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1 **Effect of CeO₂ Nanoparticles on Plant Growth and Soil**
2 **Microcosm in a Soil-Plant Interactive System**

3
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20 [#]The two authors contributed equally to the manuscript.

28 **Highlights:**

29 Both CeO₂ NPs and Ce³⁺ ions stimulated cucumber roots growth.

30 Biotransformation of CeO₂ NPs occurred in root rhizosphere.

31 CeO₂ NPs and Ce³⁺ ions altered bacterial taxonomic and compositions.

32 CeO₂ NPs showed particle-specific effects.

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55 **Abstract**

56 The impact of CeO₂ nanoparticles (NPs) on plant physiology and soil microcosm and
57 the underlying mechanism remains unclear to date. This study investigates the effect of CeO₂
58 NPs on plant growth and soil microbial communities in both the rhizosphere of cucumber
59 seedlings and the surrounding bulk soil, with CeCl₃ as a comparison to identify the
60 contribution of the particulate and ionic form to the phytotoxicity of CeO₂ NPs. The results
61 show that Ce was significantly accumulated in the cucumber tissue after CeO₂ NPs exposure.
62 In the roots, 5.3% of the accumulated Ce has transformed to Ce³⁺. This transformation might
63 take place prior to uptake by the roots since 2.5% of CeO₂ NPs was found transformed in the
64 rhizosphere soil. However, the transformation of CeO₂ NPs in the bulk soil was negligible,
65 indicating the critical role of rhizosphere chemistry in the transformation. CeO₂ NPs
66 treatment induced oxidative stress in the roots, but the biomass of the roots was significantly
67 increased, although the Vitamin C (Vc) content and soluble sugar content were decreased and
68 mineral nutrient contents were altered. The soil enzymatic activity and the microbial
69 community in both rhizosphere and bulk soil samples were altered, with rhizosphere soil
70 showing more prominent changes. CeCl₃ treatment induced similar effects although less than
71 CeO₂ NPs, suggesting that Ce³⁺ released from CeO₂ NPs contributed to the CeO₂ NPs
72 induced impacts on soil health and plant physiology.

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74 **Keywords:** CeO₂ NPs; Transformation; Soil enzymes; Soil bacterial community;
75 Rhizosphere; Cucumber seedlings

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83 1. Introduction

84 The UN Food and Agriculture Organization and the World Bank are promoting the use
85 of nanotechnology as a sustainable technology to increase crop yields to feed the growing
86 population (Asadishad *et al.*, 2018). Engineered nanomaterials (ENMs) present great
87 potentials in the agricultural application (Chen *et al.*, 2021). For example, CeO₂ nanoparticles
88 (NPs) have shown their potential in crop protection due to their intrinsic antioxidative
89 capacity (Dai *et al.*, 2020). However, unlike application in other fields, the agricultural
90 application requires large scale and high quantities which raises concerns about their adverse
91 effects on the agricultural ecosystem (e.g. soil and plant health) as well as on animal and
92 human health (Zhang *et al.*, 2021).

93 The interactions between CeO₂ NPs and terrestrial plants have been extensively studied.
94 Priester *et al.* (2012) demonstrated that high concentrations (1000 mg/kg) of CeO₂ NPs
95 significantly reduced the yield of soybean by 22.5%. Lower concentration (200 mg/kg) of
96 CeO₂ NPs was reported to reduce the photosynthetic rate and CO₂ assimilation efficiency of
97 *Clarkia unguiculata*, possibly by disrupting energy transfer from photosystem II to the Calvin
98 cycle (Conway *et al.*, 2015). However, contradictory results found no phytotoxic effects on
99 *Cucumis sativus* in the Hoagland solution at concentrations up to 2000 mg/kg (Ma *et al.*,
100 2015). Moreover, at lower doses (100 mg/kg) nano-CeO₂ showed positive impacts on the
101 photosynthesis and growth of *Lactuca sativa* (Gui *et al.*, 2015).

102 Soil ecosystem is the most important sink of nanomaterials (Nowack and Bucheli, 2007).
103 Soil microorganisms are essential to many ecological functions, particularly in soil organic
104 matter decomposition and nutrient mineralization, which has greatly impact the growth of
105 terrestrial plants (Delgado-Baquerizo *et al.*, 2016). Wang *et al.* (2018b) reported that
106 long-term exposure (210 days) of activated sludge to 1 mg/L CeO₂ NPs induced the
107 deterioration of denitrifying process by reducing the abundance of some dominant
108 denitrifying bacteria such as *Acinetobacter* and *Flexibacter*. Pan *et al.* (2020) found that
109 CeO₂ NPs exposure with Fe amendment enhanced the abundances of several functionally
110 significant bacterial phyla including *Proteobacteria* and *Bacteroidetes*, which was associated

111 with C and N cycling. The microorganism in the soil plays important roles in maintaining
112 plant health. However, so far, most research has focused only on the impact of CeO₂ NPs on
113 plant species in the hydroponic culture system, with limited study investigating effect in the
114 soil-plant system.

115 The present study aims to evaluate the effect of CeO₂ NPs on plant growth and soil
116 microcosm in a soil-plant interactive system. Effects of the ionic form of Ce was studied as a
117 comparison to identify the contribution of the particulate and ionic form to the CeO₂ NPs
118 toxicity. The chemical species of Ce was determined by X-ray absorption near-edge
119 spectroscopy (XANES) to examine the role of biotransformation in the observed biological
120 effects. Bulk and rhizosphere soils were compared to understand the role of the rhizosphere
121 in the response of bacteria to exposure.

