP7
 648 and MP18.

649

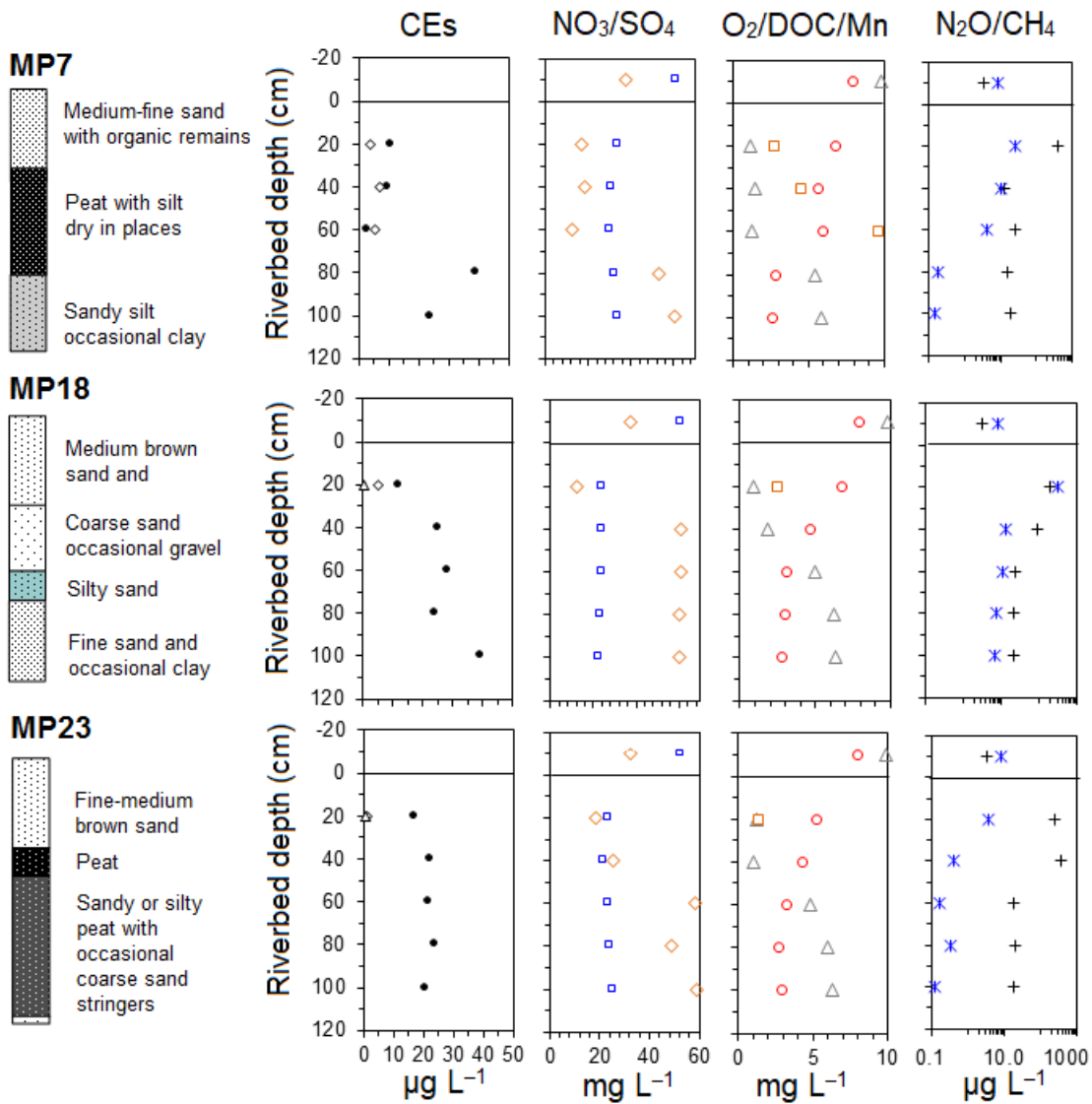
MP locations included: MP1-4, 6-8, 14, 15, 18-20, 22-25.



650

651 **Figure 3:** Vertical depth-concentration profiles for mini-piezometer locations used to
652 compare medians (n = 16) in comparison with the up-gradient aquifer (GW) (n = 7) and
653 surface water (SW) (n = 3).