122 **2. Materials and Methods**

123 **2.1 Chemicals and Nanomaterials**

124 Ce(NO₃)₃·6H₂O and CeCl₃·7H₂O (purity of 99.9%) were purchased from Sinopharm
125 Chemical Reagent Beijing Co., Ltd. (China). CeO₂ NPs were synthesized using a
126 precipitation method described previously (Xie *et al.*, 2021). Briefly, 10 mmol of
127 Ce(NO₃)₃·6H₂O was added to 320 mL of NaOH solution (78 mmol·L⁻¹), followed by
128 vigorous stirring using a magnetic stirrer for 48 h. The resulting precipitate was collected by
129 centrifugation (15000 × g), followed by several washes with deionized (DI) water and
130 ethanol. The particle morphology, size, crystal structure, hydrodynamic diameter, and zeta
131 potential in DI water and Hoagland nutrient solution, surface chemical valence states were
132 characterized using Transmission electron microscopy (TEM, JEM 200CX, Japan), powder
133 X-ray diffraction (XRD, X'pert PRO MPD, Holland), and X-ray photoelectron spectroscopy
134 (XPS, Thermo ESCALAB 250XI, USA) and dynamic light scattering (DLS, Zetasizer Nano
135 ZS90, UK), respectively.

136 **2.2 Dissolution of CeO₂ NPs**

137 The dissolution of CeO₂ NPs in DI water was analyzed by measuring the Ce³⁺ released

138 into the solution. Briefly, CeO₂ NPs suspensions (100 mg/L) in 25 mL deionized H₂O were
139 prepared and incubated for 48 h at 37 °C, followed by centrifuging at 11,000 g for 15 min.
140 The supernatants were collected and diluted with 2% nitric acids for ICP-MS analysis
141 (Thermo Elemental X7). A range of Ce standard solutions (0.1, 1, 5, 10, 50, 100, 500 µg/L)
142 were also measured for calibration. The recovery rates of Ce was tested to be 99.87%.

143 **2.3 Plant-Soil System Exposure and Sample Collection**

144 Silt loam soil (13% clay, 55% silt, 30.9% sand, and 1.1% organic matter content, pH
145 7.85) was collected from the Shangzhuang Experimental Station of China Agricultural
146 University and air-dried, followed by sieving through a 2 mm mesh and stored at 4°C.
147 Cucumber seeds (*Cucumis sativus*, Zhongnong NO.16) were purchased from the Chinese
148 Academy of Agricultural Sciences. 30 plastic pots (6.0 cm diameter×5.3 cm height) filled
149 with 60 g of the sieved soil were divided into six equal groups for different treatments:
150 unamended control, CeO₂ NPs treatment at 5.8 mmol kg⁻¹ (1000 mg/kg), and CeCl₃·6H₂O
151 treatment at 0.6 mmol kg⁻¹ (100 mg/kg), with and without plant seedling. The concentration
152 of 100 mg/kg ionic Ce was chosen under the assumption that 10% of the CeO₂ NPs would be
153 dissolved (Pagano *et al.*, 2016).

154 Cucumber seeds were germinated on moist paper towels for 4 d. Then 15 uniform
155 seedlings were selected and transferred to the corresponding pots (planted pots). The
156 remaining 15 pots were left unplanted. Then 10 mL CeO₂ NPs suspension, CeCl₃ solutions,
157 and Hoagland solution (control group) were applied in each treatment (day 1). Hoagland's
158 solution was used to water the pots every day. Both planted and unplanted pots were
159 cultivated in a climate chamber with 16 h photoperiod (light intensity of 1.76×10⁴ µmol/m² s),
160 25°C/18°C day/night temperature and 50%/70% day/night humidity.

161 Twenty days after transplanting (day 21), samples of soil and plants were harvested. The
162 soil on the root surface was manually removed and collected as rhizosphere soil. Soils from
163 the unplanted treatments were used as bulk soil samples. A portion of one soil sample was
164 stored at 4 °C for enzymatic activity measurements, and the remainder was stored at -80 °C to
165 characterize the soil bacterial community structure. Fresh plants were collected and the

166 physiological response was measured immediately. For other measurements, the plants are
167 washed, dried at 60 °C, and then weighed to acquire constant weight.

168 **2.4 Plant Physiology**

169 At day 21, the relative chlorophyll content of the cucumber leaves was measured before
170 harvest using a Konic Minolta SPAD-502 Plus (Konica Minolta Optics, Japan). Total soluble
171 sugar was determined according to the method described by Buysse and Merckx (1993). Leaf
172 nitrate-N content was analyzed by a colorimetric method (Cataldo *et al.*, 1975). Soluble
173 protein concentrations in the roots and leaves were determined using the Pierce BCA Protein
174 assay kit (Thermo Scientific). The content of Vitamin C was analyzed using an assay kit
175 (Nanjing Jiancheng Bioengineering Institute, China) according to the manufacturer's
176 instructions.

177 **2.5 Element Measurement of Ce and Mineral Nutrient in plant tissues**

178 To quantify the macro-and micro-nutrient contents (K, Ca, Na, Mg, Fe, S, P, Cu, Zn, Mn,
179 and Mo) and concentration of Ce in plants, dried roots, stems, and leaves were ground into
180 fine powders and digested with a 3:1 (v:v) mixture of HNO₃ (75%) and H₂O₂ (30%) on a
181 heating plate (80 °C for 1 h, 120 °C for 3 h, and 160 °C for 0.5 h). Elemental concentrations
182 in the digestion solution were then analyzed by ICP-MS or inductively coupled plasma
183 optical emission spectroscopy (ICP-OES, Perkin Elmer). Standard solutions (0.5-50 mg/L)
184 containing all of the selected elements were used for external calibration. Blanks were
185 analyzed between every six samples. Spiking recovery experiments and analysis of certified
186 reference materials (GBW 07602 and GBW07603 Bush Branches and Leaves) were
187 performed for analytical method validation. Recoveries and detection limits for all of the
188 elements are reported in **Table S1**. The recoveries for all elements were between 93.1% and
189 111.5% with a relative standard deviation of < 1.5% (**Table S2**).

190 **2.6 Stress Response of Cucumber to CeO₂ NPs and CeCl₃**

191 Fresh roots, stems and leaves were excised, homogenized with cold phosphate-buffered
192 saline (PBS) (50 mM, pH 7.8), and centrifuged at 10000 × *g* at 4 °C for 10 min. The
193 supernatants were collected for analyses of superoxide dismutase (SOD), peroxidase (POD),

194 catalase (CAT) activities, and the malondialdehyde (MDA) contents according to the
195 manufacturer's instructions (Nanjing Jiancheng Bioengineering Institute, Nanjing, China).

196 **2.7 Ce Speciation Analysis by XANES**

197 To analyze the chemical species of Ce in plant roots and soils, all samples were ground
198 to fine powders and pressed into thin slices (~2 mm). Ce *L_{III}* edge (5723 eV) spectra were
199 recorded at ambient temperature in fluorescence mode at beamline 1W1B of the Beijing
200 Synchrotron Radiation Facility. The storage ring was run at 2.5 GeV with a current intensity
201 of 50 mA during the spectra collection. XANES spectra of the reference compound CeO₂
202 NPs and CePO₄ were also collected. Linear combination fitting (LCF) analyses of the
203 XANES spectra were performed on the software program ATHENA to identify and quantify
204 Ce species.

205 **2.8 Determination of Enzymatic Activity in Soil**

206 Acid phosphatase, β -D-glucosidase, and arylsulfatase activities were quantified using
207 the method described by Saiya-Cork *et al.* (2002). Urease activity was evaluated by
208 measuring the release of NH₃-N (mg) per gram of dry soil in 24 h (Yang *et al.*, 2007).
209 Dehydrogenase activity was tested by a method for reductive generation of triphenyl
210 formazan (TF), expressing as TF (mg) per gram dry soil in 24 h (Ross, 1971). Peroxidase
211 activity was expressed as the amount of quinone in mg formed per g dry soil in 2 h (Mi and
212 Kim, 1994). Invertase activity was determined with sucrose as a substrate, based on
213 3,5-dinitrosalicylic acid colorimetry to detect glucose (mg) per gram dry soil in 24 h (Yang *et*
214 *al.*, 2006).

215 **2.9 DNA extraction analysis**

216 Total DNA was extracted from a 0.3 g soil sample using a Power Soil DNA extraction
217 kit (MO BIO Laboratories, Carlsbad, CA, USA). The 16S rDNA V4 region of the sample is
218 amplified by the specific primers with Barcode in the designated sequence area, which is 515F
219 (5'-GTGCCAGCMGCCGCGGTAA-3') and 806R (5'-GGACTACHVGGGTWTCTAAT-3'). The
220 polymerase chain reactions (PCR) were conducted using the following temperature profiles:
221 denaturation at 94 °C for 5 min, followed by 30 cycles of amplification at 94 °C for 1 min, 48

222 °C for 1 min and 72 °C for 1 min, followed by a final extension at 72 °C for 10 min, and
223 finally held at 4 °C. In addition, soil samples from triplicates were mixed, with the DNA
224 extracted and amplified. The amplified products were separated by 1.5 % agarose
225 electrophoresis. Purified amplicons were obtained using a QIAquick PCR purification kit
226 (Qiagen, Valencia, CA, USA), and concentrations were determined on GE NanoVue System
227 (Thermo Scientific). Then a library was constructed using Illumina TruSeq DNA PCR-Free
228 Sample Prep Kit. The paired-end sequencing was performed at Beijing Genome Institute,
229 Beijing, China, using a paired 150 bp MiSeq 2000 sequencing system (Illumina, San Diego,
230 CA, USA) according to the manufacturer's instructions.

231 **2.10 Statistical analysis**

232 All statistical analyses were conducted using the SPSS 19.0 statistical software package
233 for Windows (SPSS, Chicago, IL, USA). Data were expressed as mean \pm standard deviation
234 (SD). A one-way analysis of variance (ANOVA) was performed to compare the significance
235 of differences between different groups. The significance levels (*, # $P < 0.05$, **, ## $P <$
236 $0.01/0.001$) between the different treatments and the control were determined by the Fisher
237 Least Significant Difference (LSD) test.