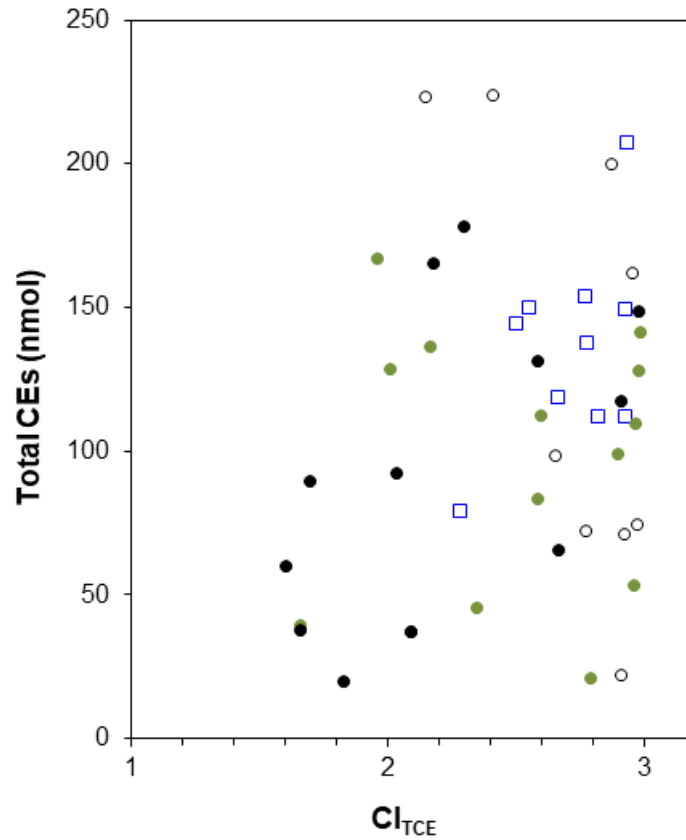
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655

656 **Figure 4:** Representative pore water biogeochemical profiles and riverbed sediment
 657 properties. TCE (black diamonds); cDCE (open black diamonds); VC (open black
 658 triangles); nitrate (open green diamonds); sulfate (open blue squares); DO (open
 659 triangles); DOC (open circles), Mn (open orange squares); blue stars (methane) and
 660 black crosses (nitrous oxide).