238

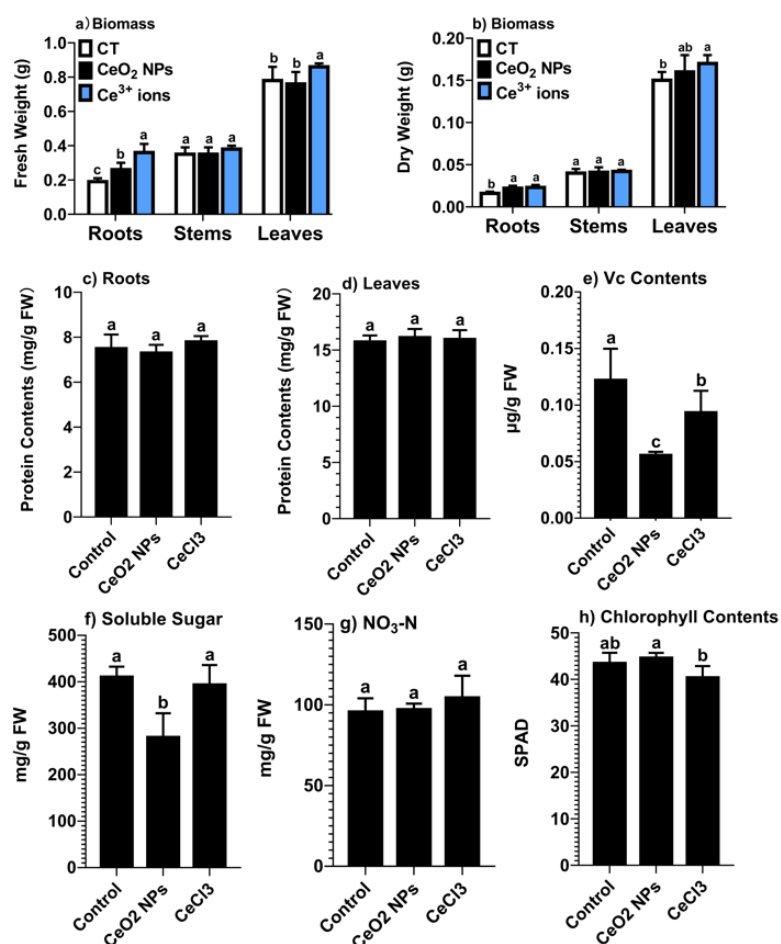
239 **3. Results and Discussion**

240 **3.1 Characterization of CeO₂ NPs**

241 The average particle size of CeO₂ NPs is 5.1 ± 0.8 nm (**Fig. S1a**). XRD analysis showed
242 that the CeO₂ NPs have a cubic fluorite structure (**Fig. S1b**). XPS spectra show that the
243 percentage of surface Ce³⁺ is 4.7% (**Fig. S1c**). The hydrodynamic size in DI water and
244 Hoagland's solution were 653 ± 166 nm and 1059 ± 139 nm, respectively. The ζ potential of
245 CeO₂ NPs in DI water and Hoagland's solutions were 5.75 ± 0.13 mV and -1.38 ± 0.26 mV,
246 respectively (**Fig. S1d**). The solubility of CeO₂ NPs (100 mg/L) in water was very low (<
247 0.1%).

248 **3.2 Plant physiological responses to CeO₂ NPs and CeCl₃ exposure**

249 As shown in **Fig. 1**, CeO₂ NPs significantly increased the biomass (fresh and dry weight)
 250 of cucumber roots but not stems and leaves (**Fig. 1a, b**). However, the organic nutrient
 251 contents were reduced. The contents of Vc and soluble sugar in leaves were reduced by 54%
 252 ($P < 0.01$) and 32% ($P < 0.05$), respectively, by CeO₂ NPs treatment (**Fig. 1e, f**). The total
 253 soluble protein, nitrate-N content, and chlorophyll contents in cucumber leaves were not
 254 affected (**Fig. 1c, d, g, h**). Similar to CeO₂ NPs, CeCl₃ induced similar trends of changes in
 255 the biomass of roots and Vc contents, suggesting that CeO₂ NPs and CeCl₃ share a similar
 256 effect and indicating that dissolution might partly contribute to the impacts of CeO₂ NPs.
 257 Therefore, the transformation of CeO₂ NPs in soil and plant was analyzed next.



258
 259 **Fig. 1** Phenotypes and contents of organic nutrients after CeO₂ NPs and CeCl₃ exposure for 20 days. **a)**
 260 Fresh weight and **b)** dry weight of plant roots, stems, and leaves, respectively. **c)** and **d)** Soluble protein of
 261 the cucumber seedlings treated with CeO₂ NPs and CeCl₃ ions in roots and leaves. **e), f), g),** and **h)** are the
 262 contents of Vc, soluble sugar, and nitrate-N content of the cucumber seedlings and relative chlorophyll
 263 contents (SPAD) in leaves treated with CeO₂ NPs and CeCl₃ ions. Different lowercase letters indicate

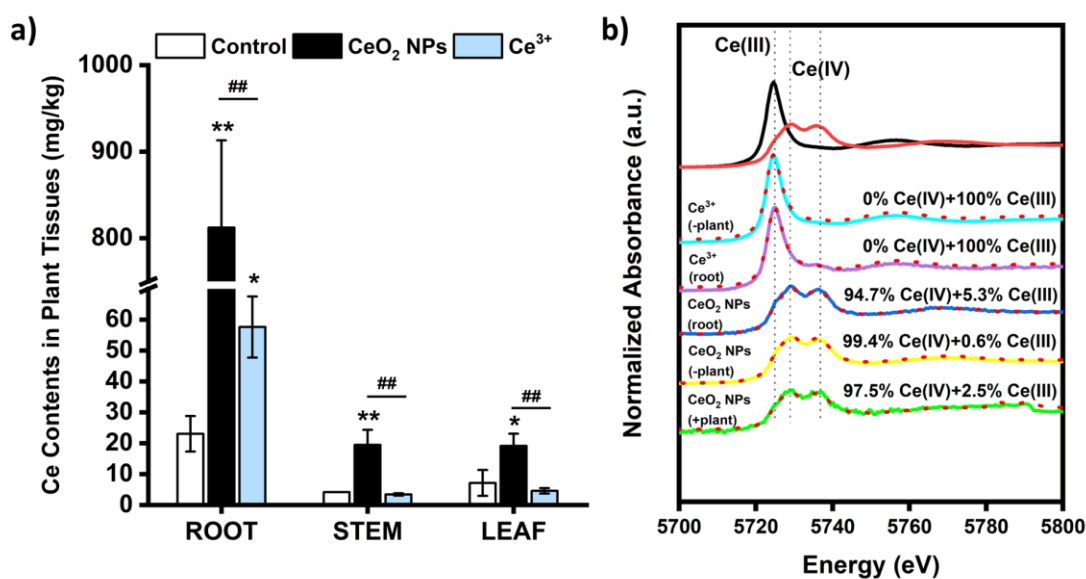
264 significant difference between different groups at $p < 0.05$ ($n = 6$).

265 **3.3 Distribution and chemical species of Ce in plant tissues**

266 Ce accumulated in the roots (812.2 ± 100.8 mg/kg), stem (19.5 ± 4.9 mg/kg) and leaves
267 (19.1 ± 3.9 mg/kg) of cucumber after CeO₂ NPs exposure. However, CeCl₃ treatment only led
268 to the accumulation of Ce in the root (57.6 ± 9.9 mg/kg), no upward translocation was
269 observed (**Fig. 2a**). Such difference might be related to the different translocation behavior of
270 particles and ions. Most of the Ce³⁺ can be easily fixed as CePO₄ on the root surface by the
271 PO₄³⁻ from the nutrients in soils, therefore, there was little chance to go upward. However,
272 the NPs usually can move upward easily with water flow (Zhang *et al.*, 2011), as
273 demonstrated by the XANES data showing that most of the Ce entering the plant roots was in
274 the form of particles (94.7%) (**Fig. 2b**).

275 Biotransformation of CeO₂ NPs is more likely to occur around the rhizosphere than in
276 the region far away from the root because it mainly occurs in acidic environment and usually
277 requires reducing agents (Rico *et al.*, 2018; Xie *et al.*, 2019). Root exudates and soil
278 microorganisms in the small rhizosphere region are considered to play crucial roles in the
279 reduction of CeO₂ NPs and the release of Ce³⁺ ions (Zhang *et al.*, 2012; Zhang *et al.*, 2017).
280 Our study found that, in the rhizosphere soil, 2.5% of CeO₂ NPs was in the form of Ce(III),
281 while only a little fraction of Ce(III) (0.6%) was observed in the bulk soil (**Fig. 2b**),
282 suggesting the crucial role of rhizosphere chemistry in the transformation of CeO₂ NPs.

283



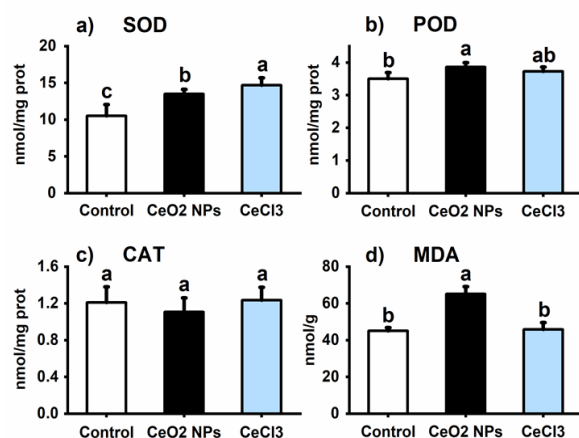
284

285 **Fig. 2 a)** Cerium contents in the root, stem, and leaf treated with CeO₂ NPs or CeCl₃ at day 21. Data are
 286 expressed as mean \pm SD (n = 6) an average of six replicates. * and ** indicates a significant difference
 287 at $p < 0.05$ and $p < 0.01$ (n = 6) compared with the control, respectively. ## indicates a significant different
 288 at $p < 0.01$ (n = 6) between CeO₂ NPs and CeCl₃ treatments; **b)** XANES normalized Ce L_{III} edge spectra of
 289 reference compounds (CePO₄ and CeO₂) and samples. (-plant) and (+plant) indicates bulk soil and
 290 rhizosphere soil, respectively. (root) means CeO₂ or CeCl₃ enriched in the root and we detected the root
 291 samples by XANES.

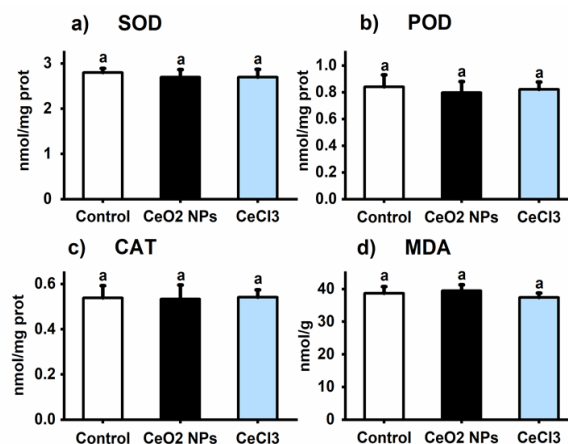
292 3.4 Antioxidative Response in Plants

293 Since Ce accumulated significantly in plant tissues under CeO₂ NPs exposure, we next
 294 examined the antioxidative responses of cucumber shoot and root after exposure. In the
 295 shoots, neither treatment-induced any changes in SOD, POD, and CAT activities and the
 296 MDA contents (**Fig. 3B**). However, in the roots, the SOD and POD activities as well as the
 297 MDA contents were significantly increased after CeO₂ NPs treatment (**Fig. 3A**), indicating
 298 the oxidative damage of cell membrane and activation of the plant defense system. In the
 299 CeCl₃ group, only SOD activity was increased. This indicates that the antioxidative responses
 300 found for CeO₂ NPs were mainly particle-specific effects. The triggered antioxidative
 301 response and oxidative stresses by CeO₂ NPs contributed to the reduction of organic nutrients
 302 in shoots. However, Ce³⁺ ions only triggered little antioxidative responses which may explain
 303 the insignificant change of organic nutrient contents.

A Roots



B Shoots



304

305 **Fig. 3** SOD, POD, CAT activities, and MDA contents in root (a) and shoot (b) after exposure to CeO₂ NPs
306 and CeCl₃ for 20 days. Different lowercase letters indicate significant difference between different groups
307 at $p < 0.05$ ($n = 6$).

308 3.5 Alteration of Mineral Nutrient Homeostasis

309 Higher plants need at least 14 mineral elements to support their growth and reproduction
310 (White and Brown, 2010; DalCorso *et al.*, 2014). Deficiency or overload of any elements
311 may lead to growth impairment or physiological disorders such as necrosis or chlorosis. To
312 further investigate the effect of CeO₂ NPs and Ce³⁺ ions, we measured the uptake of several
313 key nutrient elements that are essential for plant growth. Our results showed that both CeO₂
314 NPs and CeCl₃ treatment influence the balance of the mineral element levels. Ce
315 accumulation led to the imbalance of several key nutrient elements that are essential for plant
316 growth. The effect on the element contents in different tissues was different as shown in the
317 heatmap (**Fig. 4**). In roots, CeO₂ NPs significantly increased the K, Mg, and Mo contents
318 while reduced the Ca, S, P, Cu, and Zn contents. In stems and leaves, results show that the
319 amounts of Ca, Fe, Cu, Zn contents decreased while Mo increased. In general, the effects of
320 CeO₂ NPs and CeCl₃ on the mineral homeostasis in the cucumber seedlings were similar.