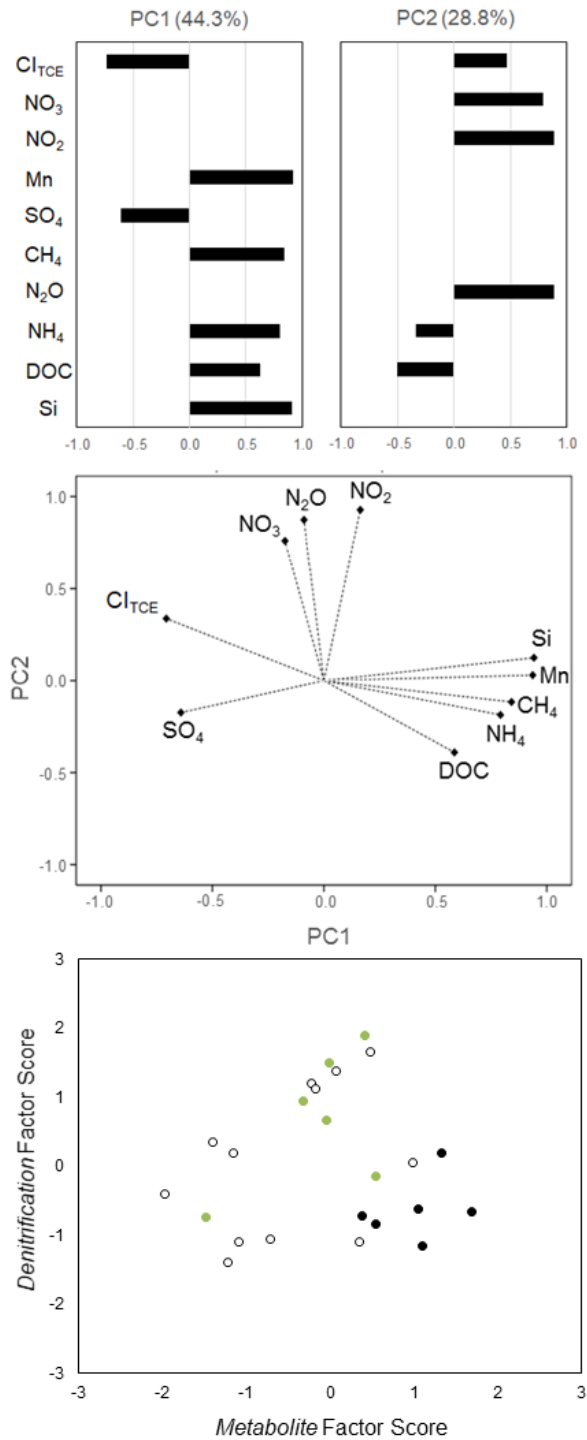
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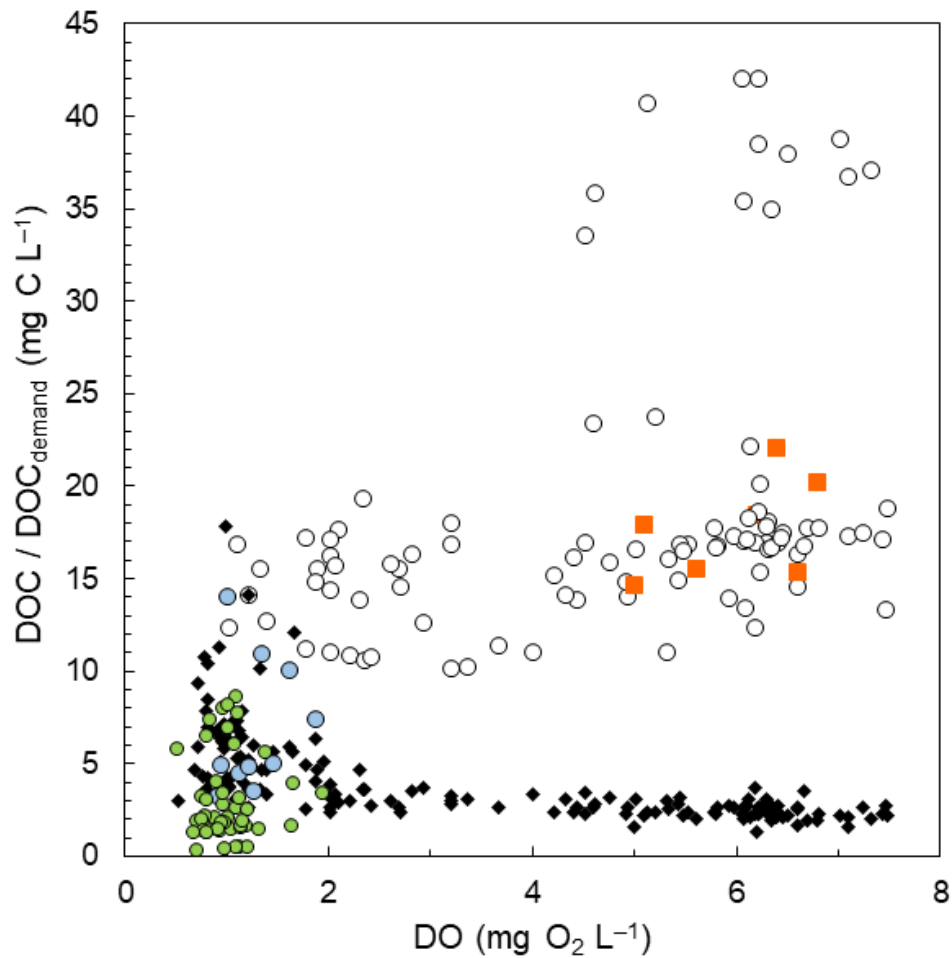
663 **Figure 5:** Relationship between total pore water CEs (in nmol) and the chlorine index of
 664 TCE (Cl_{TCE}) in all riverbed samples where $Cl_{TCE} = <3$. Blue squares (reach-scale mini-
 665 piezometer samples); open circles (upstream PW samples); green circles (mid-point PW
 666 samples); black circles (downstream PW samples).

667



668

669 **Figure 6:** Principle component analysis of high resolution pore water chemistry
 670 including factor loadings on original variables, ordination biplot and factor scores. The
 671 values in parenthesis is the percentage variance explained by PC1 and PC2. Factor
 672 scores are grouped as upstream (open circles), mid-point (green circles) and
 673 downstream (black circles) sample locations.



675

676 **Figure 7:** Calculated DOC demand for reduction of dissolved oxygen and nitrate in
 677 samples for the sandstone aquifer (orange squares); MP samples (open circles); MP
 678 samples showing TCE degradation (blue circles) and PW samples (green circles). Black
 679 diamonds denote observed DOC concentrations for each sample.