321

	Roots		Stems		Leaves		
	CeO ₂ NPs	CeCl ₃	CeO ₂ NPs	CeCl ₃	CeO ₂ NPs	CeCl ₃	
K	1.45*	1.28*	1.01	1.03	1.02	1.02	0.4
Ca	0.76*	0.79*	0.89*	1.00	0.87*	0.93	0.55
Na	1.12	0.81*	0.89*	0.83*	1.07	0.96	0.7
Mg	1.26*	1.35*	1.05	1.06	1.03	0.94	0.85
Fe	0.98	1.09	0.78*	0.95	0.68**	1.20*	1
S	0.92*	1.06	1.01	1.13	1.03	0.96	1.15
P	0.91*	0.98	0.93*	0.98	0.96	0.91*	1.3
Cu	0.41**	0.53**	0.93*	0.99	0.88*	0.81*	1.45
Zn	0.51**	0.52**	0.72**	0.94	0.71**	0.62**	1.6
Mn	0.97	1.04	1.02	1.31**	1.03	1.35**	1.75
Mo	1.75**	1.59**	1.42**	1.48**	1.46**	1.37*	1.9

322

323 **Fig. 4** Heatmap showing the changes of inorganic nutrients in roots, stems, and leaves after CeO₂ NPs and
324 CeCl₃ exposure. K, Ca, Na, Mg, S, and P were determined by ICP-OES; Fe, Cu, Zn, Mn, and Mo were
325 determined by ICP-MS. Numbers indicate the fold change of elemental content compared with the control
326 group. < 1 indicates that the content was decreased; > 1 indicates that the content was increased. * and **
327 indicates a significant difference at $p < 0.05$ and $p < 0.01$ (n = 6) compared with the control, respectively.
328

329 3.6 Enzyme Activities in the Cucumber Rhizosphere and Bulk Soil

330 The activity of soil enzymes is a valuable indicator of overall soil health and
331 functionality (Chaperon and Sauvé, 2007; Lessard *et al.*, 2013). In the bulk soil, CeO₂ NPs
332 treatment significantly increased the activities of arylsulfatase (46.8%), peroxidase (8.3%),
333 and phosphatase (93.0%). However, in the rhizosphere soil, CeO₂ NPs caused higher
334 enhancement of phosphatase (37.4%) but less enhancement of peroxidase activities (18.6%),
335 and didn't induce any change of arylsulfatase (**Table 1**). CeCl₃ treatment resulted in the
336 enhanced activity of invertase in the bulk soil (14.6%), and decreased activity of
337 dehydrogenase in the rhizosphere soil. These indicate responses of rhizosphere and bulk soils
338 to CeO₂ NPs are different. Plant roots can release root exudates, which can coat the NP
339 surface, potentially shielding the particles from reaction or chelating metal ions that are
340 released from the metal oxide NPs, consequently lessening the toxicity of particles (Tong *et al.*
341 *et al.*, 2007; Philippot *et al.*, 2013). These may partially explain the less significant effect of
342 CeO₂ NPs on the enzyme activity in the rhizosphere soil compared to that in the bulk soil.
343 Urease activity was not affected by 1000 mg/kg CeO₂ NPs and 100 mg/kg Ce³⁺ ions exposure,

344 which may be due to the fact that the microbial-secreted urease is very resistant to
 345 environmental breakdown in the soil (Zantua and Bremner, 1977) (**Table 1**). Soil phosphatase
 346 is an enzyme that can catalyze the mineralization of soil organophosphorus compounds,
 347 subsequently making phosphorus (P) available for uptake by plants (Margesin *et al.*, 2000;
 348 Belyaeva *et al.*, 2005). Interestingly, phosphatase activities in the rhizosphere in CeO₂ NPs
 349 group is much higher than that in the bulk soil in both control and CeCl₃ treatment. However,
 350 the P uptake was not increased by CeO₂ NPs (**Fig. 4**). In our study, we supplemented
 351 Hoagland solutions to the soil every day to provide nutrients including the P. The increase of
 352 soil phosphatase was thus not directly correlated with P uptake by plant. A recent study found
 353 that 100 mg/kg CeO₂ NPs inhibited urease and β-glucosidase activities but stimulate
 354 phosphatase activity (Li *et al.*, 2017). The authors hypothesized that the stimulation might be
 355 due to the changes in the phosphatase-associated microbes in the soil, potential from
 356 enhanced activity, or population size, which we will discuss in the following section.

357

358 **Table 1.** Enzyme activities in the cucumber rhizosphere and bulk soil after 20 days of exposure to
 359 CeO₂ NPs and CeCl₃. The data are means of six replicates ± standard deviation.

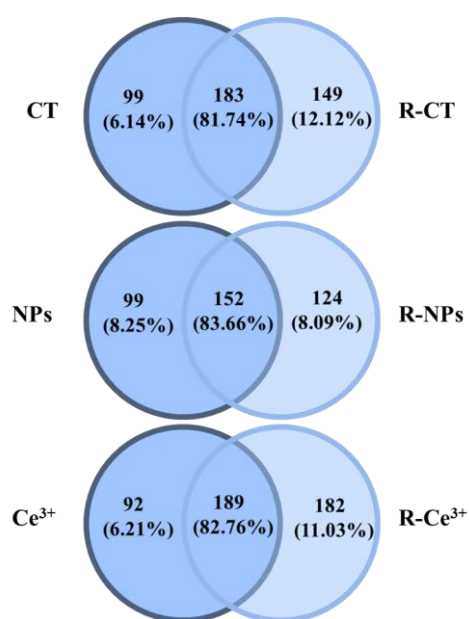
Enzyme activity	Bulk soil			Rhizosphere		
	CT	NPs	Ce ³⁺	R-CT	R-NPs	R-Ce ³⁺
Arylsulfatase (nmol MU g ⁻¹ soil h ⁻¹)	0.47±0.09	0.69±0.10**	0.44±0.03	0.46±0.08	0.55±0.11	0.52±0.09
Dehydrogenase (μmol /d/g soil)	464±54.9	469±32.3	501±9.51	501±17.4	461±42.1	462±20.3*
Invertase (mg/d/g soil)	6.59±0.58	6.59±0.21	7.55±0.69*	6.18±0.42	6.09±0.15##	6.41±0.28#
Peroxidase (mg/d/g soil)	65.2±1.63	70.6±4.37*	64.0±3.56	58.6±4.56	69.5±5.29*	65.4±9.09
Phosphatase (nmol MU g ⁻¹ soil h ⁻¹)	22.8±1.44	44.0±7.40*	20.7±6.62	33.7±0.80##	46.3±4.6**	31.7±2.27##
Urease (μg/d/g soil)	423±43.0	432±11.6	452±13.7	436±9.92	434±5.39	458±22.7
β-glucosidase (nmol MU g ⁻¹ soil h ⁻¹)	23.6±3.30	22.5±3.64	20.2±2.68	22.2±1.67	23.2±3.89	21.9±1.39

360 **Note:** CT=control, soil without CeO₂ NPs or Ce³⁺ ions added. R-CT means the soil planted cucumber
 361 seedlings. “*” means comparison with control at bulk soil and rhizosphere, respectively; “##” represents a
 362 comparison between bulk soil and rhizosphere under the same exposure conditions (*, #, *p* < 0.05; **, ##, *p*

363 < 0.01).

364 **3.7 Microbial Community Structure in Soil**

365 The difference in the enzymatic activity in the soil may result from the different soil
366 microbial communities (Brookes, 1995). The effects of metal and metal oxide NPs on soil
367 microbial activity, diversity, and abundance have been studied (Miao *et al.*, 2018; Fang *et al.*,
368 2022). However, little is known on the effects of CeO₂ NPs and Ce³⁺ ions on soil microbial in
369 the presence or absence of plants. For instance, our results showed that the operational
370 taxonomic units (OTU) number in the rhizosphere group was significantly higher than that in
371 bulk soil (**Fig. 5**), highlighting the role of plant root rhizosphere. Plant root exudates in
372 rhizosphere can provide nutrients for the microorganisms, promoting the metabolism and
373 proliferation of these populations (Weinert *et al.*, 2011; Khodakovskaya *et al.*, 2013). Besides,
374 the microbial composition in the rhizosphere soil showed difference from that in the bulk soil,
375 although 81.74% of OTU were shared by each group (**Fig. 5**). CeO₂ NPs exerted acute
376 toxicity to the bacterial community and reduced the bacterial diversity (**Fig. 5**), which was in
377 accordance to previously reported result (Miao *et al.*, 2018). However, CeCl₃ increased the
378 OTUs, which may be attributed to the unique physicochemical properties of Ce³⁺ which
379 affects the soil enzyme activities and bacterial communities. In both the bulk and rhizosphere
380 soil, the dominant phyla include *Proteobacteria*, *Bacteroidetes*, *Acidobacteria*,
381 *Actinobacteria*, *Verrucomicrobia*, and *Firmicutes*, accounting for > 85% of the total
382 microbial community (**Fig. 6a, Fig. S2**). The abundance of *Proteobacteria* (the most
383 dominant phylum (>60%)) in the rhizosphere soil was lower than that in bulk soil (**Fig. 6a**).
384 The difference in microbial composition between the bulk and rhizosphere soil group was
385 also observed at the class and genus levels (**Fig. 6b, 6c**).

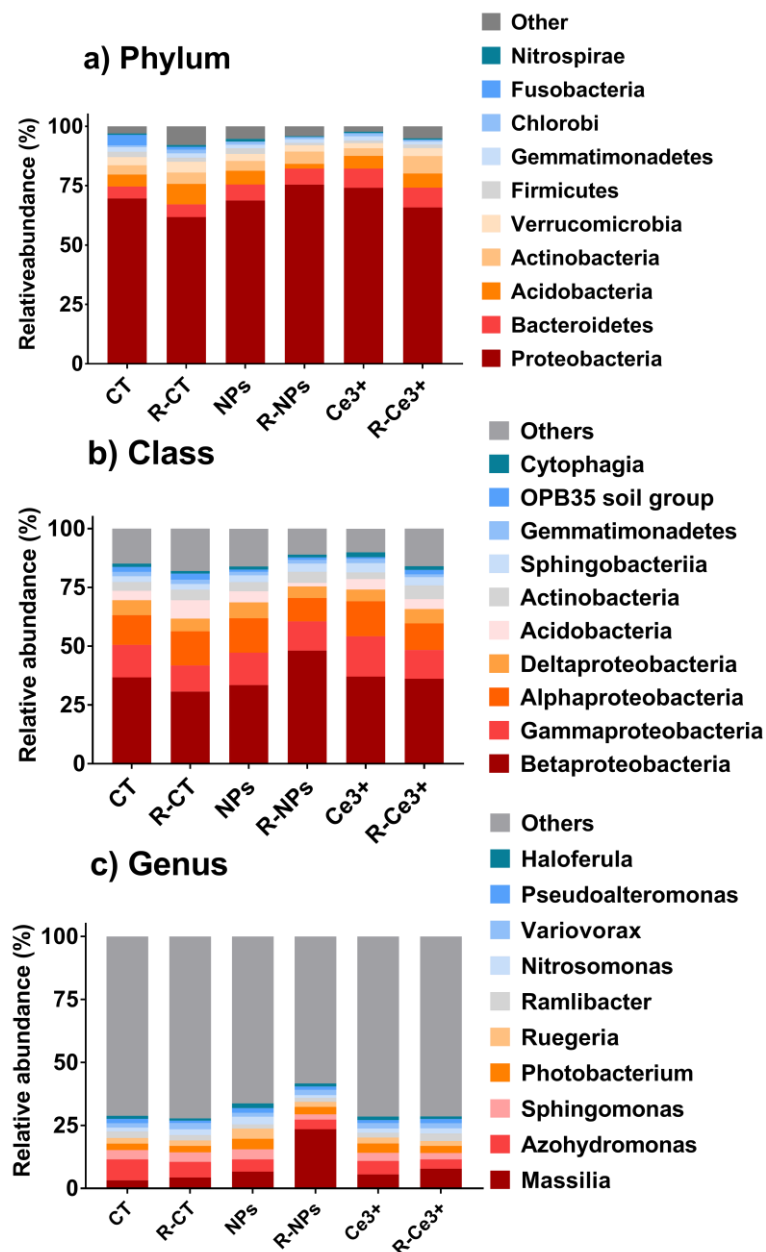


386

387 **Fig. 5** The analysis of common and different OTUs between control and treated groups was obtained from
 388 the sequencing data. CT, NPs, and Ce³⁺ are the three treatments in bulk soil, and the prefix R- represents
 389 soil incubation with the plant.

390 CeO₂ NPs treatment significantly changed the microbial populations in the rhizosphere
 391 soil but not in the bulk soil (**Fig. 5, 6**). A previous study also found that CeO₂ NPs (100
 392 mg/kg) altered soil bacterial communities in soybean-planted soil but did not affect bacterial
 393 communities in unplanted soil (Ge *et al.*, 2014). The relative abundance of *Proteobacteria*,
 394 *Bacteroidetes*, and *Actinobacteria* was increased while the abundance of *Acidobacteria* was
 395 decreased after CeO₂ NPs treatment in the rhizosphere soil (**Fig. 6**). Similarly, Pan *et al.*
 396 (2020) found that the abundances of *Proteobacteria* and *Bacteroidetes* at the phylum level
 397 increased after CeO₂ NPs exposure with the ferrous amendment. *Actinobacteria* plays vital
 398 roles in the decomposition of organic matter and the carbon cycle (Lewin *et al.*, 2016; Chen
 399 *et al.*, 2021). Phosphatase encoding gene (*pho D*) mainly existed in *Actinobacteria* (Luo *et al.*,
 400 2017). Therefore, the increased enzymatic activity of phosphatase may result from the
 401 increased abundance of *Actinobacteria*. However, the relative abundances of *Nitrospira* at the
 402 phylum level were decreased under CeO₂ NPs and Ce³⁺ treatment in the rhizosphere (**Fig. 6**).
 403 Compared with R-CT (0.8%), the relative abundance of *Nitrospira* at the phylum level was
 404 significantly lower than in the CeO₂ NPs treatment (0.37%) and Ce³⁺ ions treatment (0.6%).

405 These findings are in accordance with previous studies that the CeO₂ NPs impaired the soil
406 microbial community and soil organic carbon mineralization (Luo *et al.*, 2020). A recent
407 study also found that the relative abundance of *Nitrospira* was reduced after exposure to
408 10 mg/L CeO₂ NPs for xx days (Wang *et al.*, 2018a). Negative effects on the relative
409 abundance of *Nitrospira* were also reported for rGO, MWCNTs, and C60. *Nitrospira* is
410 involved in plant's nitrification processes (Hao *et al.*, 2018). The effects of CeO₂ NPs and
411 Ce³⁺ ions on *Nitrospira* indicate that they may negatively impact soil nitrogen cycling.
412 However, because of the lack of long-term experiments, the relationship between the change
413 of soil microbial communities and its soil enzyme activities, and whether it will eventually
414 have a positive or a negative impact on plant growth remains still unknown. We found that
415 the relative abundances of the rare bacteria (*e.g.*, *Euryarchaeota*, *Fibrobacteres*) at the
416 phylum level for the total bacterial community are also more sensitive to environmental
417 factors because of their response to the soil rhizosphere. The bacterial composition at the
418 class and genus level also shifted markedly upon CeO₂ NPs exposure (**Fig. S3** and **Fig. S4**).
419 Overall, although sharing some similarities, CeO₂ NPs showed particle-specific effects on
420 soil microorganisms compared with Ce³⁺ ions.



421
 422 **Fig. 6** Relative abundance of major phyla (a), class (b), and genus (c) in bulk soil, rhizosphere, or soil
 423 treated with CeO₂ NPs and CeCl₃ and the respective controls. CT, CeO₂ NPs, and Ce³⁺ are the three
 424 treatments in bulk soil, and Prefix R- represents soil incubation with the plant, respectively.

425 **Conclusions**

426 With the increasing application of nanomaterials in environmental remediation and
 427 agriculture, a thorough understanding of their environmental impacts are critical for their

428 sustainable design and safe use. In this study, the responses of soil microbial communities,
429 soil enzyme activities, and cucumber seedling growth in a soil-plant interactive system were
430 systematically investigated. CeO₂ NPs shared some similarities in the effects with CeCl₃,
431 which was attributed to the biotransformation of CeO₂ NPs in the rhizosphere. However,
432 CeO₂ NPs also show distinct nano-specific effects on the antioxidant system, organic nutrient
433 accumulation, and soil enzyme activities. Distinct microbial response in the rhizosphere with
434 that in bulk soils highlights the critical role of rhizosphere chemistry in nanomaterial-induced
435 soil impacts. This study indicates that any environmental factors that alter the rhizosphere
436 chemistry may affect the behavior and biological effects of NPs in soil-plant system. It
437 should be noted that the present study was a short-term study, the long-term effects of NPs
438 exposure on the resiliency of soil microbial communities and their functions should be
439 evaluated in the future.

440 **CRedit authorship contribution statement**

441 **Changjian Xie:** Investigation, Formal analysis, Writing-original draft, Funding
442 acquisition. **Zhiling Guo:** Formal analysis, Writing-review & editing. **Peng Zhang:**
443 Resource and Funding acquisition. **Jie Yang:** Investigation. **Junzhe Zhang:** Investigation.
444 **Yuhui Ma:** Investigation. **Xiao He:** Investigation. **Iseult Lynch:** Investigation. **Zhiyong**
445 **Zhang:** Conceptualization, Methodology, Formal analysis, Writing-original draft,
446 Writing-review & editing, Resources, Funding acquisition.

447 **Declaration of competing interest**

448 We declare we have no competing interests.

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